

Efficient Access to 1,5-Diketones from N-Sulfonyl Ketimines and 3-Chloropropiophenones

M.Sc. Thesis

by

ARJUN KUMBHAKAR



**DEPARTMENT OF CHEMISTRY
INDIAN INSTITUTE OF TECHNOLOGY
INDORE
2022-2023**

Efficient Access to 1,5-Diketones from N-Sulfonyl Ketimines and 3-Chloropropiophenones

A THESIS

Submitted in partial fulfillment of the requirements for the award of the degree of
Master of Science

by

ARJUN KUMBHAKAR



**DEPARTMENT OF CHEMISTRY
INDIAN INSTITUTE OF TECHNOLOGY
INDORE
2022-2023**



INDIAN INSTITUTE OF TECHNOLOGY INDORE

CANDIDATE'S DECLARATION

I hereby certify that the work which is being presented in the thesis entitled **“Efficient Access to 1,5-Diketones from N-Sulfonyl Ketimines and 3-Chloropropiophenones ”** in the partial fulfillment of the requirements for the award of the degree of **MASTER OF SCIENCE** and submitted in the **DEPARTMENT OF CHEMISTRY, Indian Institute of Technology Indore**, is an authentic record of my own work carried out during the time period from June 2022 to June 2023 under the supervision of **Prof. Sampak Samanta, IIT Indore**.

The matter presented in this thesis has not been submitted by me for the award of any other degree of this or any other institute.

Arjun Kumbhakar

Signature of the student
(**Arjun Kumbhakar**)

This is to certify that the above statement made by the candidate is correct to the best of my/our knowledge.

Sampak Samanta

Signature of Supervisor
(**Prof. Sampak Samanta**)

ARJUN KUMBHAKAR has successfully given his M.Sc. Oral Examination held on **16.05.2023**.

Sampak Samanta

Signature of Supervisor
(**Prof. Sampak Samanta**)
Date:16.05.2023

Dr. Umesh A. Kshirsagar

Convener, DPGC
(**Dr. Umesh A. Kshirsagar**)
Date: 16.05.2023

Venkatesh. c

Signature of PSPC Member
(**Dr. Chelvam Venkatesh**)
Date:16.05.2023

Dr. Selvakumar Sermadurai

Signature of PSPC Member
(**Dr. Selvakumar Sermadurai**)
Date: 16.05.2023

ACKNOWLEDGEMENTS

I wish to express my sincere thanks to those whose constant support and motivation made it possible to present my research work in form of a dissertation.

At this moment I acknowledge, first of all, I would like to say special thanks to thesis supervisor **Prof. Sampak Samanta** (Professor, Department of Chemistry, IIT Indore) for valuable guidance, suggestions and supervision throughout my work. Further, I would like to thank my PSPC members **Dr. Chelvam Venkatesh** and **Dr. Selvakumar Sermadurai** for their valuable suggestions and support.

I wish to express my gratitude to **Prof. Suhas S. Joshi** (Director, IIT Indore) for his continuous encouragement, help, and support in every aspect.

I would also like to thank HOD, Department of Chemistry for the support and guidance.

I would like to thank my mentor **Mr. Meher Prakash** and all the lab mates for their valuable suggestions, and help. Without their support would not possible.

I would like to thank Department of Chemistry, IIT Indore for working in the laboratory. I am also thankful to SIC, IIT Indore and its member for their technical help and support.

I would also like to thank my friends and family members specially my father and mother who always assisted me directly or indirectly during my masters also supported me mentally and emotionally throughout my post-graduation program without them I would unable to achieve this stage.

Arjun Kumbhakar
Department of Chemistry

Abstract

A series of 1,5-diketones were synthesized in good to high yields. This C-C bond-forming reaction proceeded between 4-alkyl N-sulfonyl ketimines and 3-chloropropiophenones in the presence of Cs_2CO_3 at heating conditions. Moreover, the current method is simple, has good substrate scope, and excellent tolerance of functionalities.

TABLE OF CONTENT

LIST OF FIGURES	VI-IX
LIST OF SCHEMES	X
LIST OF TABLES	XI
ACRONYMS	XII
NOMENCLATURE	XIII
CHAPTER 1: INTRODUCTION	
1.1 General Introduction	1
1.2 Review work	1-3
1.3 Conclusion of review work	3
1.4 The objectives of the present work	3
CHAPTER 2: RESULT AND DISCUSSIONS	
2.1 Optimization table	5-6
2.2 Substrate scope	7
2.3 Plausible reaction mechanism	8
2.4 Applications	9
CHAPTER 3: EXPERIMENTAL WORK	
3.1 Required materials and instrumentation	11
3.2 Standard procedure of synthesis	
3.2.1 Synthesis of N-sulfonyl ketimines	11-14
3.2.2 Synthesis of 3-chloropropiophenones	14
3.2.3 One-pot procedure synthesis of 1,5-diketones	15
CHAPTER 4: SYTHESIZED COMPOUND	
4.1 Data of all synthesized compounds	15-26
CHAPTER 5: CONCLUSION	27
APPENDIX A	29-69
REFERENCES	70

LIST OF FIGURES

Figure 1: ^1H NMR spectrum (500MHz) of 1aa in CDCl_3	29
Figure 2: ^{13}C NMR spectrum (125MHz) of 1aa in CDCl_3	29
Figure 3: ^1H NMR spectrum (500MHz) of 1ab in CDCl_3	30
Figure 4: ^{13}C NMR spectrum (125MHz) of 1ab in CDCl_3	30
Figure 5: ^1H NMR spectrum (500MHz) of 1ac in CDCl_3	31
Figure 6: ^{13}C NMR spectrum (125MHz) of 1ac in CDCl_3	31
Figure 7: ^1H NMR spectrum (500MHz) of 1ad in CDCl_3	32
Figure 8: ^{13}C NMR spectrum (125MHz) of 1ad in CDCl_3	32
Figure 9: ^1H NMR spectrum (500MHz) of 1ae in CDCl_3	33
Figure 10: ^{13}C NMR spectrum (125MHz) of 1ae in CDCl_3	33
Figure 11: ^1H NMR spectrum (500MHz) of 1af in CDCl_3	34
Figure 12: ^{13}C NMR spectrum (125MHz) of 1af in CDCl_3	34
Figure 13: ^1H NMR spectrum (500MHz) of 1ag in CDCl_3	35
Figure 14: ^{13}C NMR spectrum (125MHz) of 1ag in CDCl_3	35
Figure 15: ^1H NMR spectrum (500MHz) of 1ah in CDCl_3	36
Figure 16: ^{13}C NMR spectrum (125MHz) of 1ah in CDCl_3	36
Figure 17: ^1H NMR spectrum (500MHz) of 1ai in CDCl_3	37
Figure 18: ^{13}C NMR spectrum (125MHz) of 1ai in CDCl_3	37
Figure 19: ^1H NMR spectrum (500MHz) of 2aa in CDCl_3	38
Figure 20: ^{13}C NMR spectrum (125MHz) of 2aa in CDCl_3	38
Figure 21: ^1H NMR spectrum (500MHz) of 3aa in CDCl_3	39
Figure 22: ^{13}C NMR spectrum (125MHz) of 3aa in CDCl_3	39
Figure 23: ^1H NMR spectrum (500MHz) of 3ab in CDCl_3	40

Figure 24: ^{13}C NMR spectrum (125MHz) of 3ab in CDCl_3	40
Figure 25: ^1H NMR spectrum (500MHz) of 3ac in CDCl_3	41
Figure 26: ^{13}C NMR spectrum (125MHz) of 3ac in CDCl_3	41
Figure 27: ^1H NMR spectrum (500MHz) of 3ad in CDCl_3	42
Figure 28: ^{13}C NMR spectrum (125MHz) of 3ad in CDCl_3	42
Figure 29: ^1H NMR spectrum (500MHz) of 3ba in CDCl_3	43
Figure 30: ^{13}C NMR spectrum (125MHz) of 3ba in CDCl_3	43
Figure 31: ^1H NMR spectrum (500MHz) of 3bd in CDCl_3	44
Figure 32: ^{13}C NMR spectrum (125MHz) of 3bd in CDCl_3	44
Figure 33: ^1H NMR spectrum (500MHz) of 3ca in CDCl_3	45
Figure 34: ^{13}C NMR spectrum (125MHz) of 3ca in CDCl_3	45
Figure 35: ^1H NMR spectrum (500MHz) of 3cd in CDCl_3	46
Figure 36: ^{13}C NMR spectrum (125MHz) of 3cd in CDCl_3	46
Figure 37: ^1H NMR spectrum (500MHz) of 3da in CDCl_3	47
Figure 38: ^{13}C NMR spectrum (125MHz) of 3da in CDCl_3	47
Figure 39: ^1H NMR spectrum (500MHz) of 3db in CDCl_3	48
Figure 40: ^{13}C NMR spectrum (125MHz) of 3db in CDCl_3	48
Figure 41: ^1H NMR spectrum (500MHz) of 3dc in CDCl_3	49
Figure 42: ^{13}C NMR spectrum (125MHz) of 3dc in CDCl_3	49
Figure 43: ^1H NMR spectrum (500MHz) of 3dd in CDCl_3	50
Figure 44: ^{13}C NMR spectrum (125MHz) of 3dd in CDCl_3	50
Figure 45: ^1H NMR spectrum (500MHz) of 3ea in CDCl_3	51
Figure 46: ^{13}C NMR spectrum (125MHz) of 3ea in CDCl_3	51
Figure 47: ^1H NMR spectrum (500MHz) of 3eb in CDCl_3	52
Figure 48: ^{13}C NMR spectrum (125MHz) of 3eb in CDCl_3	52

Figure 49: ^1H NMR spectrum (500MHz) of 3ee in CDCl_3	53
Figure 50: ^{13}C NMR spectrum (125MHz) of 3ee in CDCl_3	53
Figure 51: ^1H NMR spectrum (500MHz) of 3ed in CDCl_3	54
Figure 52: ^{13}C NMR spectrum (125MHz) of 3ed in CDCl_3	54
Figure 53: ^1H NMR spectrum (500MHz) of 3ae in CDCl_3	55
Figure 54: ^{13}C NMR spectrum (125MHz) of 3ae in CDCl_3	55
Figure 55: ^1H NMR spectrum (500MHz) of 3af in CDCl_3	56
Figure 56: ^{13}C NMR spectrum (125MHz) of 3af in CDCl_3	56
Figure 57: ^1H NMR spectrum (500MHz) of 3fa in CDCl_3	57
Figure 58: ^{13}C NMR spectrum (125MHz) of 3fa in CDCl_3	57
Figure 59: ^1H NMR spectrum (500MHz) of 3fd in CDCl_3	58
Figure 60: ^{13}C NMR spectrum (125MHz) of 3fd in CDCl_3	58
Figure 61: ^1H NMR spectrum (500MHz) of 3ha in CDCl_3	59
Figure 62: ^{13}C NMR spectrum (125MHz) of 3ha in CDCl_3	59
Figure 63: ^1H NMR spectrum (500MHz) of 3ak in CDCl_3	60
Figure 64: ^{13}C NMR spectrum (125MHz) of 3ak in CDCl_3	60
Figure 65: ^1H NMR spectrum (500MHz) of 3ai in CDCl_3	61
Figure 66: ^{13}C NMR spectrum (125MHz) of 3ai in CDCl_3	61
Figure 67: ^1H NMR spectrum (500MHz) of 3aj in CDCl_3	62
Figure 68: ^{13}C NMR spectrum (125MHz) of 3aj in CDCl_3	62
Figure 69: ^1H NMR spectrum (500MHz) of 3ag in CDCl_3	63
Figure 70: ^{13}C NMR spectrum (125MHz) of 3ag in CDCl_3	63
Figure 71: ^1H NMR spectrum (500MHz) of 3al in CDCl_3	64
Figure 72: ^{13}C NMR spectrum (125MHz) of 3al in CDCl_3	64
Figure 73: ^1H NMR spectrum (500MHz) of 3am in CDCl_3	65

Figure 74: ^{13}C NMR spectrum (125MHz) of 3am in CDCl_3	65
Figure 75: ^1H NMR spectrum (500MHz) of 3aa' in CDCl_3	66
Figure 76: ^{13}C NMR spectrum (125MHz) of 3aa' in CDCl_3	66
Figure 77: ^1H NMR spectrum (500MHz) of 4aa in CDCl_3	67
Figure 78: ^{13}C NMR spectrum (125MHz) of 4aa in CDCl_3	67
Figure 79: ^1H NMR spectrum (500MHz) of 6aa in CDCl_3	68
Figure 80: ^{13}C NMR spectrum (125MHz) of 6aa in CDCl_3	68
Figure 81: ^1H NMR spectrum (500MHz) of 7aa in CDCl_3	69
Figure 82: ^{13}C NMR spectrum (125MHz) of 7aa in CDCl_3	69

LIST OF SCHEMES

Scheme 1: synthesis of 1,5-diketones containing ferrocenyl via Michael addition under solvent free condition	1
Scheme 2: One-pot synthesis of 1,5-diketones using barium isopropoxide catalyst	2
Scheme 3: Intramolecular acylation in the presence of hexamethylditin	2
Scheme 4: Synthesis of 1,5-diketones from using ketoxentane	2
Scheme 5: Synthesis of 1,5-diketones from vinyl cyclobutanol and aldehyde using rhodium-catalyst.	3
Scheme 6: Control experiments.	7
Scheme 7: Reaction pathway for the synthesis of 1,5-diketone derivatives	8
Scheme 8: Application of 1,5-diketones	9
Scheme 9: Synthesis of N-Sulfonyl ketimines (1a)	11
Scheme 10: Synthesis of 3-chloropropiophenone	14
Scheme 11: Synthesis of 1,5-diketones (3a).	15

LIST OF TABLES

Table 1: Optimization of the reaction conditions	5
Table 2: Substrate scope	7

ACRONYMS

ACN	Acetonitrile
CDCl₃	Chloroform-D
¹³C NMR	Carbon-13 NMR Spectroscopy
DMSO	Dimethyl sulfoxide
DCM	Dichloromethane
DBU	1,8-Diazabicyclo[5.4.0]undec-7-ene
DABCO	1,4-Diazabicyclo[2,2,2]octane
DBN	1,5-Diazabicyclo[4.3.0]non-5-ene
DMF	N, N-dimethyl formamide
EtOAc	Ethyl acetate
HRMS	High resolution mass spectroscopy
2-MeTHF	2-Methyltetrahydrofuran
M	Molar
NMR	Nuclear magnetic resonance
ppm	Parts per million
UV	Ultra visible spectroscopy

NOMENCLATURE

δ	Chemical shift
cm	Centimetre
$^{\circ}\text{C}$	Degree Celsius
mmol	Millimole
mL	Millilitre
rt	Room temperature

Chapter 1

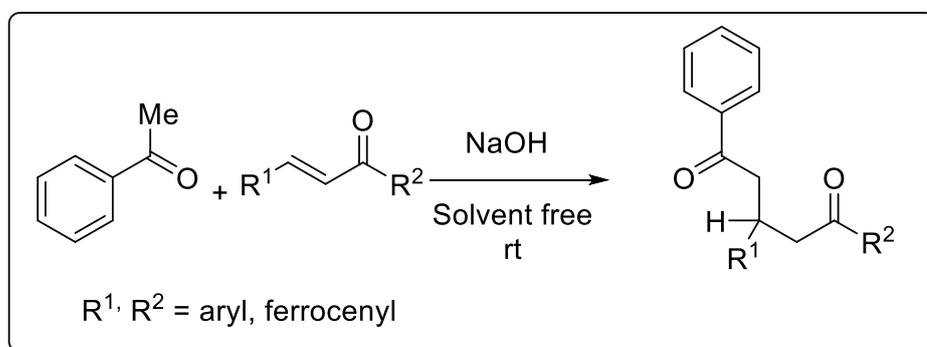
1 INTRODUCTION

1.1 General Introduction

Among the functional groups, ketones are the most important and valuable functional groups in organic molecules. Particularly, 1,5-diketones are the building blocks in the synthesis of a variety of heterocyclic molecules and other related skeletons which are essential subunits of many natural products. Particularly, they have promising applications in the synthesis of vital classes of aza-heterocyclic compounds such as pyridine [1], pyrazolo[1,5-a]pyrimidines [2], etc. Thus, many methods have been developed for synthesizing 1,5-diketones. Some of them methods are included in the next section.

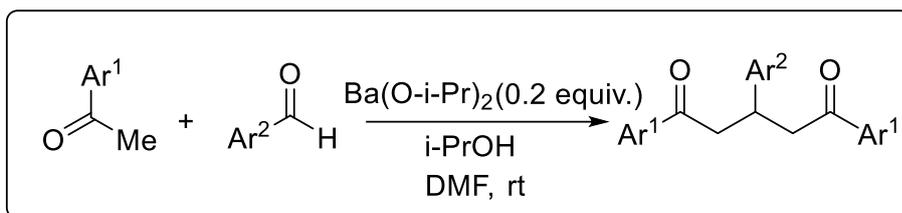
1.2. Review work

In 2001, Liu et al. established a very significant environmental-friendly method for the synthesis 1,5-diketone in good yields from acetophenones and α , β -unsaturated ketones using NaOH as a base under solvent-free conditions in Scheme 1. [3]



Scheme 1: Base-promoted access to 1,5-diketones

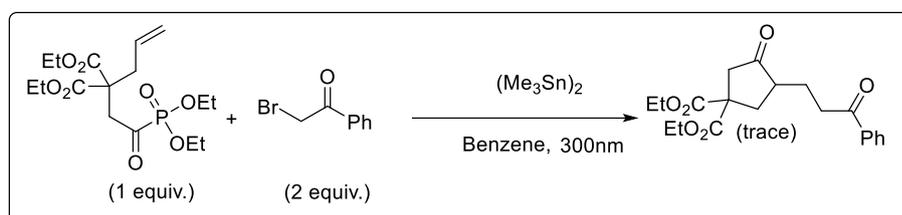
In 2007, Yanagisawa et al. developed an efficient way to synthesize 1,5-diketones, which has been achieved by a catalytic amount of barium isopropoxide in a mixture of acetophenone derivatives and benzaldehyde derivatives in DMF at room temperature (Scheme 2) [4]



Scheme 2: One-pot synthesis of 1,5-diketones using barium isopropoxide catalyst.

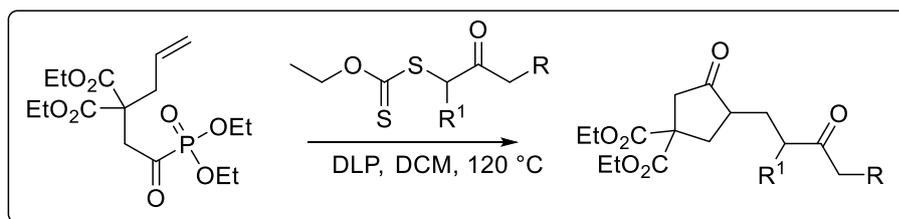
In 2013, Zard et al. modified the methods to synthesize 1,5-diketones via aldol-Michael between α -halo ketones and alkenylacylphosphonate as a carbonyl group acceptor in the presence of hexamethylditin.

(Scheme: 3) [5]



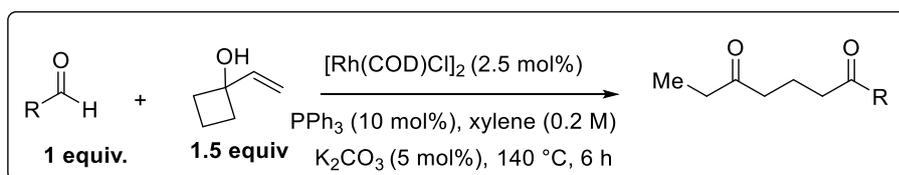
Scheme: 3 Intramolecular acylation in the presence of hexamethylditin.

Because of the toxicity of tin and low yield, they modified the method by using ketoxanthates instead of α -haloketone which proceeded through the radical pathway. **(Scheme: 4)** [5]



Scheme 4: Synthesis of 1,5-diketones from using ketoxanthane.

In 2017, Gua et al. developed rhodium catalyst hydroacylation using vinyl cyclobutanol and aldehyde that afforded 1,5-diketones. **(Scheme: 5)** [6]



Scheme 5: Synthesis of 1,5-diketones from vinyl cyclobutanols and aldehydes using rhodium-catalyst.

1.3 Conclusions

Literature study revealed that many efficient techniques have been established for accessing 1,5-diketones. However, they have their own drawbacks, including the use of toxic elements, the use of expensive metal catalysts, poor yields (few cases), limited structural scope, and the formation of by-products. Therefore, there is an ample scope to develop a new transition-metal-free protocol for assembling a diverse set of 1,5-diketones from simple reactants under mild conditions.

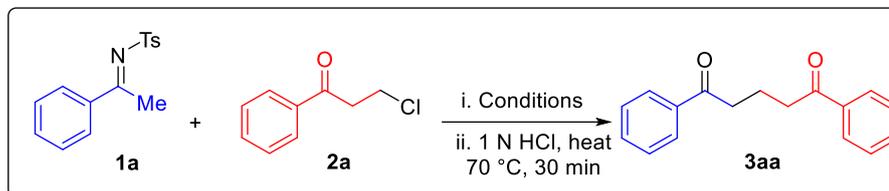
1.4 Objectives of the present work

The synthesis of 1,5-diketone derivatives is subject of growing interest in recent years due to their great importance in organic synthesis. For this purpose, here in we report a novel C-C bond forming reaction between 4-methyl-N-(1-arylethylidene)benzenesulfonamides and 3-chloropropiophenones promoted by Cs_2CO_3 in 2-MeTHF at heating conditions to deliver 1,5-diketones.

Chapter 2

2. RESULTS AND DISCUSSION:

2.1 Optimization of the reaction conditions^a



Entry	Base	Solvent	Temperature(°C)	Time(h)	Yield ^b (3aa)(%)
1.	Cs ₂ CO ₃	2-MeTHF	rt	12	0
2.	Cs ₂ CO ₃	2-MeTHF	50	6	49
3.	Cs₂CO₃	2-MeTHF	70	6	72
4.	Cs ₂ CO ₃	2-MeTHF	100	6	63
5.	Cs ₂ CO ₃	THF	70	6	61
6.	Cs ₂ CO ₃	MeCN	70	6	46
7.	Cs ₂ CO ₃	DCE	70	6	47
8.	Cs ₂ CO ₃	Toluene	70	6	57
9.	Cs ₂ CO ₃	DMSO	70	6	31
10.	Cs ₂ CO ₃	DMF	70	6	ND
11.	Cs ₂ CO ₃	1,4-Dioxane	70	6	34
12.	Cs ₂ CO ₃	EtOH	70	6	59
13.	DBU	2-MeTHF	70	6	61
14.	DBN	2-MeTHF	70	6	55
15.	DABCO	2-MeTHF	70	6	ND
16.	NEt ₃	2-MeTHF	70	6	27
17.	DEIPA	2-MeTHF	70	6	41
18.	K ₂ CO ₃	2-MeTHF	70	6	ND
19.	NaOH	2-MeTHF	70	6	ND
20.	KOH	2-MeTHF	70	6	57
21.	t-BuOK	2-MeTHF	70	6	15

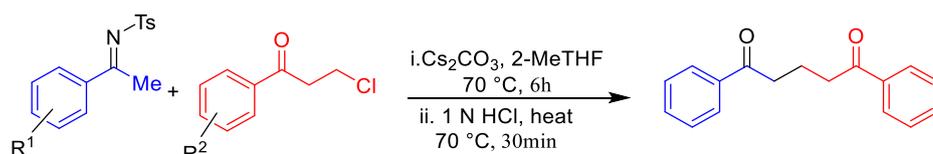
^aAll the reactions are carried out with **1a** (0.18 mmol, 1 equiv.), **2a** (0.18 mmol, 1 equiv.), and base (1.2 equiv.) in dry solvent (0.5 mL) at different temperatures. ^bYield refers to isolated product **3aa** after column chromatography.

We did the model reaction between N-sulfonyl ketimine (**1a**) and 3-chloropropiophenone (**1b**) in 2-MeTHF using Cs₂CO₃ as a robust base at different temperatures and observed by TLC. After 6h, the reaction was completed and **3aa** was isolated in good yield at 70 °C. To improve the yield further, several common solvents such as THF, DCE, MeCN, 1,4-dioxane, toluene, DMF, DMSO, and EtOH were tested for this reaction. Most of the solvents gave moderate to good yields. Thus,

considering the yield, 2-MeTHF was best solvent for this reaction at 70 °C (entry 3). Further, we screened other bases such as DBU, DBN, DABCO, DEIPA, NEt₃, K₂CO₃, NaOH, KOH, and *t*-BuOK. Therefore, considering the yield of the desired product **3aa**, Cs₂CO₃ in 2-MeTHF at 70°C is the best-optimized condition for the reaction. It should be noted that using DABCO as a base, we isolated MBH adduct **4aa** instead of **3aa**.

2.2 Substrate Scope

Having optimal reaction parameters in hand, we demonstrated the substrate scope of the reaction by taking various N-sulfonyl ketimines and 3-chloropropiophenones in 2-MeTHF at 70 °C in the presence of Cs₂CO₃ under present conditions. The results are included in Table 2. It was found that a variety of aryl-substituted 3-chloropropiophenones have converted into vinyl ketones which were attacked by several 4-methyl-N-(1-arylethylidene)benzenesulfonamides as carbonucleophiles under present basic conditions, followed by hydrolysis of imine bonds. All of the reactions afforded the corresponding symmetrical and unsymmetrical 1,5-diarylpentane-1,5-diones in good to high yields. The current method tolerates a bunch of functionalities Me, MeO, Cl, Br, I, C=O, F, thiophene, cyclohexyl etc.



R¹ = alkyl, aryl,... R² = alkyl, aryl,.....

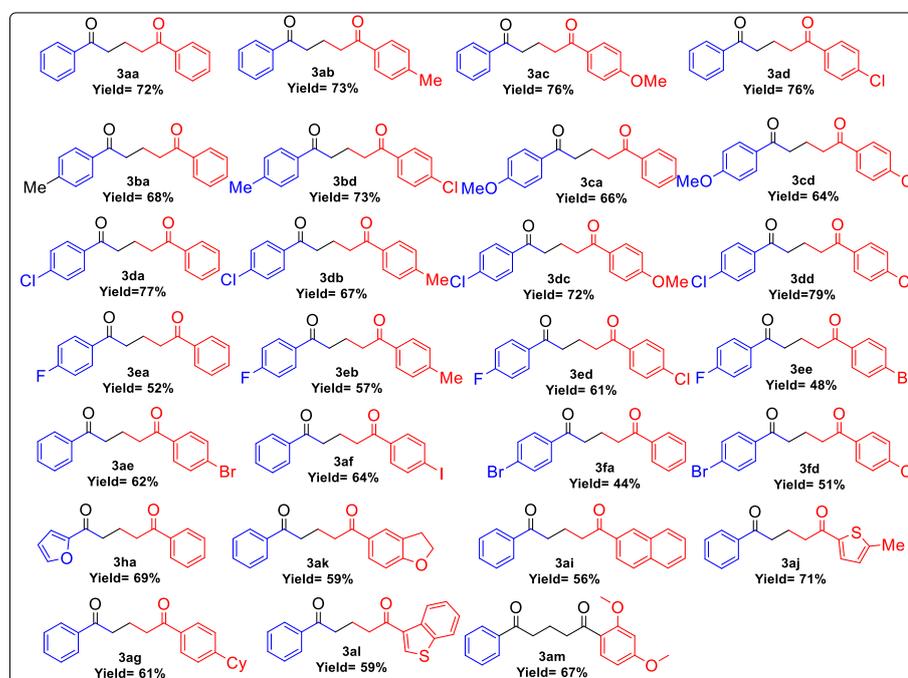
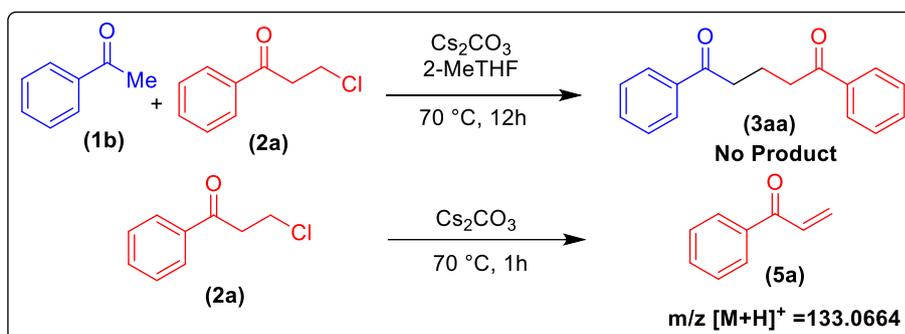


Table 2: Substrate Scope.

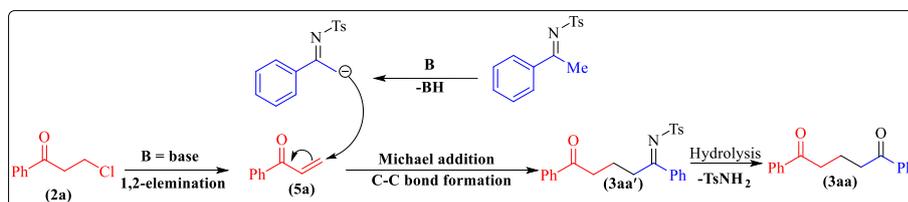
Control experiments were carried out to gain insight the pathway of the reaction (**Scheme 8**). Notably, the reaction between acetophenone (**1b**) and 3-chloropropiophenone (**2a**) was carried out under the best-optimized conditions, we didn't get any desired product **3aa**. Moreover, 3-chloropropiophenone (**2a**) was heated with Cs₂CO₃, at 70 °C, it led to vinyl ketone (**5a**).



Scheme 6: Control experiments

2.3 Plausible Reaction Mechanism

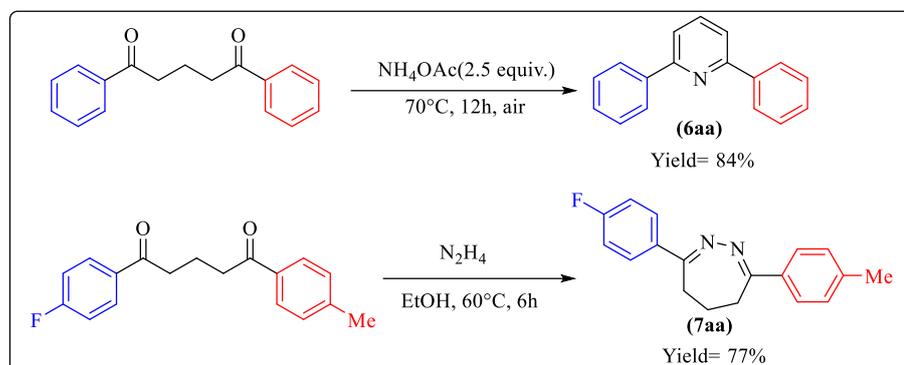
Based on the above control experimental data the plausible mechanism is drawn in **Scheme 9**. Initially, 3-chloropropiophenone undergoes elimination and thereby forms vinyl ketone (**5a**) which then undergoes Michael addition by the nucleophile **A**, generated from N-sulfonyl ketimine, and forms the intermediate **3aa'** (4-methyl-N-(5-oxo-1,5-diphenylpentylidene)benzenesulfonamide). Then intermediate **3aa'** undergoes hydrolysis to form the desired product **3aa** (1,5-diphenylpentane-1,5-dione).



Scheme 7: Reaction pathway for the synthesis of 1,5-diketone derivatives.

2.4 Applications:

Here we have shown some applications of 1,5-diketone derivatives which provide good yields.



Scheme 8: Application of 1,5-diketones.

Chapter 3

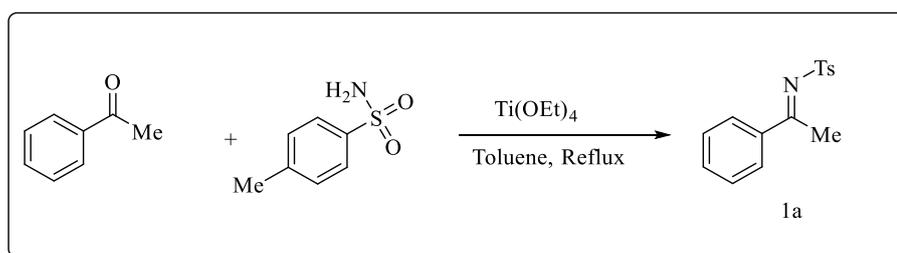
3. EXPERIMENTAL WORK

3.1 Required Materials and Instrumentation:

All the chemicals were bought from Sigma Aldrich and Spectrochem. All the reactions were in a closed tube and observed by TLC using Merck 60 F254 pre-coated silica gel plate and the product was apprehended by UV detection. Silica gel (60-120 mesh) was used for column chromatography. ^1H NMR (500 MHz) and ^{13}C NMR (125 MHz) data were taken in CDCl_3 solvent using Bruker Advance 500. NMR data processed by MestReNova.

3.2 Standard procedure for the synthesis:

3.2.1 Synthesis of N-sulfonyl ketimines:

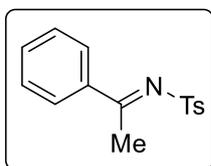


Scheme 9: Synthesis of N-Sulfonyl ketimines (**1a**).

Firstly, a mixture of acetophenone (0.97 mL, 8.33 mmol) and p-toluene sulfonamide (1.71 gm, 10 mmol) were taken in a double-neck 250 mL round bottom flask and then brought together with a water-cooled reflux condenser. A balloon infused with argon was put over the top of the condenser. On the other side in a 50 mL beaker, toluene (10 mL) was taken and titanium ethoxide (1.9 gm, 8.33 mmol) dissolved in it. After that, the toluene-titanium solution was added to the mixture (inert condition) under rapid stirring at room temperature. Then the mixture is stirred and refluxed. After 6 hours, it was permitted to settle down to rt and then it was quenched with 5-10 mL of NaHCO_3 solution. The mixture was filtered and the workup of the filtrate was done by ethyl acetate. The organic layer was dried by Na_2SO_4 . Column

chromatography (1:9; ethyl acetate/hexane) was used to purify the crude product. [7,8]

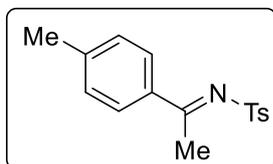
4-Methyl-N-(1-phenylethylidene)benzenesulfonamide (1aa): ^1H



NMR (500 MHz, CDCl_3) δ 7.93 (d, $J = 8.0$ Hz, 2H), 7.89 (d, $J = 7.9$ Hz, 2H), 7.52 (t, $J = 7.3$ Hz, 1H), 7.40 (t, $J = 7.7$ Hz, 2H), 7.34 (d, $J = 8.0$ Hz, 2H), 2.98 (s, 3H), 2.43 (s, 3H). ^{13}C **NMR (125**

MHz, CDCl_3) δ 179.8, 143.5, 138.7, 137.5, 133.2, 129.5, 128.6, 128.3, 127.1, 21.6, 21.1 ppm.

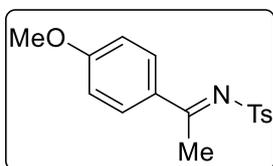
4-Methyl-N-(1-(4-methylphenyl)ethylidene)benzenesulfonamide



(1ab): ^1H **NMR (500 MHz, CDCl_3)** δ 7.92 (d, $J = 8.0$ Hz, 2H), 7.80 (d, $J = 8.1$ Hz, 2H), 7.33 (d, $J = 7.9$ Hz, 2H), 7.20 (d, $J = 8.0$ Hz, 2H), 2.95 (s, 3H), 2.44 (s, 3H), 2.38 (s, 3H) ppm; ^{13}C

NMR (125 MHz, CDCl_3) δ 179.7, 144.2, 143.4, 138.9, 134.8, 129.4, 128.4, 129.3, 127.0, 21.61, 21.0 ppm.

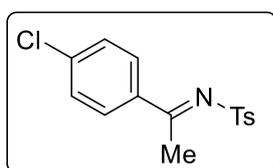
N-(1-(4-methoxyphenyl)ethylidene)-4-methylbenzenesulfonamide



(1ac): ^1H **NMR (500 MHz, CDCl_3)** δ 7.93-7.88 (m, 4H), 7.32 (d, $J = 8.0$ Hz, 2H), 6.88 (d, $J = 8.3$ Hz, 2H), 3.84 (s, 3H), 2.93 (s, 3H), 2.43 (s, 3H) ppm; ^{13}C **NMR (125 MHz,**

CDCl_3) δ 178.7, 163.9, 143.3, 139.0, 130.6, 130.6, 129.4, 126.9, 113.9, 55.5, 21.5, 20.7 ppm.

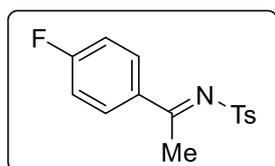
N-(1-(4-chlorophenyl)ethylidene)-4-methylbenzenesulfonamide



(1ad): ^1H **NMR (500 MHz, CDCl_3)** δ 7.91 (d, $J = 8.2$ Hz, 2H), 7.83 (d, $J = 8.6$ Hz, 2H), 7.36 (m, 4H), 2.96 (s, 3H), 2.45 (s, 3H) ppm; ^{13}C **NMR (125 MHz, CDCl_3)** δ 178.4, 143.7,

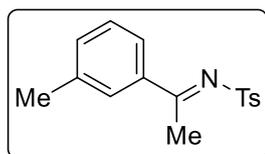
139.7, 138.4, 135.9, 129.6, 129.5, 128.9, 127.1, 21.6, 21.0 ppm.

N-(1-(4-fluorophenyl)ethylidene)-4-methylbenzenesulfonamide



(1ae): $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 7.96-7.89 (m, 4H), 7.35 (d, $J = 8.2$ Hz, 2H), 7.08 (m, 2H), 2.96 (s, 3H), 2.44 (s, 3H) ppm; $^{13}\text{C NMR}$ (126 MHz, CDCl_3) δ 178.2, 165.9 (d, $J = 255.6$ Hz), 143.6, 138.6, 133.7 (d, $J = 2.9$ Hz), 130.8 (d, $J = 9.2$ Hz), 129.5, 127.1, 115.8 (d, $J = 22.0$ Hz), 21.6, 21.0 ppm.

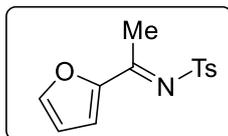
4-Methyl-N-(1-(3-methylphenyl)ethylidene)benzenesulfonamide



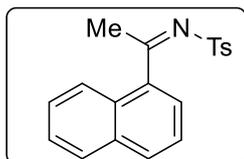
(1af): $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 7.93 (d, $J = 8.3$ Hz, 2H), 7.69 (d, $J = 6.3$ Hz, 2H), 7.34 (d, $J = 8.4$ Hz, 3H), 7.31-7.27 (m, 1H), 2.97 (s, 3H), 2.44 (s, 3H), 2.37 (s, 3H). $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 180.2, 143.5, 138.7, 138.4, 137.6, 134.0, 129.4, 128.7, 128.5, 127.1, 125.5, 21.6, 21.3 ppm.

N-(1-(furan-2-yl)ethylidene)-4-methylbenzenesulfonamide (1ag):

$^1\text{H NMR}$ (500 MHz, CDCl_3) δ 7.90 (d, $J = 8.3$ Hz, 2H), 7.62 (s, 1H), 7.32 (d, $J = 8.0$ Hz, 2H), 7.26-7.23 (m, 1H), 6.56-6.53 (m, 1H), 2.85 (s, 3H), 2.43 (s, 3H). $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 168.2, 152.3, 147.9, 143.5, 138.5, 129.4, 127.1, 119.0, 113.1, 21.5, 19.7 ppm.

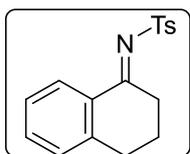


4-Methyl-N-(1-(naphthalen-1-yl)ethylidene)benzenesulfonamide



(1ah): $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 8.19 (d, $J = 7.4$ Hz, 1H), 7.96 (d, $J = 7.4$ Hz, 2H), 7.92 (d, $J = 8.2$ Hz, 1H), 7.88-7.85 (m, 1H), 7.63-7.59 (m, 1H), 7.54-7.45 (m, 3H), 7.32 (d, $J = 8.2$ Hz, 2H), 3.12 (s, 3H), 2.41 (s, 3H). $^{13}\text{C NMR}$ (126 MHz, CDCl_3) δ 184.0, 143.8, 138.3, 137.7, 133.8, 131.6, 129.5, 129.5, 128.6, 127.4, 127.2, 126.4, 126.3, 124.9, 124.7, 26.0, 21.6 ppm.

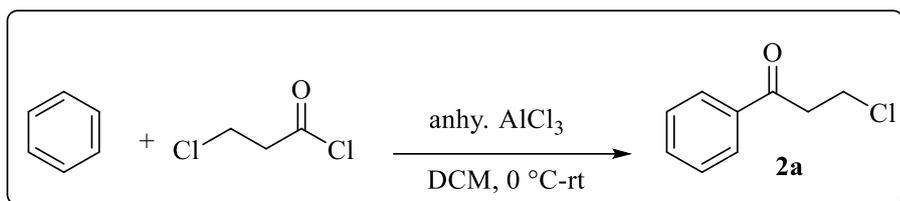
N-(3,4-dihydronaphthalen-1(2*H*)-ylidene)-4-



methylbenzenesulfonamide (1ai): $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 8.07 (d, $J = 7.3$ Hz, 1H), 7.93 (d, $J = 8.3$ Hz, 2H), 7.45-7.40 (m, 1H), 7.33 (d, $J = 8.0$ Hz, 2H), 7.20 (t, $J = 7.3$ Hz, 2H), 3.41 (s, 2H), 2.89 (s, 2H), 2.44 (s, 3H), 2.09-1.99 (m, 2H). $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 179.9, 144.3, 143.3, 139.0, 133.7, 131.9, 129.4, 129.0, 127.7, 126.9, 126.7, 33.0, 29.3, 22.4, 21.6 ppm.

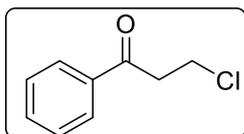
3.2.2 Synthesis of 3-chloropropiophenones (2a):

Firstly, in a 250 mL double-neck round bottom flask anhydrous AlCl_3 (6.14 gm, 46.15 mmol) was taken in DCM (10 mL) and connected to a balloon filled with argon. The atmosphere inside the flask made it inert. After that 3-chloropropionyl chloride (3.98 mL, 42.31 mmol) was added to anhydrous AlCl_3 in stirring condition at 0°C and then benzene (3.43 mL, 46.15 mmol) was added to the mixture under stirring condition for 3 hours and allowed the reaction to come to rt. After 3 hours the reaction was checked using a TLC plate. Then the workup was done using DCM. The DCM layer was dried using Na_2SO_4 and filtered. The filtrate was concentrated using a rotary evaporator to yield crude and purified using column chromatography. (Scheme: 12) [9,10]



Scheme 10: Synthesis of 3-chloropropiophenone.

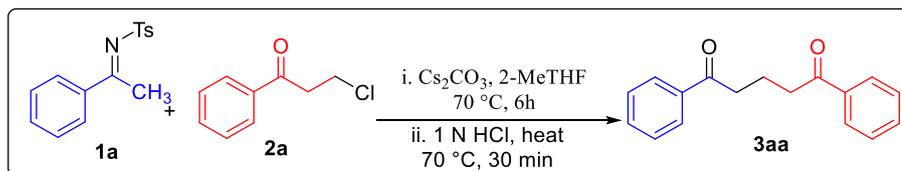
3-Chloro-1-phenylpropan-1-one (2aa): $^1\text{H NMR}$ (500 MHz, CDCl_3)



δ 7.99-7.92 (m, 2H), 7.59 (t, $J = 7.4$ Hz, 1H), 7.48 (t, $J = 7.8$ Hz, 2H), 3.92 (t, $J = 6.8$ Hz, 2H), 3.45 (t, $J = 6.8$ Hz, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 196.72, 136.3, 133.5, 128.7, 128.0, 41.2, 38.7 ppm.

3.2.3 One-pot procedure for the synthesis of 1,5-diketones:

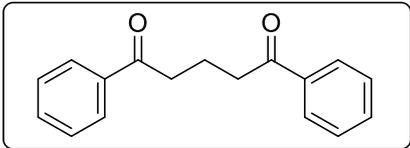
In a dry closed test tube, mixture of 4-methyl-N-(1-phenylethylidene) benzene sulfonamide (50 mg, 0.183 mmol, 1.0 equivalent), 3-chloropropiophenone (31 mg, 0.183 mmol, 1.0 equivalent), cesium carbonate (50 mg, 0.220 mmol) and 2-MeTHF (0.5 mL) taken and then it stirred under 70°C for 12h. The completion of the reaction judged by TLC, afterwards, the reaction was quenched by 1 N HCl and stirred for 30 min in 70°C. Then the reaction mixture was extracted with ethyl acetate (30 mL), washed with brine and dried using Na₂SO₄. After that, concentrated using a rotary evaporator under reduced pressure to yield the crude, which was purified using column chromatography over silica gel (60-120 mesh) with EtOAc/hexane (1:9, v/v). ¹H and ¹³C data were use to characterize the product with MastReNova.



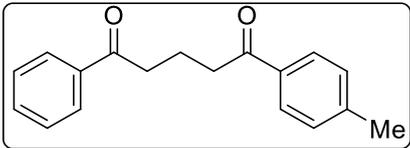
Scheme 11: Synthesis of 1,5-diketones (**3aa**).

4.2. Data of all synthesized compounds

1,5-Diphenylpentane-1,5-dione (3aa): White solid; mp 58-60 °C; yield

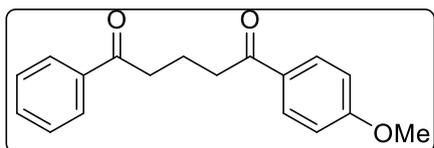
 = 72%; R_f = 0.69 (ethyl acetate/hexane = 1:9); ¹H NMR (500 MHz, CDCl₃) δ 7.98 (d, J = 7.6 Hz, 2H), 7.59-7.52 (t, J = 7.2 Hz, 2H), 7.45 (t, J = 7.6 Hz, 1H), 3.12 (t, J = 6.9 Hz, 4H), 2.24-2.18 (m, 2H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 199.8, 136.8, 133.0, 128.6, 128.0, 37.6, 18.7 ppm; HRMS (ESI-TOF): m/z [M+H]⁺ calcd for C₁₇H₁₆O₂ 253.1223, found 253.1210.

1-Phenyl-5-(4-methylphenyl)pentane-1,5-dione (3ab): White solid;

 mp 64-66 °C; yield = 73%; R_f = 0.71 (ethyl acetate/hexane = 1:9); ¹H NMR (500 MHz, CDCl₃) δ 8.00-7.96 (m, 2H), 7.88 (d, J = 8.2 Hz, 2H), 7.55 (t, J = 7.4 Hz, 2H), 7.45 (t,

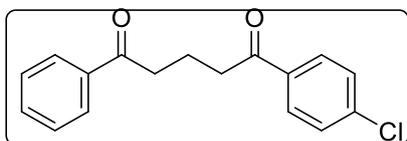
$J = 7.7$ Hz, 2H), 7.25 (d, $J = 8.0$ Hz, 2H), 3.13-3.06 (m, 2H), 2.40 (s, 3H), 2.22-2.15 (m, 2H) ppm; ^{13}C NMR (125 MHz, CDCl_3) δ 199.9, 199.5, 143.8, 136.8, 134.4, 133.0, 129.2, 128.6, 128.2, 128.0, 37.6, 37.5, 21.6, 18.8 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{18}\text{H}_{18}\text{O}_2$ 267.1380, found 267.1380.

1-(4-Methoxyphenyl)-5-phenylpentane-1,5-dione (3ac): White solid;



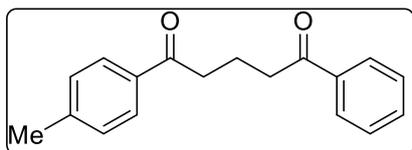
mp 78-80 °C; yield = 76%; $R_f = 0.61$ (ethyl acetate/hexane = 1:9); ^1H NMR (500 MHz, CDCl_3) δ 8.02-7.93 (m, 4H), 7.58-7.52 (m, 1H), 7.45 (t, $J = 7.5$ Hz, 2H), 6.93 (d, $J = 8.3$ Hz, 2H), 3.86 (s, 3H), 3.11 (t, $J = 6.9$ Hz, 2H), 3.06 (t, $J = 6.9$ Hz, 2H), 2.22-2.15 (m, 2H) ppm; ^{13}C NMR (125 MHz, CDCl_3) δ 199.9, 198.4, 163.4, 136.8, 133.0, 130.3, 129.9, 128.6, 128.0, 113.7, 55.4, 37.7, 37.2, 18.9 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{Na}]^+$ calcd for $\text{C}_{18}\text{H}_{18}\text{O}_3$ 305.1148, found 305.1120.

1-(4-Chlorophenyl)-5-phenylpentane-1,5-dione (3ad): White solid;



mp 70-72 °C; yield = 76%; $R_f = 0.67$ (ethyl acetate/hexane = 1:9); ^1H NMR (500 MHz, CDCl_3) δ 7.97 (d, $J = 7.3$ Hz, 2H), 7.92 (d, $J = 8.6$ Hz, 2H), 7.55 (t, $J = 7.4$ Hz, 1H), 7.48-7.40 (m, 4H), 3.11 (t, $J = 6.9$ Hz, 2H), 3.08 (t, $J = 7.0$ Hz, 2H), 2.22-2.15 (m, 2H) ppm; ^{13}C NMR (125 MHz, CDCl_3) δ 199.7, 198.6, 139.4, 136.8, 135.1, 133.1, 129.5, 128.9, 128.6, 128.0, 37.6, 37.4, 18.6 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{15}\text{ClO}_2$ 289.0833, found 289.0832.

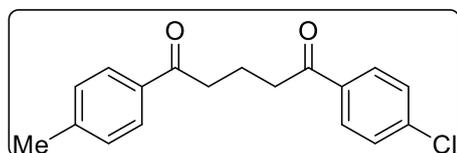
1-Phenyl-5-(4-methylphenyl)pentane-1,5-dione (3ba): White solid;



mp 64-66 °C; yield = 68%; $R_f = 0.71$ (ethyl acetate/hexane = 1:9); ^1H NMR (500 MHz, CDCl_3) δ 8.00-7.96 (m, 2H), 7.88 (d, $J = 8.2$ Hz, 2H), 7.55 (t, $J = 7.4$ Hz, 2H), 7.45 (t, $J = 7.7$ Hz, 2H), 7.25 (d, $J = 8.0$ Hz, 2H), 3.13-3.06 (m, 2H), 2.40 (s, 3H), 2.22-2.14 (m, 2H) ppm; ^{13}C NMR (125 MHz, CDCl_3) δ 199.9, 199.5, 143.8, 136.8, 134.4, 133.0, 129.2, 128.6, 128.2, 128.0, 37.6, 37.5,

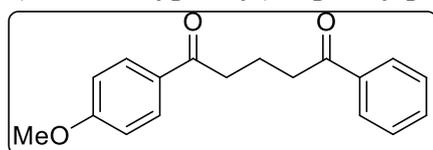
21.6, 18.8 ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{18}H_{18}O_2$ 267.1380, found 267.1378.

1-(4-Chlorophenyl)-5-(4-methylphenyl)pentane-1,5-dione (3bd):



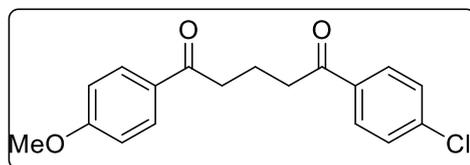
White solid; mp 74-76 °C; yield = 73%; R_f = 0.68 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 7.92 (d, J = 8.6 Hz, 2H), 7.87 (d, J = 8.2 Hz, 2H), 7.45-7.41 (m, 2H), 7.25 (d, J = 8.4 Hz, 2H), 3.08 (m, 4H), 2.41 (s, 3H), 2.21-2.15 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.4, 198.7, 143.9, 139.4, 135.1, 134.3, 129.5, 129.3, 128.9, 128.1, 37.6, 37.3, 21.6, 18.7 ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{18}H_{17}ClO_2$ 301.0990, found 301.1010.

1-(4-Methoxyphenyl)-5-phenylpentane-1,5-dione (3ca): White solid;



mp 80-82 °C; yield = 66%; R_f = 0.61 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 8.01-7.93 (m, 4H), 7.57-7.52 (m, 1H), 7.45 (t, J = 7.5 Hz, 2H), 6.93 (d, J = 8.3 Hz, 2H), 3.86 (s, 3H), 3.11 (t, J = 6.9 Hz, 2H), 3.06 (t, J = 6.9 Hz, 2H), 2.24-2.14 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.9, 198.4, 163.4, 136.8, 133.0, 130.3, 129.9, 128.6, 128.0, 113.7, 55.4, 37.7, 37.2, 18.9 ppm; HRMS (ESI-TOF): m/z $[M+Na]^+$ calcd for $C_{18}H_{18}O_3$ 305.1148, found 305.1120.

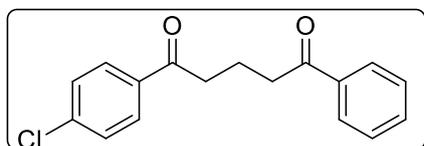
1-(4-Chlorophenyl)-5-(4-methoxyphenyl)pentane-1,5-dione (3cd):



White solid; mp 86-87 °C; yield = 64%; R_f = 0.60 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 7.96 (d, J = 8.8 Hz, 2H), 7.92 (d, J = 8.5 Hz, 2H), 7.43 (d, J = 8.5 Hz, 2H), 6.93 (d, J = 8.8 Hz, 2H), 3.87 (s, 3H), 3.10-3.03 (m, 4H), 2.22-2.13 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 198.7, 198.3, 163.5, 139.4, 135.1, 130.3, 129.9, 129.5, 128.9, 113.7, 55.4, 37.6, 37.1, 18.8

ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{18}H_{17}ClO_3$ 317.0939, found 317.0938.

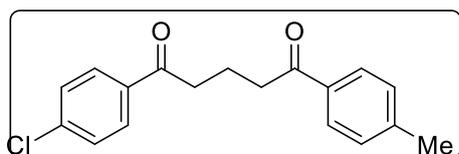
1-(4-Chlorophenyl)-5-phenylpentane-1,5-dione (3da): White solid;



mp 70-72 °C; yield = 77%; R_f = 0.67 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 7.97

(d, J = 7.3 Hz, 2H), 7.92 (d, J = 8.6 Hz, 2H), 7.55 (t, J = 7.4 Hz, 1H), 7.48-7.40 (m, 4H), 3.11 (t, J = 6.9 Hz, 2H), 3.08 (t, J = 7.0 Hz, 2H), 2.21-2.14 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.7, 198.6, 139.4, 136.8, 135.1, 133.1, 129.5, 128.9, 128.6, 128.0, 37.6 37.4, 18.6 ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{17}H_{15}ClO_2$ 289.0833, found 289.0832.

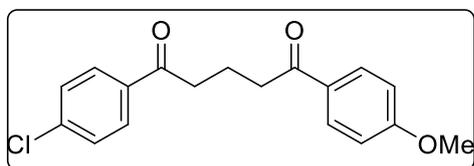
1-(4-Chlorophenyl)-5-(4-methylphenyl)pentane-1,5-dione(3db):



White solid; mp 68-70 °C; yield = 67%; R_f = 0.68 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 7.92 (d, J =

8.6 Hz, 2H), 7.87 (d, J = 8.2 Hz, 2H), 7.45-7.41 (m, 2H), 7.25 (d, J = 8.4 Hz, 2H), 3.11-3.05 (m, 4H), 2.41 (s, 3H), 2.21-2.14 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.4, 198.7, 143.9, 139.4, 135.1, 134.3, 129.5, 129.3, 128.9, 128.1, 37.6, 37.3, 21.6, 18.7 ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{18}H_{17}ClO_2$ 301.0990, found 301.1010.

1-(4-Chlorophenyl)-5-(4-methoxyphenyl)pentane-1,5-dione (3dc):

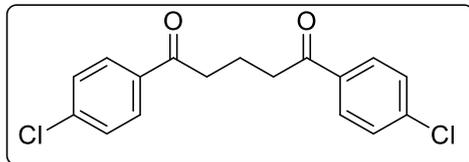


White solid; mp 86-88 °C; yield = 72%; R_f = 0.60 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 7.96 (d, J =

8.8 Hz, 2H), 7.92 (d, J = 8.5 Hz, 2H), 7.43 (d, J = 8.5 Hz, 2H), 6.93 (d, J = 8.8 Hz, 2H), 3.87 (s, 3H), 3.10-3.03 (m, 4H), 2.22-2.13 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 198.7, 198.3, 163.5, 139.4, 135.1, 130.3, 129.9, 129.5, 128.9, 113.7, 55.4, 37.6, 37.1, 18.8 ppm; HRMS

(ESI-TOF): m/z $[M+H]^+$ calcd for $C_{18}H_{17}ClO_3$ 317.0939, found 317.09365.

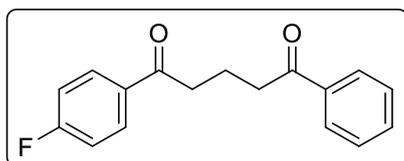
1,5-Bis(4-chlorophenyl)pentane-1,5-dione (3dd): White solid; mp 76-78 °C; Yield = 79%; R_f = 0.65 (Ethyl acetate/ Hexane = 1:9); 1H NMR



(500 MHz, $CDCl_3$) δ 7.92 (d, J = 8.6 Hz, 4H), 7.44 (d, J = 8.6 Hz, 4H), 3.09 (t, J = 6.9 Hz, 4H),

2.18 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 198.5, 139.6, 135.1, 129.4, 128.9, 37.4, 18.5 ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{17}H_{14}Cl_2O_2$ 321.0444; found 321.0440.

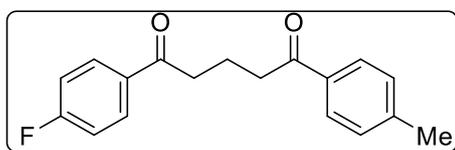
1-(4-Fluorophenyl)-5-phenylpentane-1,5-dione (3ea): White solid;



mp 60-62 °C; yield = 52%; R_f = 0.68 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 8.04-7.96 (m, 4H), 7.56 (t, J = 7.6 Hz, 1H),

7.46 (t, J = 7.8 Hz, 2H), 7.12 (t, J = 8.5 Hz, 2H), 3.15-3.06 (t, J = 6.9 Hz, 2H), 3.09 (t, J = 7.0 Hz, 2H), 2.23-2.15 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.7, 198.2, 165.7 (d, J_{C-F} = 254.6 Hz), 136.8, 133.2 (d, J_{C-F} = 3.2 Hz), 133.1 (s), 130.7 (d, J_{C-F} = 9.3 Hz), 128.6, 128.0, 115.6 (d, J_{C-F} = 21.7 Hz), 37.5, 37.5, 18.6 ppm; m/z $[M+H]^+$ calcd for $C_{17}H_{15}FO_2$ 271.1129, found 271.1133.

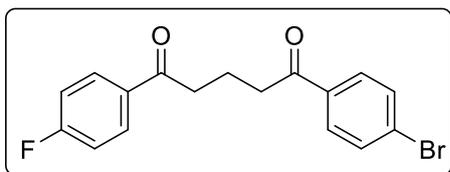
1-(4-Fluorophenyl)-5-(4-methylphenyl)pentane-1,5-dione (3eb):



White solid; mp 66-68 °C; yield = 57%; R_f = 0.69 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 8.03-7.96

(m, 2H), 7.86 (d, J = 8.1 Hz, 2H), 7.24 (d, J = 7.8 Hz, 2H), 7.14-7.06 (m, 2H), 3.11-3.02 (m, 4H), 2.39 (s, 3H), 2.20-2.12 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.4, 198.2, 165.7 (d, J = 254.5 Hz), 143.8, 134.3, 133.3 (d, J = 2.8 Hz), 130.7 (d, J = 9.3 Hz), 129.2, 128.1, 115.6 (d, J = 21.9 Hz), 37.5, 37.3, 21.6, 18.7 ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{18}H_{17}FO_2$ 285.1285, found 285.1287.

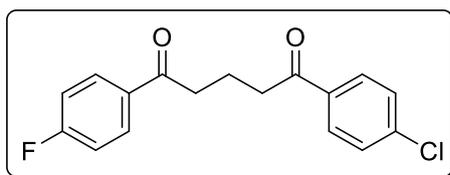
1-(4-Bromophenyl)-5-(4-fluorophenyl)pentane-1,5-dione (3ee):



White solid; mp 70-72 °C; yield = 48%; $R_f = 0.69$ (ethyl acetate/hexane = 1:9); $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 8.02-7.96

(m, 2H), 7.82 (d, $J = 8.5$ Hz, 2H), 7.57 (d, $J = 8.5$ Hz, 2H), 7.14-7.07 (m, 2H), 3.10-3.03 (m, 4H), 2.20-2.11 (m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 198.7, 198.0, 165.7 (d, $J = 254.8$ Hz), 135.5, 133.2 (d, $J = 3.0$ Hz), 131.9, 130.6 (d, $J = 9.3$ Hz), 129.5, 128.2, 115.7 (d, $J = 21.6$ Hz), 37.4, 37.3, 18.5 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{14}^{79}\text{BrO}_2$ 349.0234, found 349.0236; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{14}^{81}\text{BrO}_2$ 351.0214, found 351.0201.

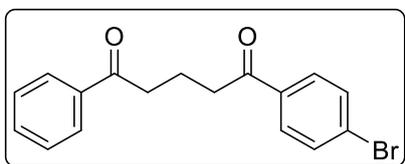
1-(4-chlorophenyl)-5-(4-fluorophenyl)pentane-1,5-dione (3ed):



White solid; mp 74-76 °C; yield = 48%; $R_f = 0.68$ (ethyl acetate/hexane = 1:9); $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 8.03-7.98

(m, 2H), 7.92 (d, $J = 8.5$ Hz, 2H), 7.43 (d, $J = 8.5$ Hz, 2H), 7.13 (t, $J = 8.6$ Hz, 2H), 3.11-3.06 (m, 4H), 2.21-2.15 (m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 198.5, 198.1, 165.7 (d, $J = 254.8$ Hz), 139.5, 135.1, 133.2 (d, $J = 2.9$ Hz), 130.7 (d, $J = 9.2$ Hz), 129.4, 128.9, 115.7 (d, $J = 22.0$ Hz), 37.5, 37.4, 18.5 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for 305.0739 found 305.0766.

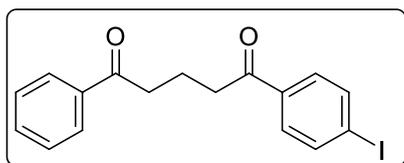
1-(4-Bromophenyl)-5-phenylpentane-1,5-dione (3ae): White solid;



mp 68-70 °C; yield = 62%; $R_f = 0.71$ (ethyl acetate/hexane = 1:9); $^1\text{H NMR}$ (500 MHz, CDCl_3) δ 7.92-7.87

(m, 2H), 7.78-7.74 (m, 2H), 7.54-7.46 (m, 3H), 7.38 (t, $J = 7.7$ Hz, 2H), 3.04 (t, $J = 6.9$ Hz, 2H), 3.00 (t, $J = 7.0$ Hz, 2H), 2.14-2.08 (m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 199.7, 198.8, 136.8, 135.5, 133.1, 131.9, 129.6, 128.6, 128.2, 128.0, 37.5, 37.4, 18.6 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{15}\text{BrO}_2$ 331.0328, found 331.0324.

1-(4-Iodophenyl)-5-phenylpentane-1,5-dione (3af): White solid; mp



66-68 °C; yield = 64%; R_f = 0.73

(ethyl acetate/hexane = 1:9); ^1H

NMR (500 MHz, CDCl_3) δ 7.99-

7.96 (m, 2H), 7.83-7.80 (m, 2H),

7.70-7.66 (m, 2H), 7.58-7.53 (m, 1H), 7.46 (t, J = 7.7 Hz, 2H), 3.11 (t,

J = 6.9 Hz, 2H), 3.07 (t, J = 7.0 Hz, 2H), 2.22-2.14 (m, 2H) ppm; ^{13}C

NMR (125 MHz, CDCl_3) δ 199.7, 199.1, 137.9, 136.8, 136.0, 133.1,

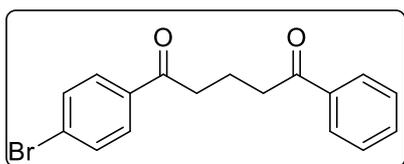
129.5, 128.6, 128.0, 101.0, 37.5, 37.4, 18.5 ppm; HRMS (ESI-TOF): m/z

$[\text{M}+\text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{15}^{79}\text{BrO}_2$ 331.0328, found 331.0321; HRMS

(ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{15}^{81}\text{BrO}_2$ 333.2287, found

333.2289.

1-(4-Bromophenyl)-5-phenylpentane-1,5-dione (3fa): White solid;



mp 68-70 °C; yield = 44%; R_f = 0.71

(Ethyl acetate/ Hexane = 1:9); ^1H

NMR (500 MHz, CDCl_3) δ 7.92-7.87

(m, 2H), 7.78-7.74 (m, 2H), 7.54-7.46

(m, 3H), 7.38 (t, J = 7.7 Hz, 2H), 3.04 (t, J = 6.9 Hz, 2H), 3.00 (t, J =

7.0 Hz, 2H), 2.14-2.07 (m, 2H) ppm; ^{13}C **NMR (125 MHz, CDCl_3)** δ

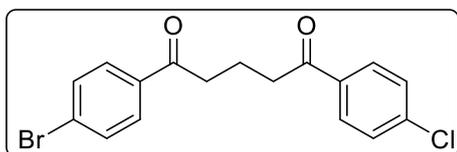
199.7, 198.8, 136.8, 135.5, 133.1, 131.9, 129.6, 128.6, 128.2, 128.0,

37.5, 37.4, 18.6 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for

$\text{C}_{17}\text{H}_{15}^{79}\text{BrO}_2$ 331.0328, found 331.0321; HRMS (ESI-TOF): m/z

$[\text{M}+\text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{15}^{81}\text{BrO}_2$ 333.2287, found 333.2289.

1-(4-Bromophenyl)-5-(4-chlorophenyl)pentane-1,5-dione (3fd):



White solid; mp 72-74 °C; yield

= 51%; R_f = 0.70 (ethyl

acetate/hexane = 1:9); ^1H **NMR**

(500 MHz, CDCl_3) δ 7.92 (d, J =

8.6 Hz, 2H), 7.84 (d, J = 8.6 Hz, 2H), 7.60 (d, J = 8.6 Hz, 2H), 7.43 (d,

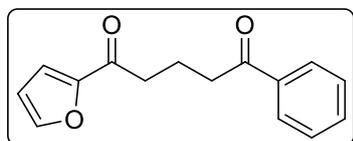
J = 8.6 Hz, 2H), 3.10-3.05 (m, 4H), 2.21-2.14 (m, 2H) ppm; ^{13}C **NMR**

(125 MHz, CDCl_3) δ 198.7, 198.5, 139.5, 135.5, 135.1, 131.9, 129.6,

129.4, 128.9, 128.3, 37.4, 18.4 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$

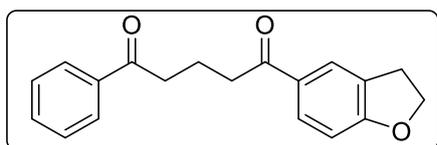
calcd for $C_{17}H_{14}^{79}BrClO_2$ 364.9938, found 368.9960; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{17}H_{14}^{81}BrClO_2$ 365.9845, found 365.9821.

1-(Furan-2-yl)-5-phenylpentane-1,5-dione (3ha): White solid; mp 72-



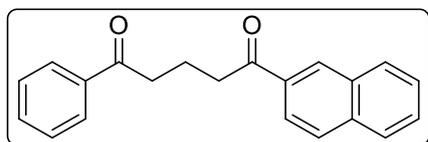
74 °C; yield = 69%; R_f = 0.65 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 8.00-7.95 (m, 2H), 7.58-7.53 (m, 2H), 7.46 (t, J = 7.7 Hz, 2H), 7.22 (d, J = 3.5 Hz, 1H), 6.54-6.51 (m, 1H), 3.10 (t, J = 7.0 Hz, 2H), 2.97 (t, J = 7.1 Hz, 2H), 2.22-2.14 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.7, 189.0, 152.6, 146.3, 136.8, 133.0, 128.6, 128.0, 117.1, 112.2, 37.5, 37.4, 18.6 ppm; m/z $[M+H]^+$ calcd for $C_{15}H_{14}O_3$ 243.1016, found 213.1004.

1-(2,3-Dihydrobenzofuran-5-yl)-5-phenylpentane-1,5-dione (3ak):



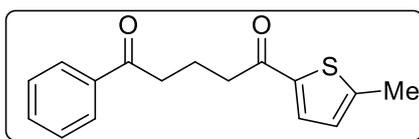
White solid; mp 82-84 °C; yield = 59%; R_f = 0.66 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 7.94-7.85 (m, 2H), 7.78 (s, 1H), 7.74 (d, J = 8.3 Hz, 1H), 7.51-7.43 (m, 1H), 7.40-7.34 (m, 2H), 6.73-6.69 (m, 1H), 4.60-4.53 (m, 2H), 3.18-3.11 (m, 2H), 3.05-2.99 (m, 2H), 2.98-2.93 (m, 2H), 2.14-2.06 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.9, 198.3, 164.3, 136.8, 133.0, 130.3, 130.0, 128.6, 128.0, 127.6, 125.4, 72.1, 37.7, 37.3, 29.0, 19.0 ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{19}H_{18}O_3$ 295.1329, found 295.1314.

1-(Naphthalen-2-yl)-5-phenylpentane-1,5-dione (3ai): White solid;



mp 72-73 °C; yield = 56%; R_f = 0.77 (ethyl acetate/hexane = 1:9); 1H NMR (500 MHz, $CDCl_3$) δ 8.41 (s, 1H), 7.98-7.83 (m, 4H), 7.81-7.74 (m, 2H), 7.47 (m, 3H), 7.36 (t, J = 7.5 Hz, 2H), 3.16 (t, J = 6.9 Hz, 2H), 3.06 (t, J = 6.8 Hz, 2H), 2.21-2.13 (m, 2H) ppm; ^{13}C NMR (125 MHz, $CDCl_3$) δ 199.9, 199.8, 136.8, 135.6, 134.1, 133.1, 132.5, 129.8, 129.6, 128.6, 128.4, 128.1, 127.7, 126.7, 123.8, 37.6, 37.6, 18.8 ppm; HRMS (ESI-TOF): m/z $[M+H]^+$ calcd for $C_{21}H_{18}O_2$ 303.1080, found 303.1081.

1-(5-Methylthiophen-2-yl)-5-phenylpentane-1,5-dione (3aj): White



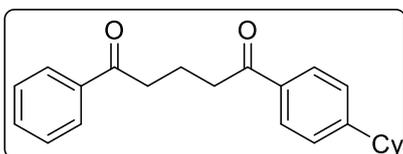
solid; mp 70-72 °C; yield = 71%; R_f

= 0.69 (ethyl acetate/hexane = 1:9);

$^1\text{H NMR}$ (500 MHz, CDCl_3) δ 7.89

(d, J = 7.3 Hz, 2H), 7.49-7.44 (m, 2H), 7.37 (t, J = 7.7 Hz, 2H), 6.72-6.69 (m, 1H), 3.02 (t, J = 7.0 Hz, 2H), 2.90 (t, J = 7.0 Hz, 2H), 2.44 (s, 3H), 2.10 (m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 199.8, 192.5, 149.6, 142.0, 136.8, 133.0, 132.6, 128.6, 128.0, 126.8, 37.8, 37.5, 19.2, 16.0 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{16}\text{H}_{16}\text{O}_2\text{S}$ 273.0944, found 273.0941.

1-(4-Cyclohexylphenyl)-5-phenylpentane-1,5-dione (3ag): White



solid; mp 62-64 °C; yield = 61%; R_f =

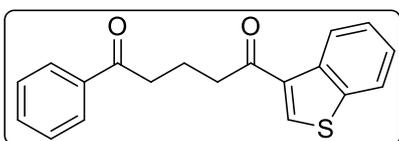
0.72 (ethyl acetate/hexane = 1:9); ^1H

NMR (500 MHz, CDCl_3) δ 8.00-

7.96 (m, 2H), 7.91 (d, J = 8.3 Hz, 2H),

7.55 (t, J = 7.4 Hz, 1H), 7.46 (t, J = 7.6 Hz, 2H), 7.29 (d, J = 8.2 Hz, 2H), 3.13-3.07 (m, 4H), 2.25-2.15 (m, 2H), 1.94-1.72 (m, 6H), 1.50-1.34 (m, 5H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 199.9, 199.5, 153.7, 136.8, 134.7, 133.0, 128.6, 128.3, 128.0, 127.0, 44.6, 37.6, 37.5, 34.1, 26.7, 26.0, 18.8 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{23}\text{H}_{26}\text{O}_2$ 335.2006, found 335.2021.

1-(Benzo[*b*]thiophen-3-yl)-5-phenylpentane-1,5-dione (3al): White



solid; mp 76-78 °C; yield = 59%; R_f =

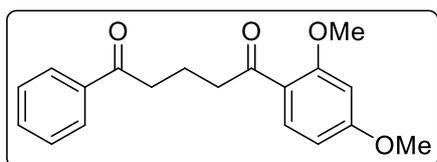
0.70 (ethyl acetate/hexane = 1:9); ^1H

NMR (500 MHz, CDCl_3) δ 8.78 (d, J

= 8.2 Hz, 1H), 8.40 (s, 1H), 7.99 (d, J

= 7.3 Hz, 2H), 7.87 (d, J = 8.0 Hz, 1H), 7.56 (t, J = 7.3 Hz, 1H), 7.52-7.39 (m, 4H), 3.19-3.11 (m, 4H), 2.30-2.22 (m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 199.9, 195.2, 139.8, 137.0, 136.8, 136.6, 135.0, 133.1, 128.6, 128.0, 125.8, 125.6, 125.4, 122.2, 39.2, 37.5, 19.2 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{19}\text{H}_{16}\text{O}_2\text{S}$ 309.0944, found 309.0924.

1-(2,4-Dimethoxyphenyl)-5-phenylpentane-1,5-dione (3am): White



solid; mp 94-96 °C; yield = 67%;

$R_f = 0.41$ (ethyl acetate/hexane =

1:9); $^1\text{H NMR}$ (500 MHz, CDCl_3)

δ 7.98 (d, $J = 7.4$ Hz, 2H), 7.82 (d,

$J = 8.7$ Hz, 1H), 7.55 (t, $J = 7.3$ Hz, 1H), 7.45 (t, $J = 7.6$ Hz, 2H), 6.54-

6.50 (m, 1H), 6.45-6.43 (m, 1H), 3.86 (s, 3H), 3.85 (s, 3H), 3.10-3.04

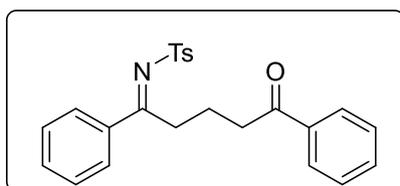
(m, 4H), 2.17-2.10 (m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 200.3,

199.9, 164.4, 160.8, 136.9, 132.9, 132.6, 128.5, 128.1, 121.0, 105.1,

98.3, 55.5, 55.4, 42.6, 38.0 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd

for $\text{C}_{19}\text{H}_{20}\text{O}_4$ 313.1434, found 313.1437.

4-Methyl-N-(5-oxo-1,5-diphenylpentylidene) (3aa'): White solid;



yield = 23%; $R_f = 0.63$ (ethyl

acetate/hexane = 1:9); $^1\text{H NMR}$ (500

MHz, CDCl_3) δ 8.05 (d, $J = 5.0$ Hz,

2H), 7.99 (d, $J = 7.5$ Hz, 2H), 7.93 (d,

$J = 7.9$ Hz, 2H), 7.60-7.51 (m, 2H), 7.51-7.41 (m, 4H), 7.35 (d, $J = 7.8$

Hz, 2H), 3.53 (s, 2H), 3.23 (t, $J = 6.2$ Hz, 2H), 2.45 (s, 3H), 2.27-2.18

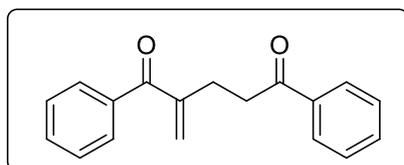
(m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 199.4, 183.0, 143.5,

138.8, 136.7, 133.2, 129.4, 128.9, 128.8, 128.8, 128.5, 128.0, 127.1,

38.0, 33.5, 22.8, 21.6 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for

$\text{C}_{24}\text{H}_{23}\text{NO}_3\text{S}$ 406.1471, found 406.1468.

2-Methylene-1,5-diphenylpentane-1,5-dione (4aa): White solid; $R_f =$



0.74 (ethyl acetate/hexane = 1:9); ^1H

NMR (500 MHz, CDCl_3) δ 8.00 –

7.96 (m, 2H), 7.76-7.72 (m, 2H),

7.59-7.51 (m, 2H), 7.45 (m, 4H),

5.97 (s, 1H), 5.68 (s, 1H), 3.25 (t, $J = 7.3$ Hz, 2H), 2.92 (t, $J = 7.3$ Hz,

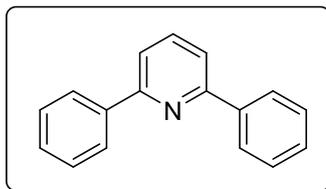
2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 199.29, 198.1, 146.8, 137.7,

136.7, 133.1, 132.2, 129.5, 128.6, 128.2, 128.1, 127.3, 37.2, 27.4 ppm;

HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{18}\text{H}_{16}\text{O}_2$ 265.1223, found

263.1210.

2,6-Diphenylpyridine (6aa): Crystalline white solid; yield = 84%; mp



92-94 °C; R_f = 0.45 (ethyl acetate/hexane

= 1:19); $^1\text{H NMR}$ (500 MHz, CDCl_3) δ

8.07 (d, J = 7.3 Hz, 4H), 7.72 (t, J = 7.7

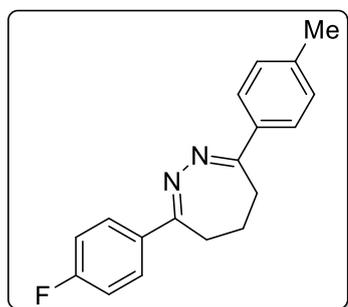
Hz, 1H), 7.60 (d, J = 7.7 Hz, 2H), 7.41 (t,

J = 7.2 Hz, 4H), 7.37-7.32 (m, 2H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3)

δ 156.8, 139.5, 137.5, 129.0, 128.7, 127.0, 118.6 ppm; HRMS (ESI-

TOF): m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{17}\text{H}_{13}\text{N}$ 232.1127, found 232.1133.

3-(4-Fluorophenyl)-7-(4-methylphenyl)-5,6-dihydro-4H-1,2-



diazepine(7aa): White solid; Yield =

77%; mp 68-70 °C; R_f = 0.34 (ethyl

acetate/hexane = 1:10); $^1\text{H NMR}$ (500

MHz, CDCl_3) δ 7.87 (m, 2H), 7.78 (d, J

= 7.9 Hz, 2H), 7.30 – 7.23 (m, 2H), 7.15-

7.09 (m, 2H), 2.71-2.62 (m, 4H), 2.50-

2.44 (m, 2H), 2.40 (s, 3H) ppm; $^{13}\text{C NMR}$ (125 MHz, CDCl_3) δ 164.8,

162.8, 159.0 (d, J = 121.0 Hz), 140.2, 134.0, 133.2 (d, J = 3.1 Hz), 129.4,

128.7 (d, J = 8.4 Hz), 126.7, 115.6 (d, J = 21.8 Hz), 32.7, 26.5, 26.4,

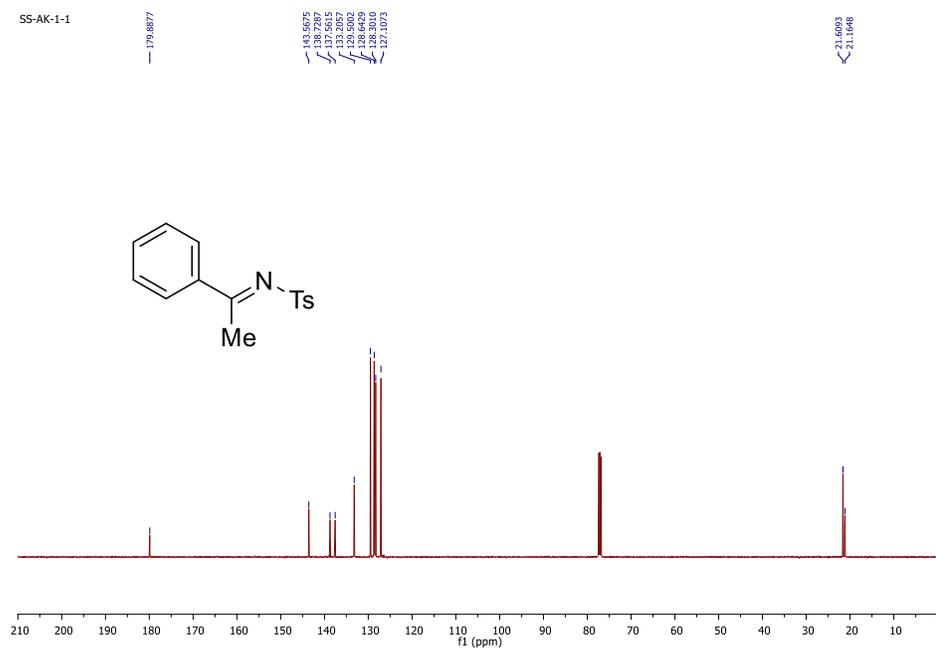
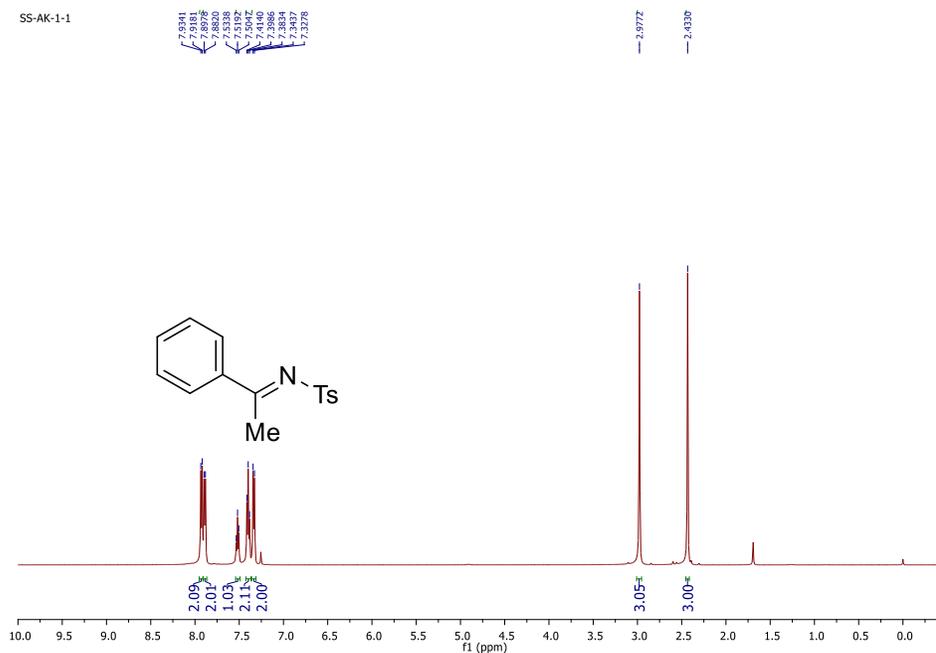
21.3 ppm; HRMS (ESI-TOF): m/z $[\text{M}+\text{H}]^+$ calcd for 289.1449, found

289.1458.

Conclusion

Finally, we have developed a simple, one-pot synthesis of symmetrical and unsymmetrical 1,5-diarylpropane-1,5-diones from several N-sulfonyl ketimines and 3-chloropropiophenones in the presence of Cs_2CO_3 . The method has many positive features such as good yields, transition metal-free and good substrate scope.

Appendix A



SS-AK-1-14

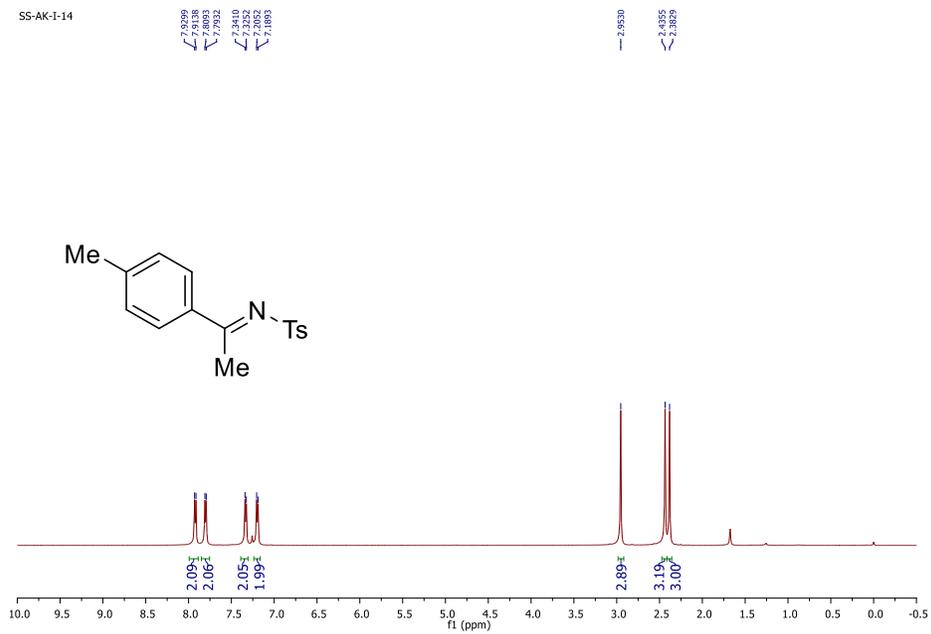


Figure 3: ^1H NMR spectrum (500MHz) of **1ab** in CDCl_3

SS-AK-1-14

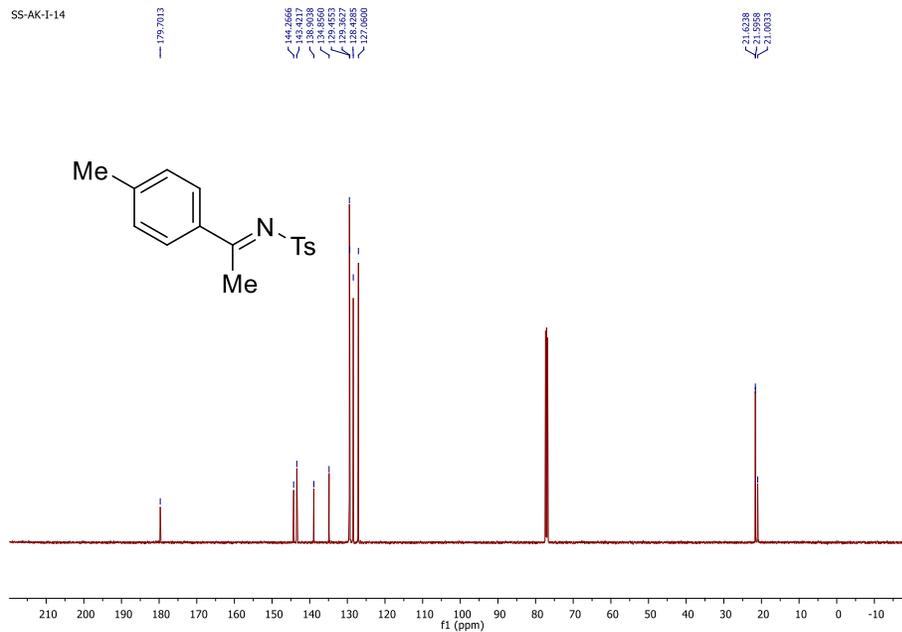


Figure 4: ^{13}C NMR spectrum (125MHz) of **1ab** in CDCl_3

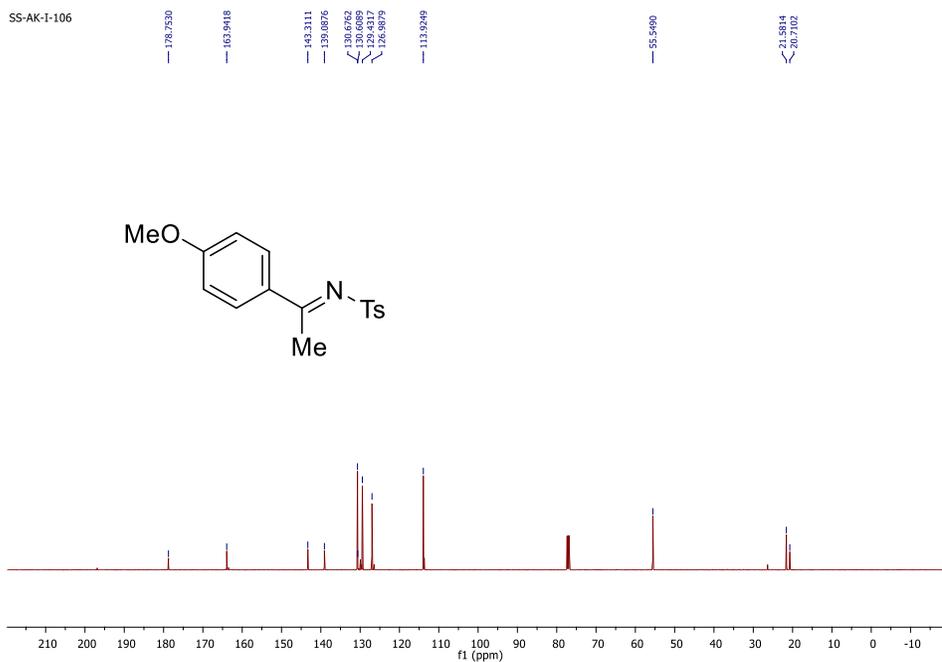


Figure 5: ^{13}C NMR spectrum (500MHz) of **1ac** in CDCl_3

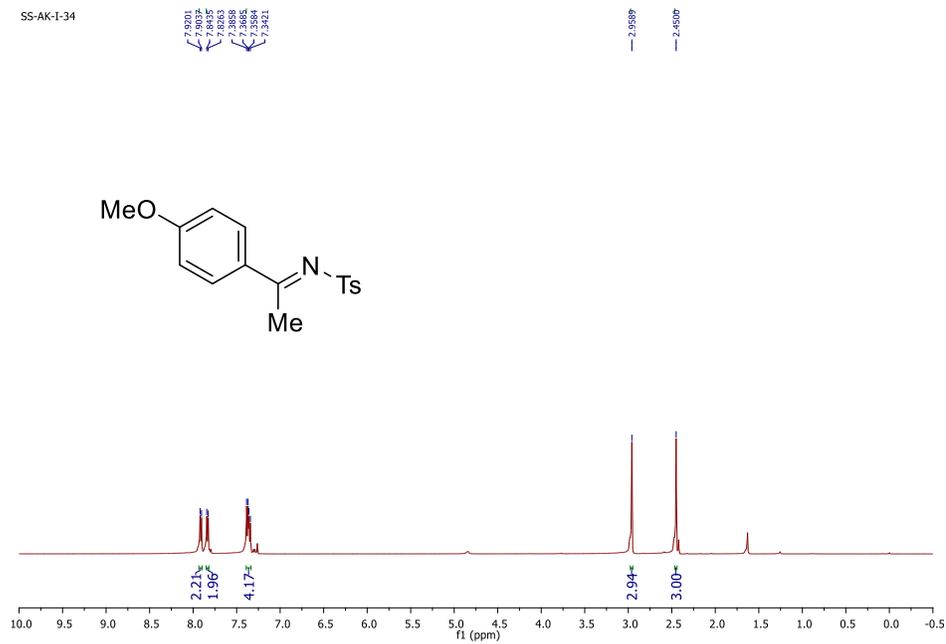


Figure 6: ^1H NMR spectrum (125MHz) of **1ac** in CDCl_3

SS-AK-I-71

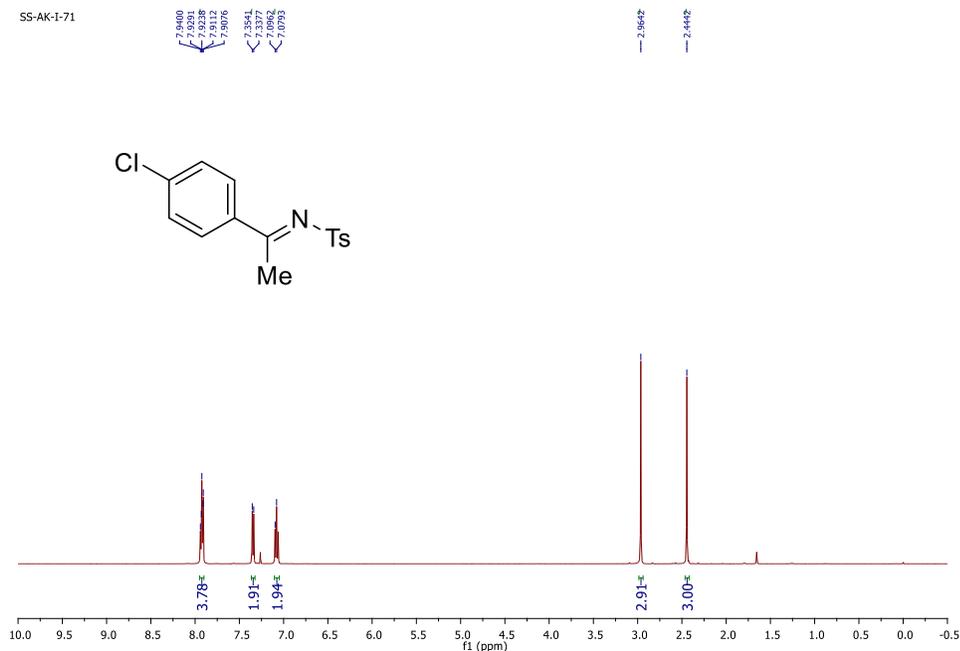


Figure 7: ^1H NMR spectrum (500MHz) of **1ad** in CDCl_3

SS-AK-I-34

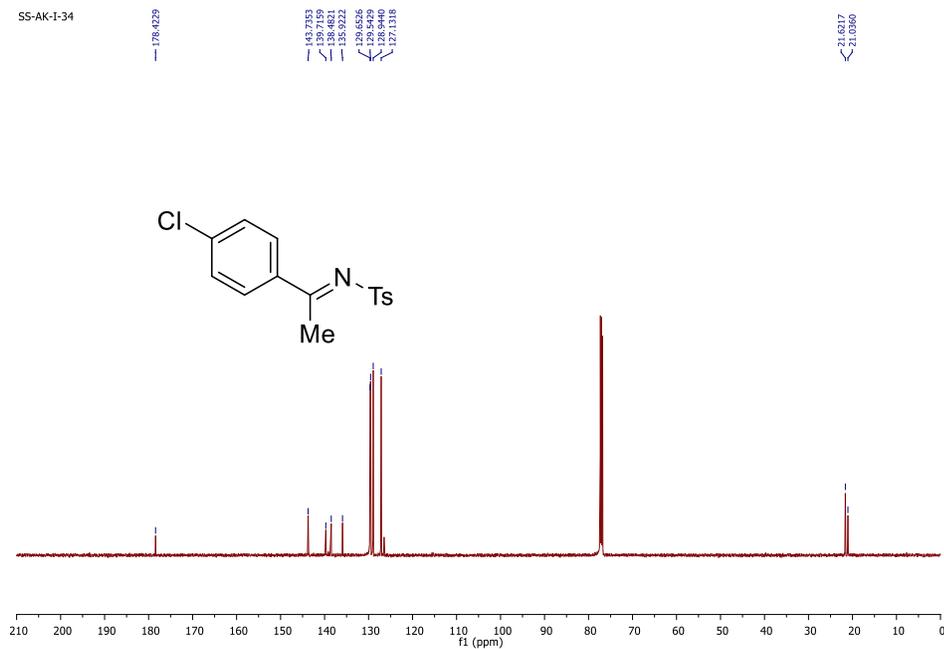


Figure 8: ^{13}C NMR spectrum (125MHz) of **1ad** in CDCl_3

SS-AK-I-113

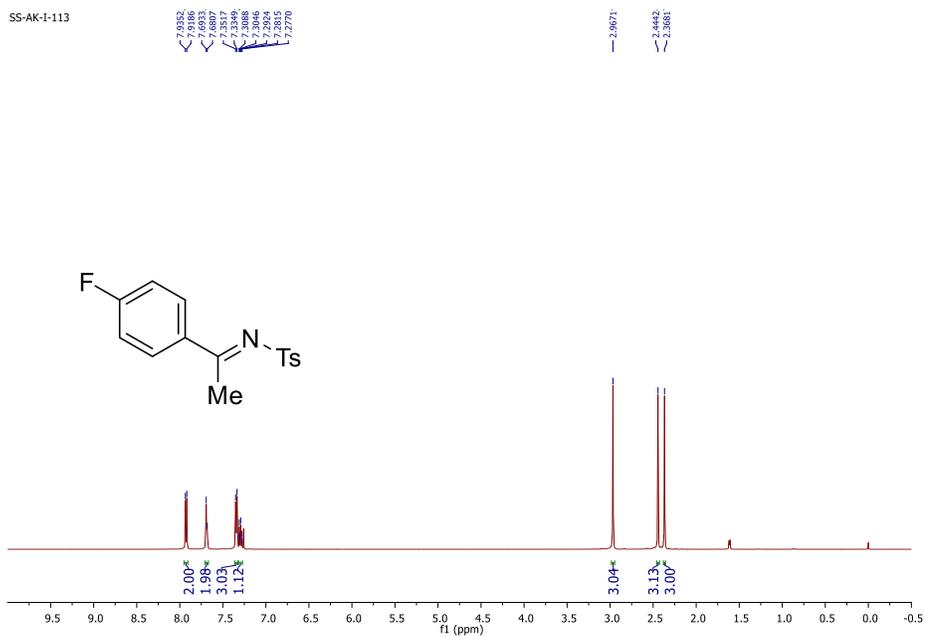


Figure 9: ¹H NMR spectrum (500MHz) of 1ae in CDCl₃

SS-AK-I-71

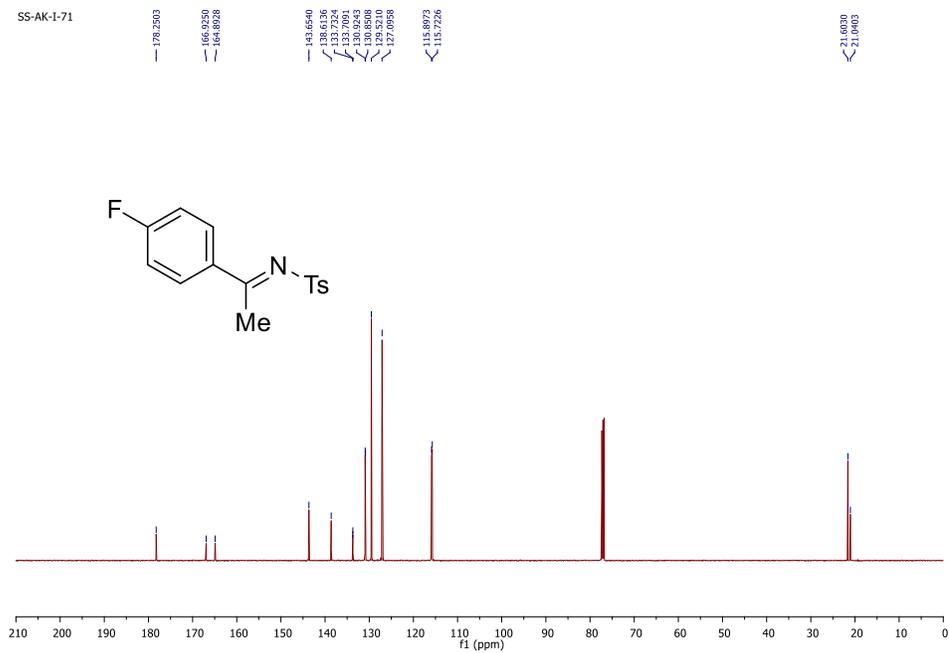


Figure 10: ¹³C NMR spectrum (125MHz) of 1ae in CDCl₃

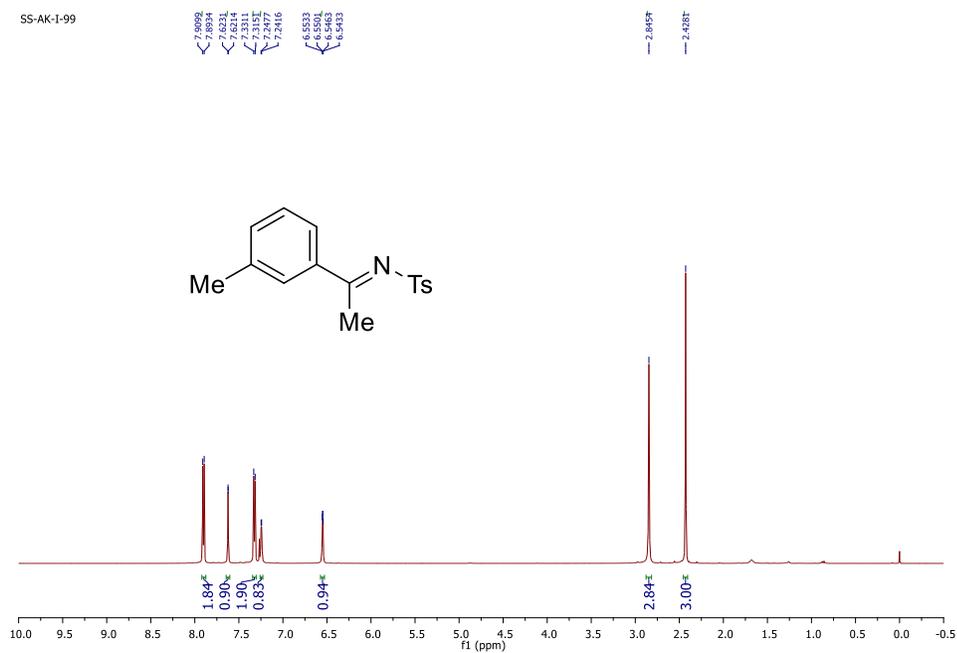


Figure 11: ^1H NMR spectrum (500MHz) of **1af** in CDCl_3

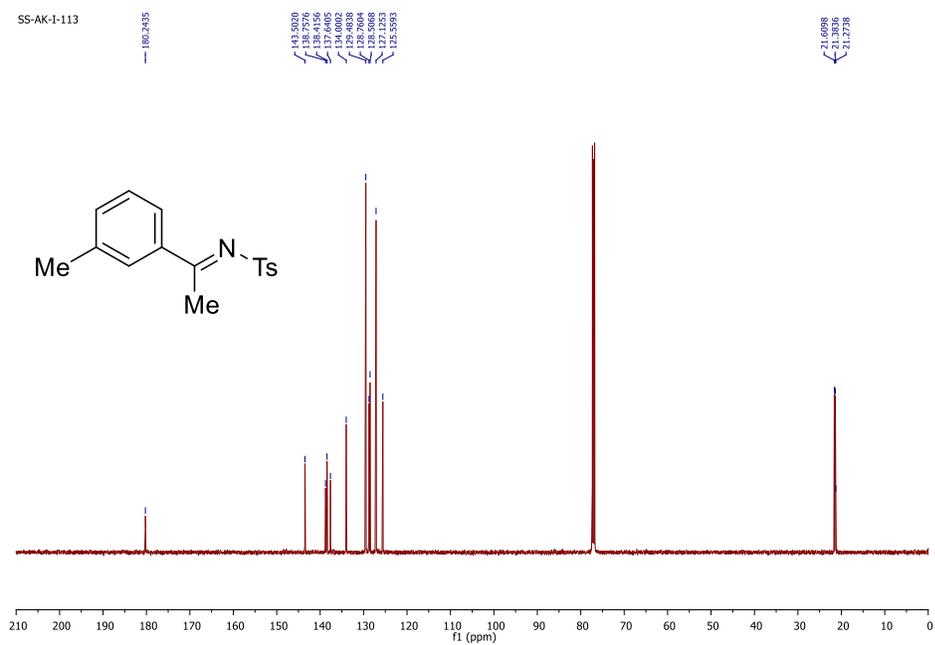


Figure 12: ^{13}C NMR spectrum (125MHz) of **1af** in CDCl_3

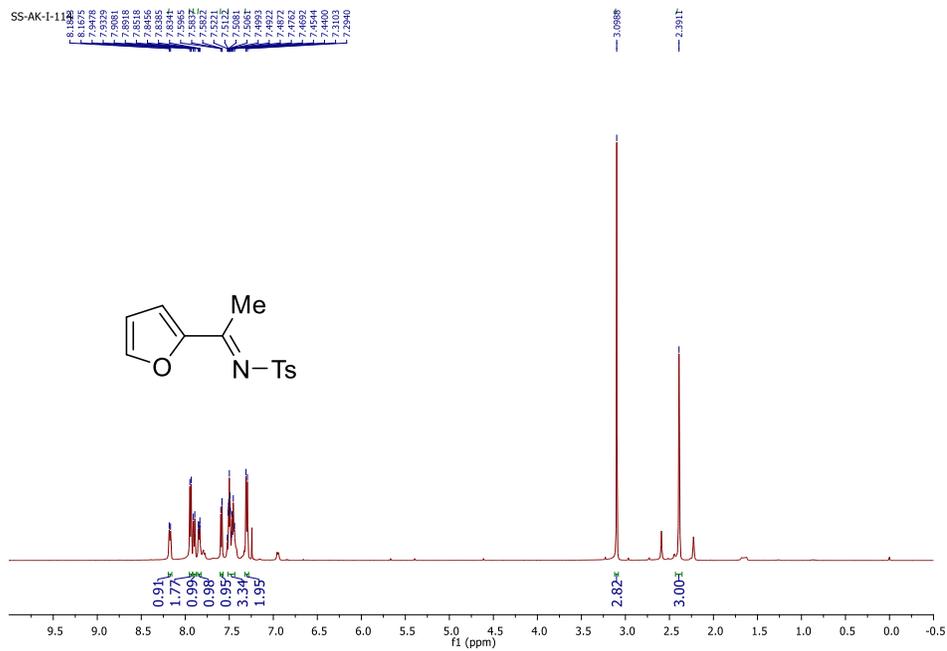


Figure 13: ^1H NMR spectrum (500MHz) of **1ag** in CDCl_3

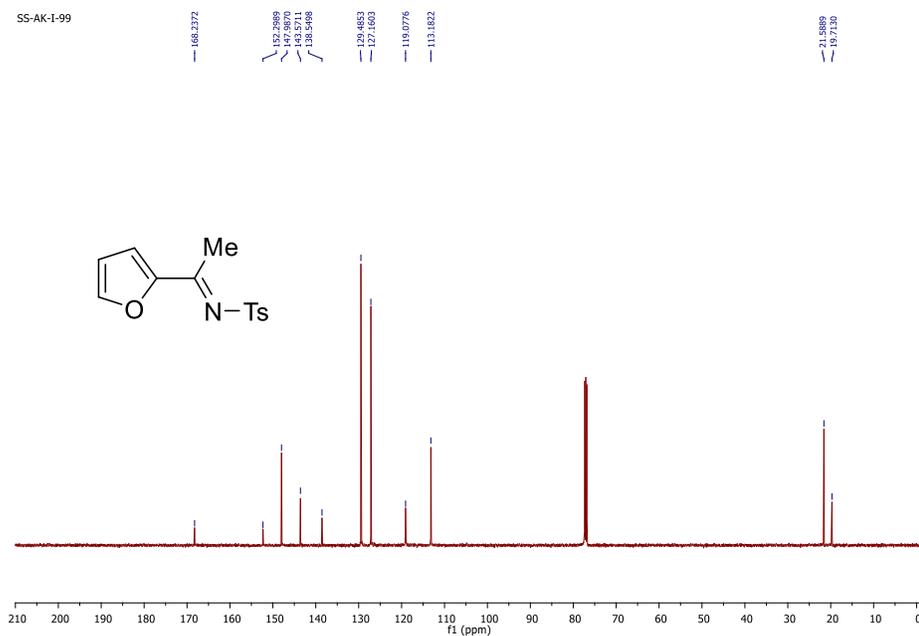


Figure 14: ^{13}C NMR spectrum (125MHz) of **1ag** in CDCl_3

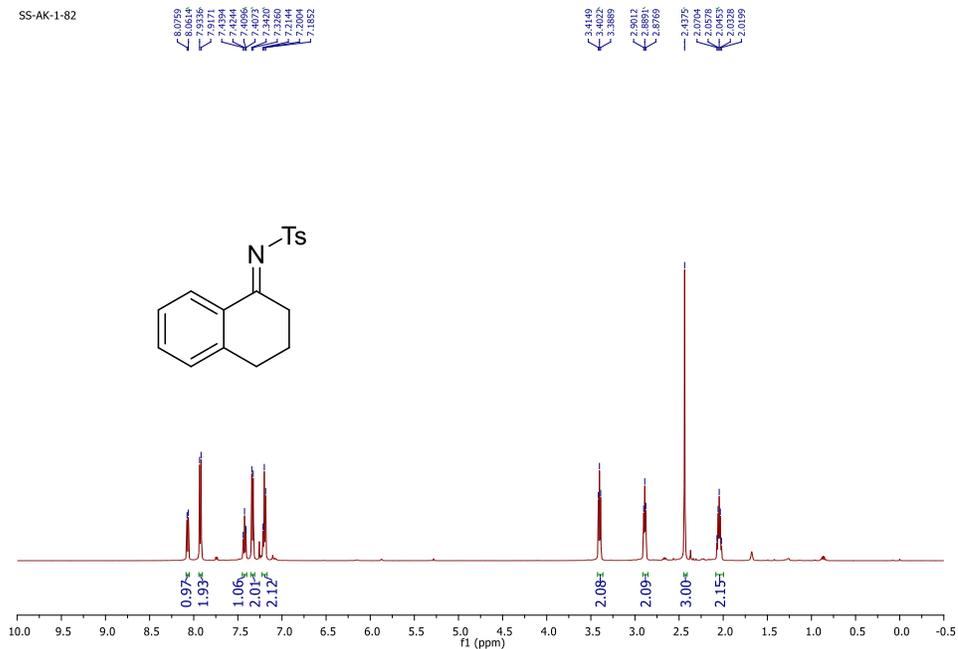


Figure 17: ^1H NMR spectrum (500MHz) of **1ai** in CDCl_3

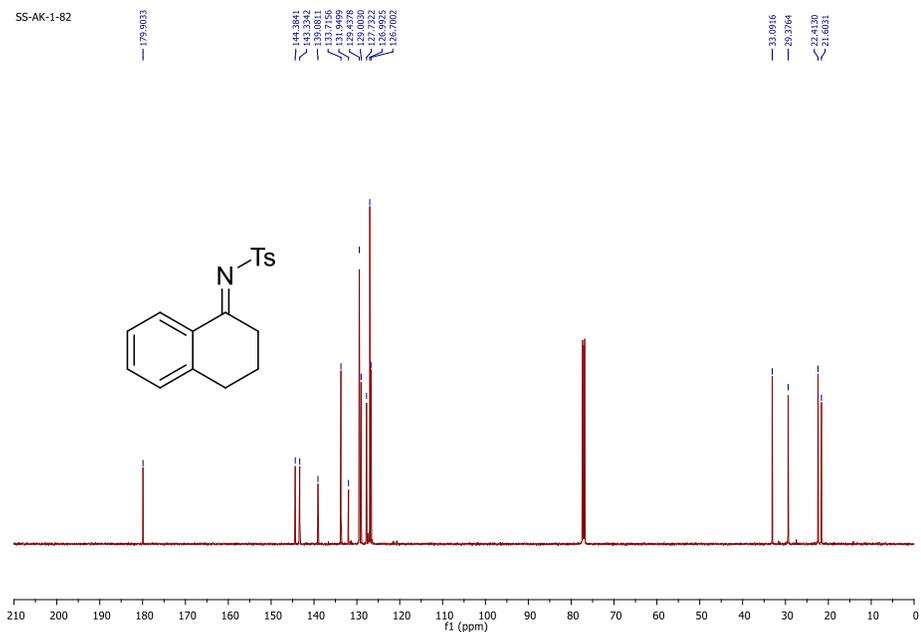


Figure 18: ^{13}C NMR spectrum (125MHz) of **1ai** in CDCl_3

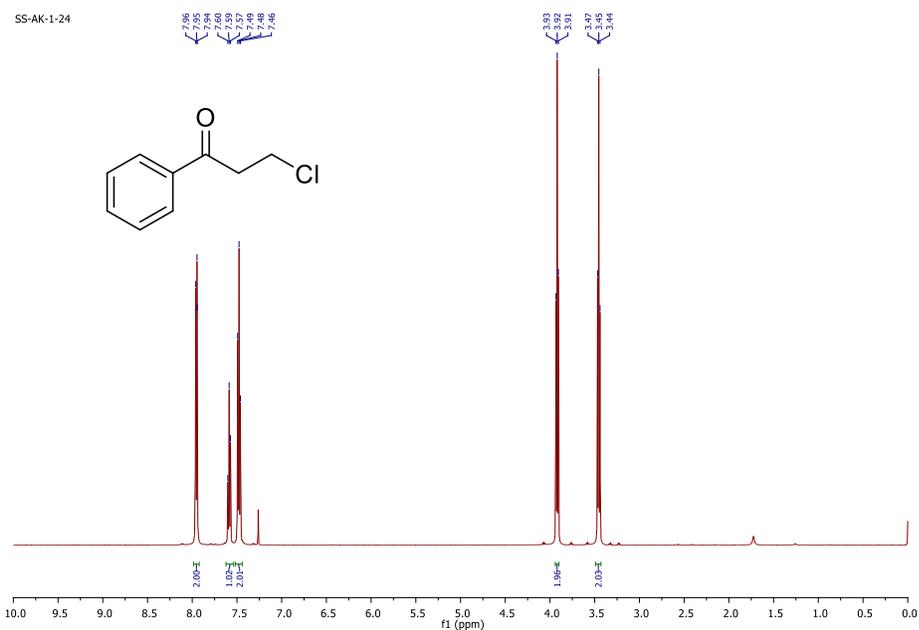


Figure 19: ^1H NMR spectrum (500MHz) of **2aa** in CDCl_3

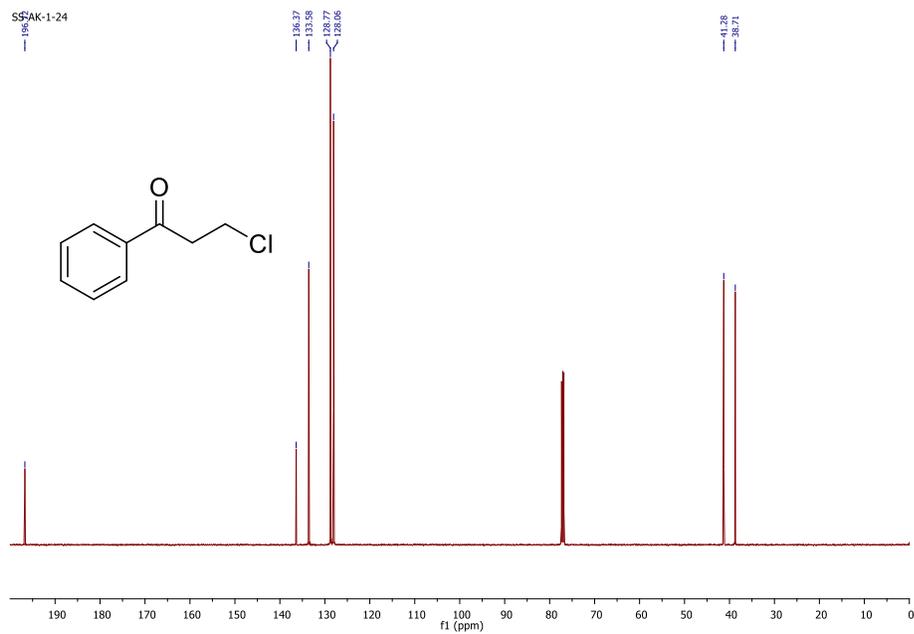


Figure 20: ^{13}C NMR spectrum (125MHz) of **2aa** in CDCl_3

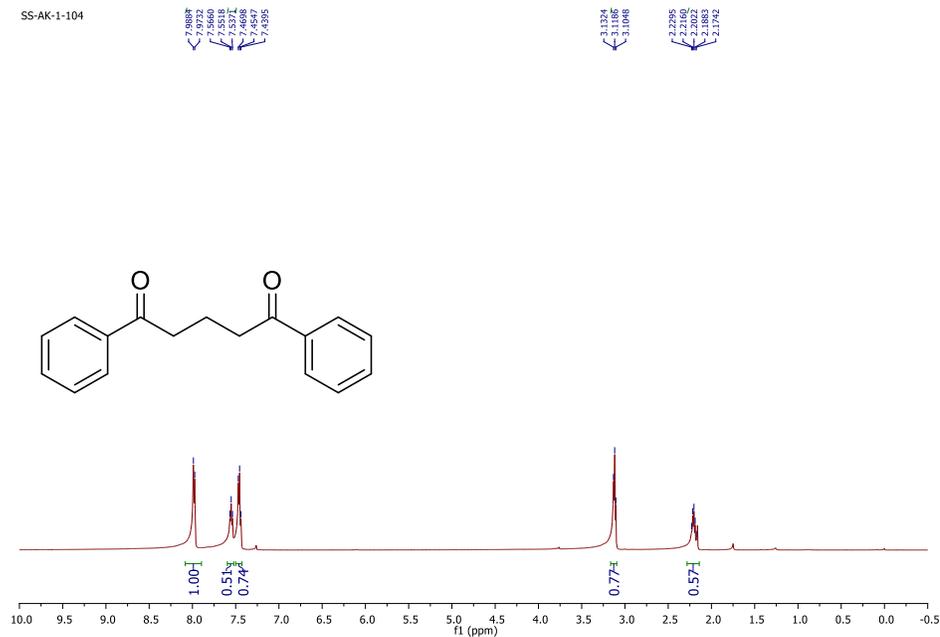


Figure 21: ^1H NMR spectrum (500MHz) of **3aa** in CDCl_3

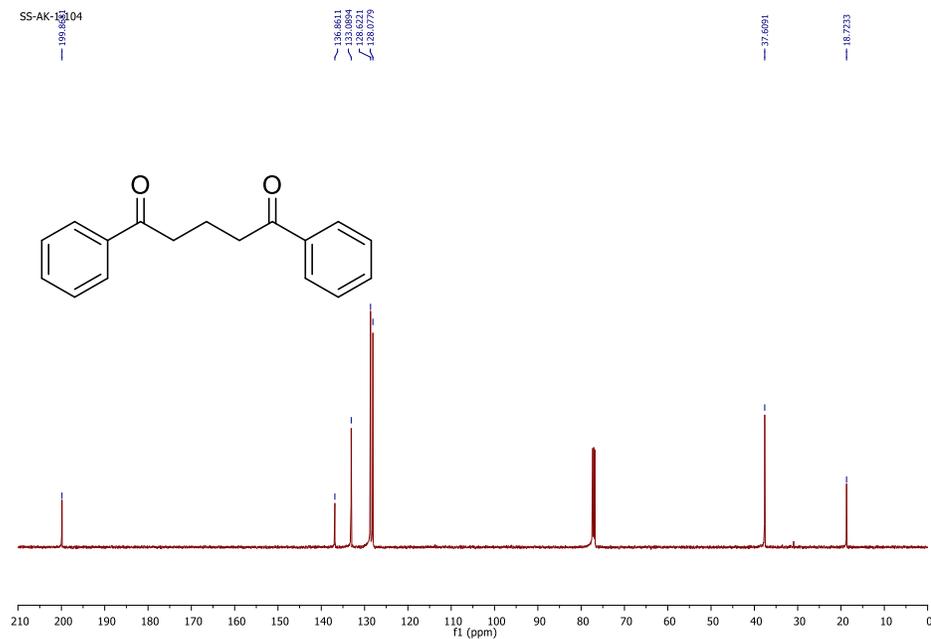


Figure 22: ^{13}C NMR spectrum (125MHz) of **3aa** in CDCl_3

SS-AK-1-33

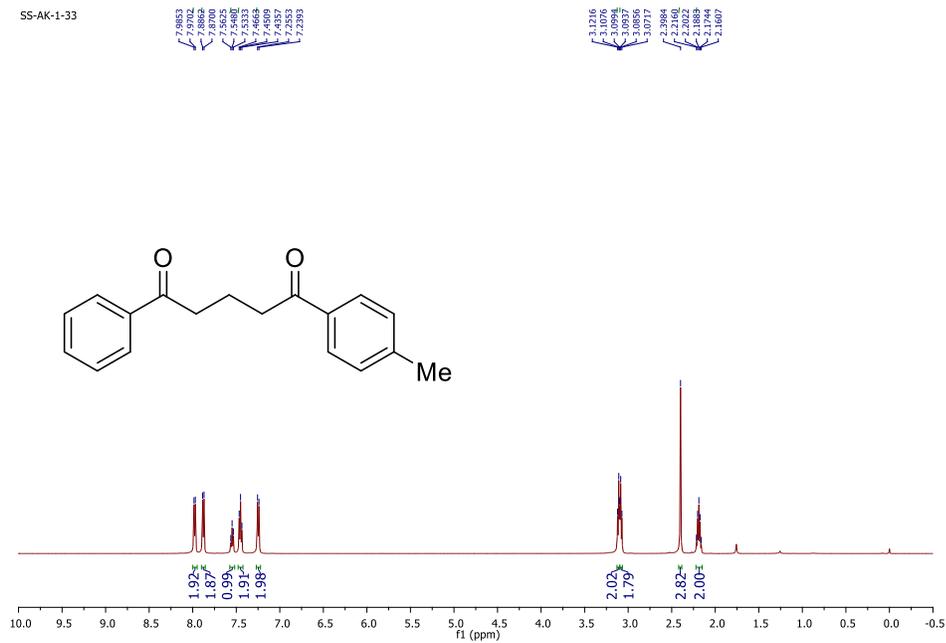


Figure 23: ¹H NMR spectrum (500MHz) of 3ab in CDCl₃

SS-AK-1-33

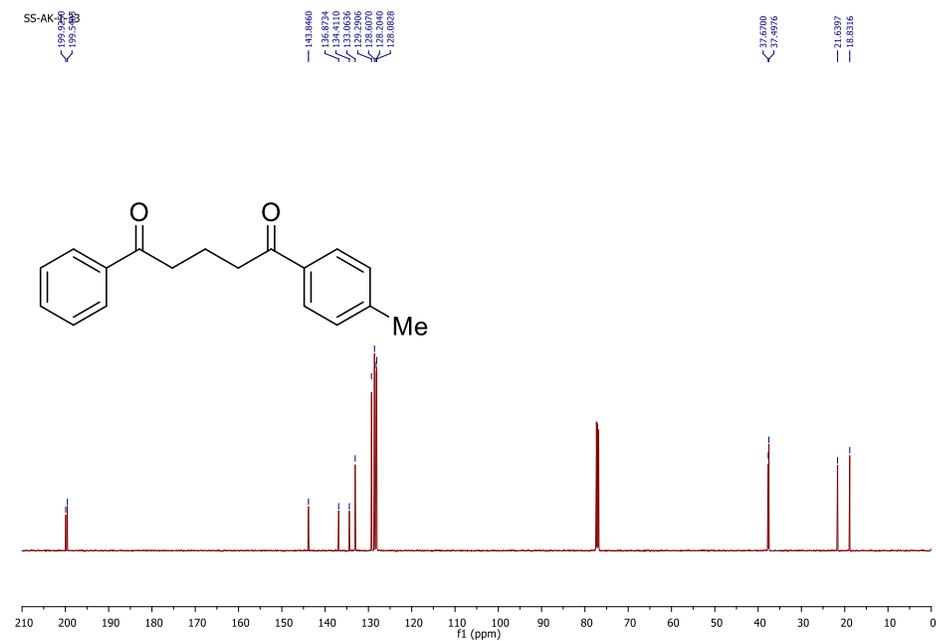


Figure 24: ¹³C NMR spectrum (125MHz) of 3ab in CDCl₃

SS-AK-I-9

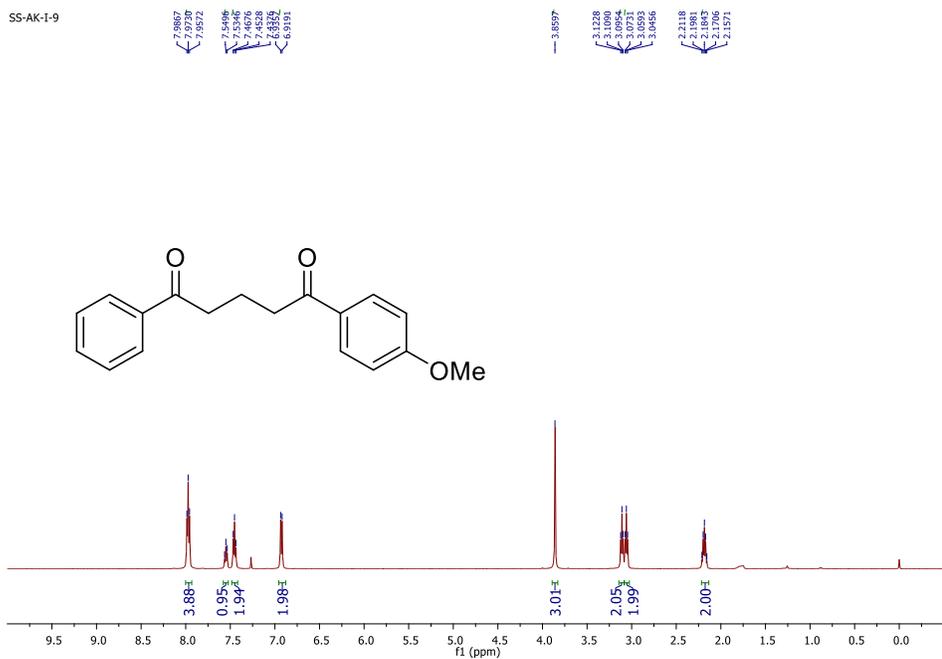


Figure 25: ^1H NMR spectrum (500MHz) of 3ac in CDCl_3

SS-AK-I-10

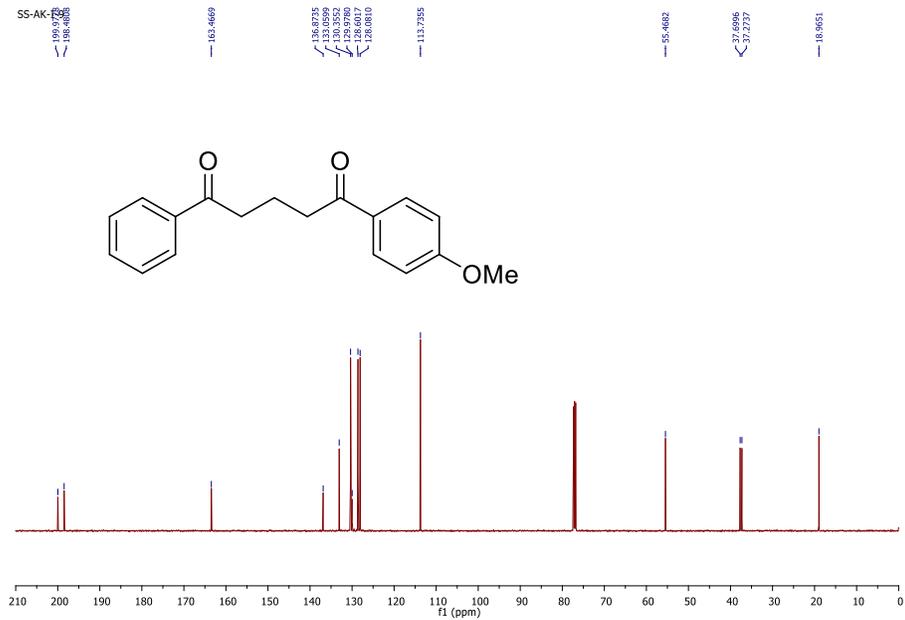


Figure 26: ^{13}C NMR spectrum (125MHz) of 3ac in CDCl_3

SS-AK-1-26

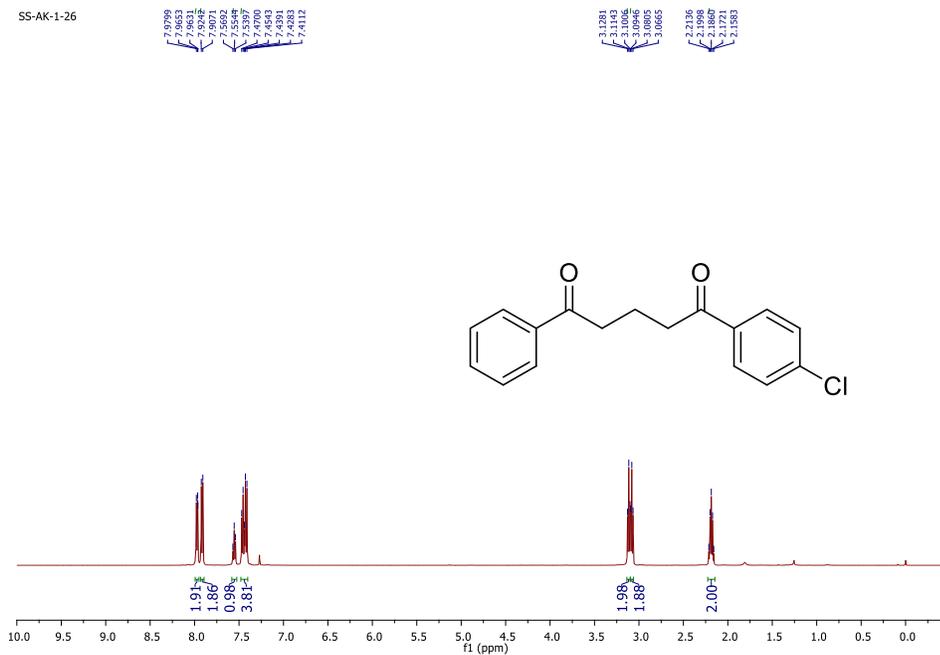


Figure 27: ¹H NMR spectrum (500MHz) of 3ad in CDCl₃

SS-AK-1-26
199.7471
198.628

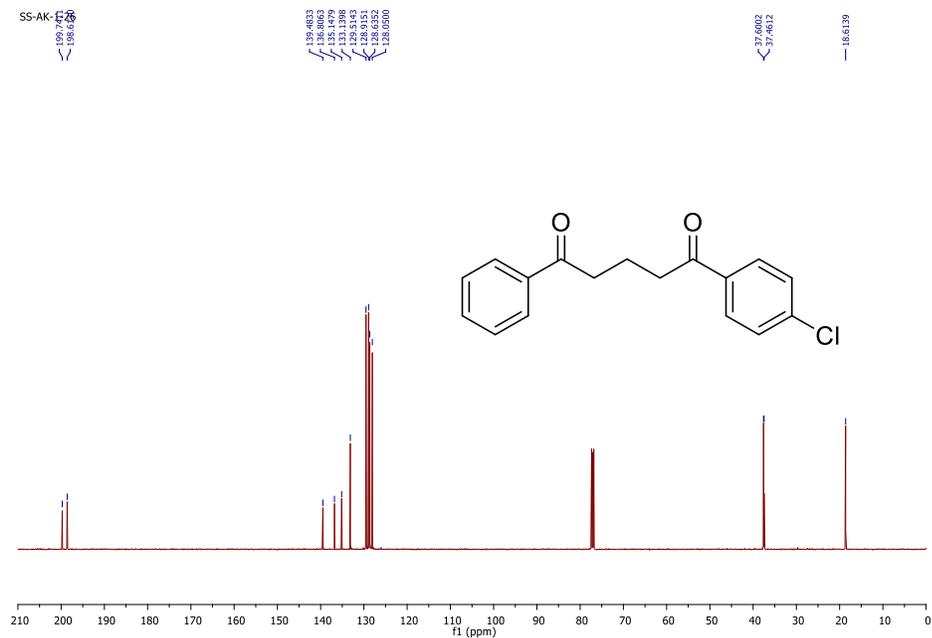


Figure 28: ¹³C NMR spectrum (125MHz) of 3ad in CDCl₃

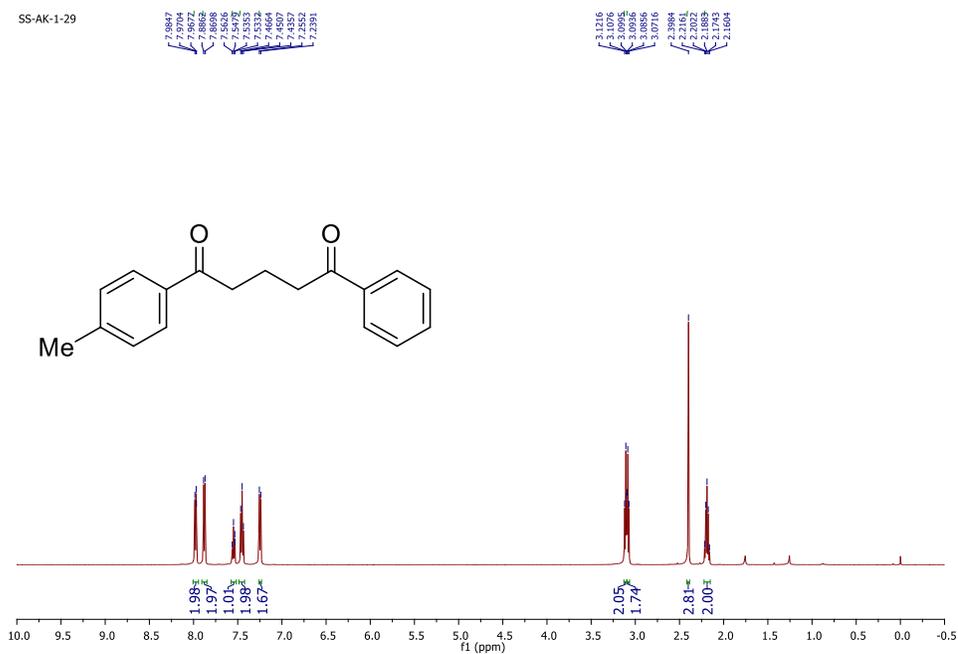


Figure 29: ¹H NMR spectrum (500MHz) of 3ba in CDCl₃

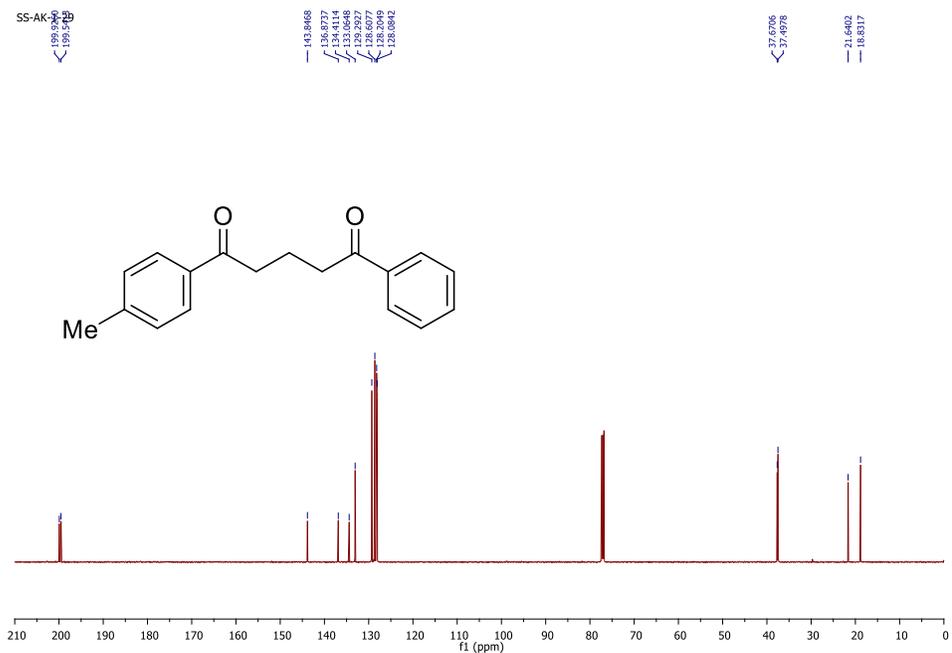


Figure 30: ¹³C NMR spectrum (125MHz) of 3ba in CDCl₃

SS-AK-I-35

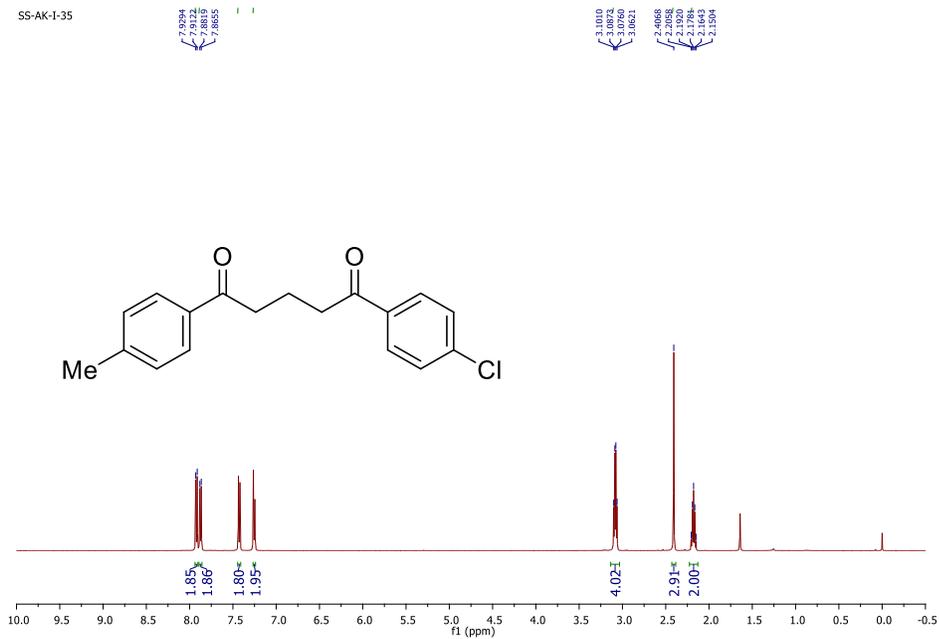


Figure 31: $^1\text{H NMR}$ spectrum (500MHz) of **3bd** in CDCl_3

SS-AK-I-35

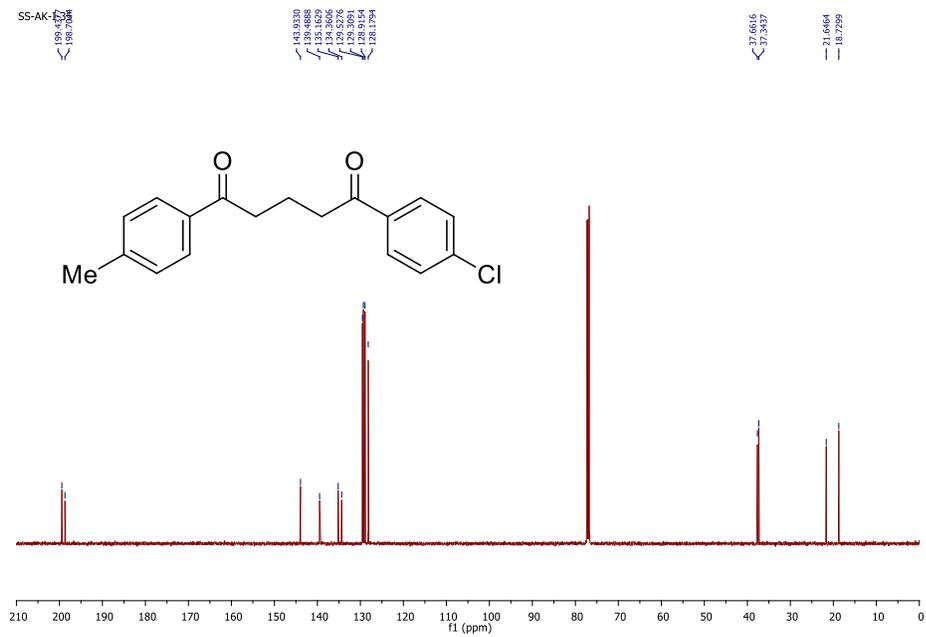


Figure 32: $^{13}\text{C NMR}$ spectrum (125MHz) of **3bd** in CDCl_3

SS-AK-I-9

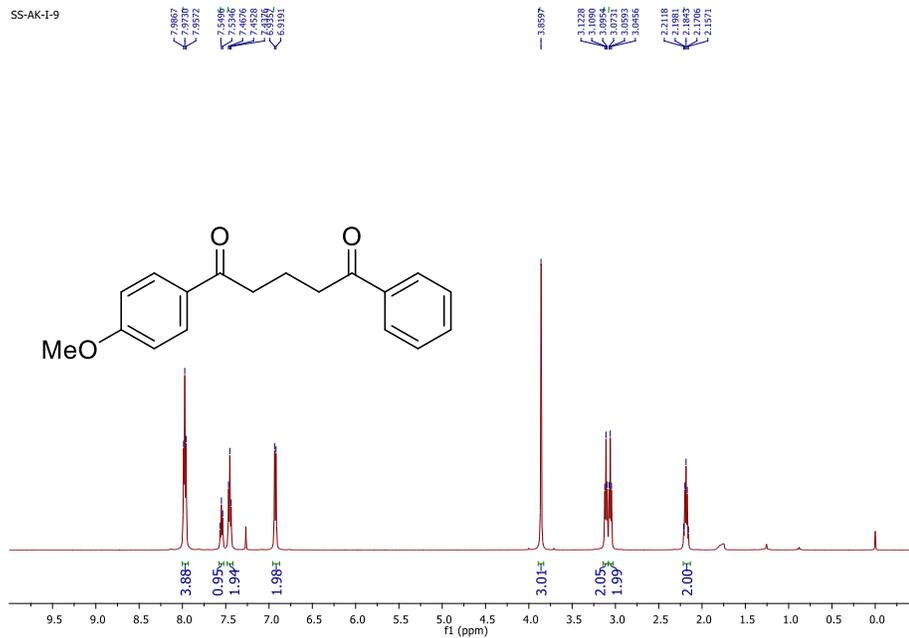


Figure 33: ¹H NMR spectrum (500MHz) of 3ca in CDCl₃

SS-AK-I-9

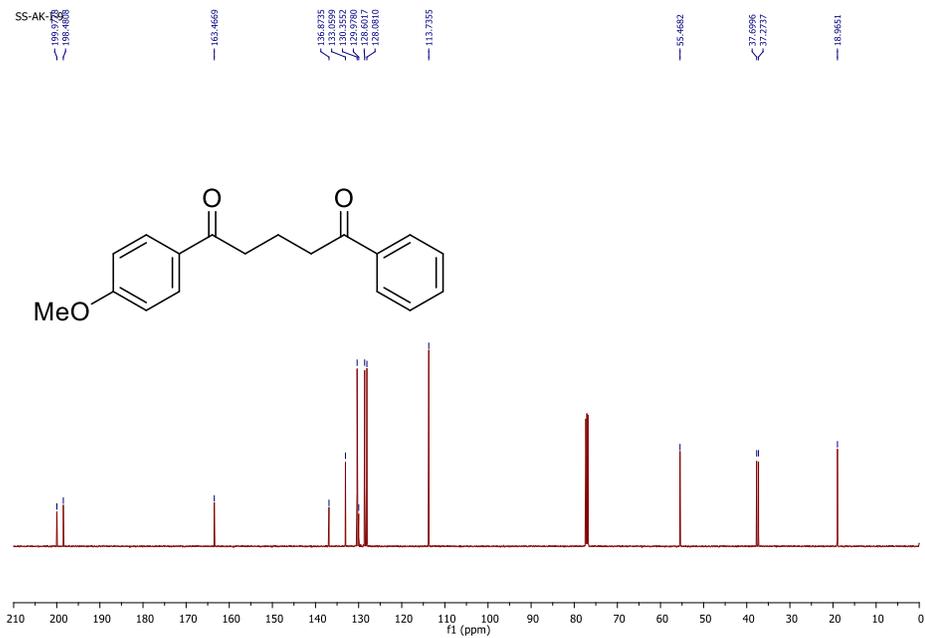


Figure 34: ¹³C NMR spectrum (125MHz) of 3ca in CDCl₃

SS-AK-1-13

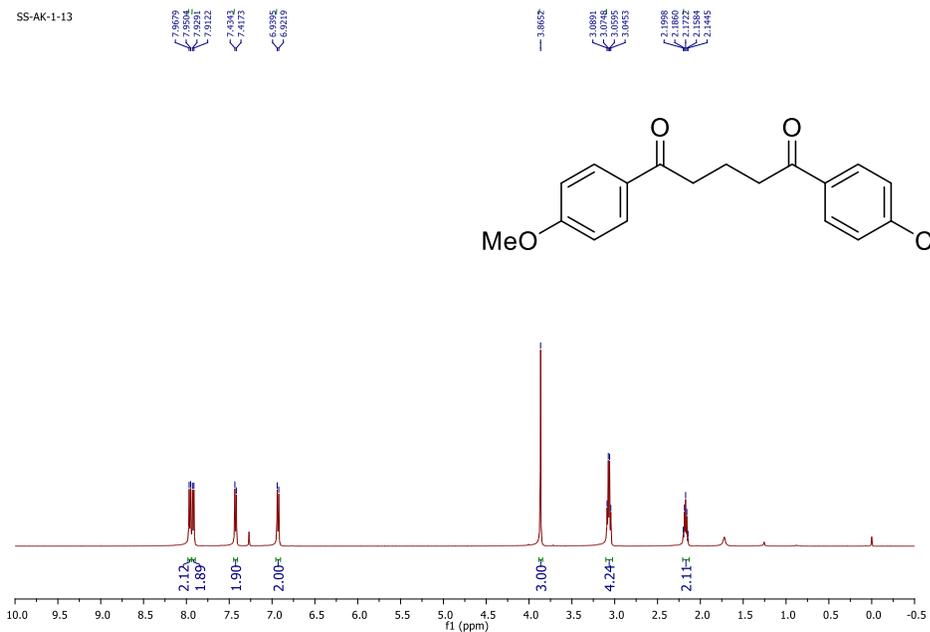


Figure 35: ¹H NMR spectrum (500MHz) of 3cd in CDCl₃

SS-AK-1-13

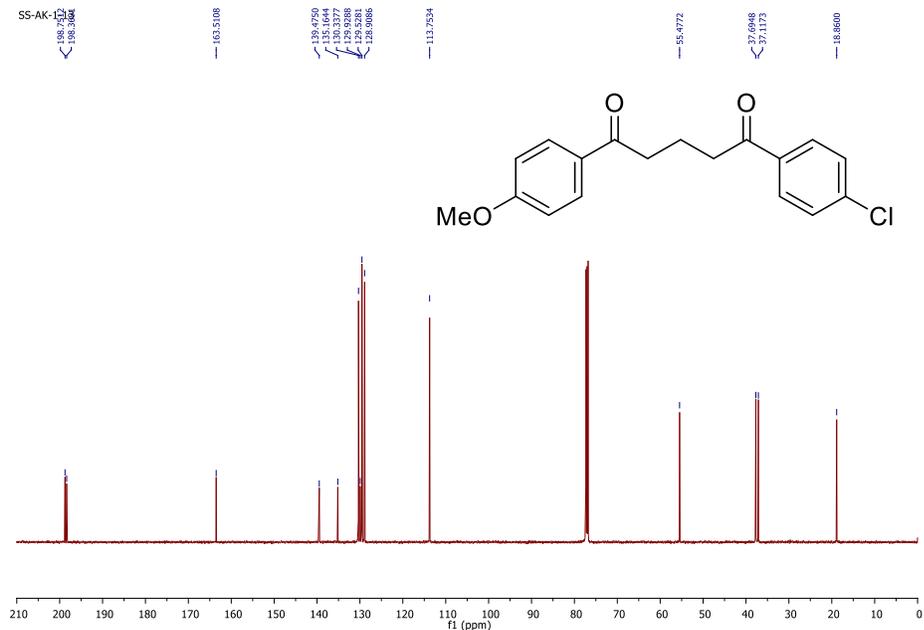


Figure 36: ¹³C NMR spectrum (125MHz) of 3cd in CDCl₃

SS-AK-1-35

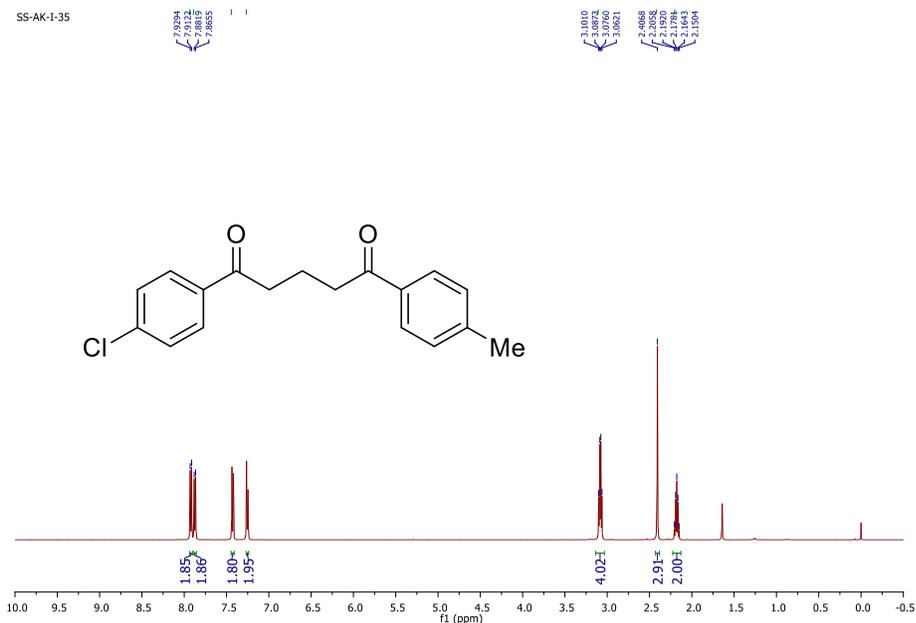


Figure 39: ^1H NMR spectrum (500MHz) of **3db** in CDCl_3

SS-AK-1-35

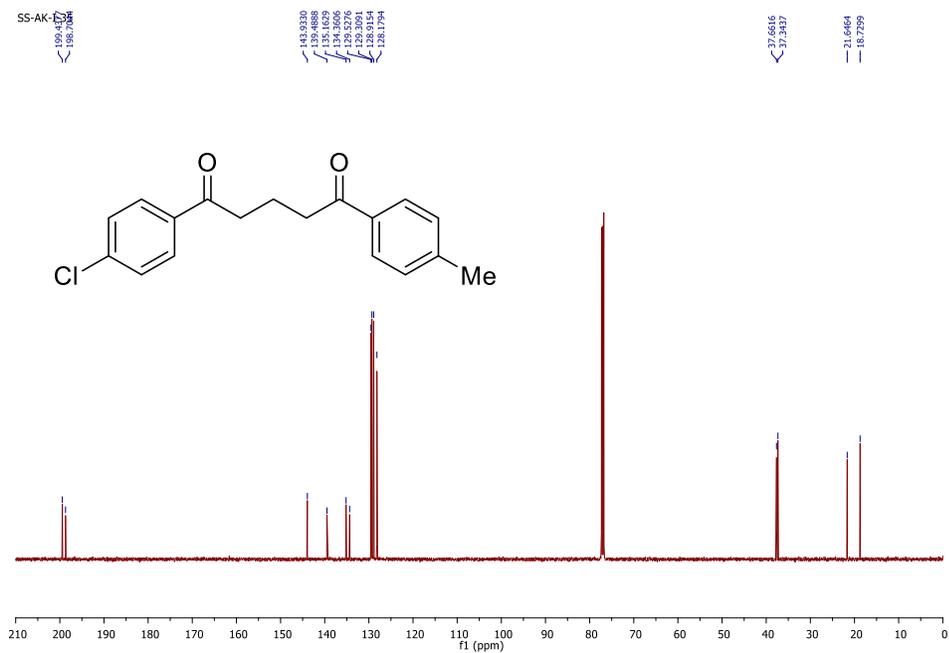


Figure 40: ^{13}C NMR spectrum (125MHz) of **3db** in CDCl_3

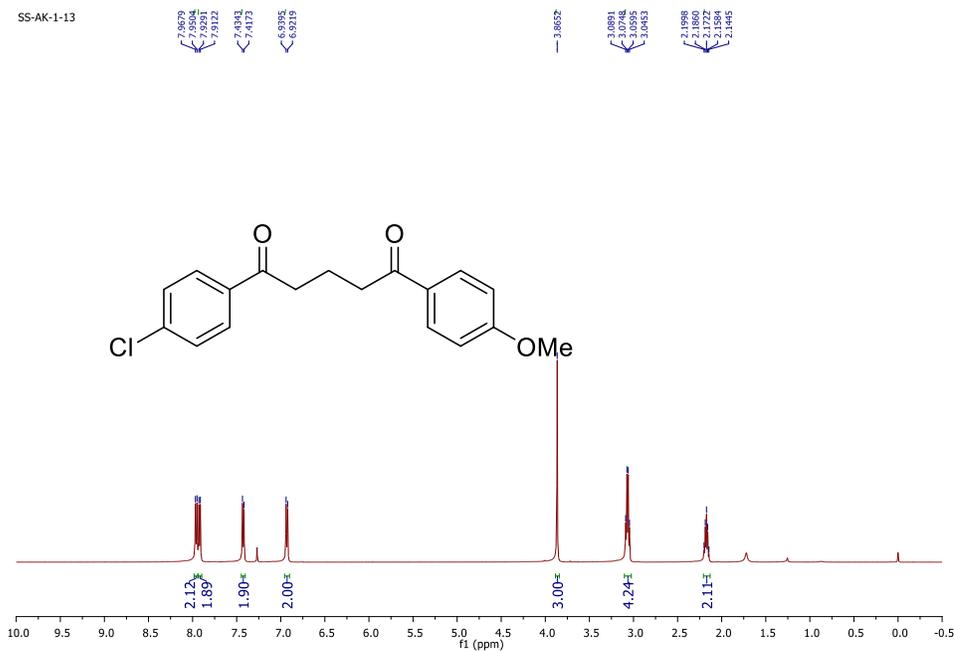


Figure 41: ^1H NMR spectrum (500MHz) of 3dc in CDCl_3

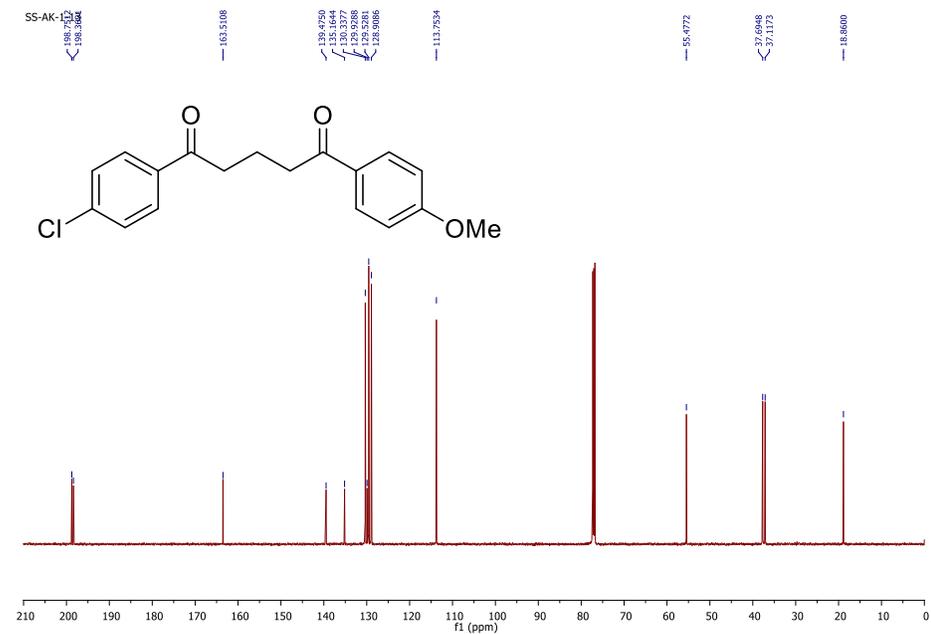


Figure 42: ^{13}C NMR spectrum (125MHz) of 3dc in CDCl_3

SS-AK-1-36

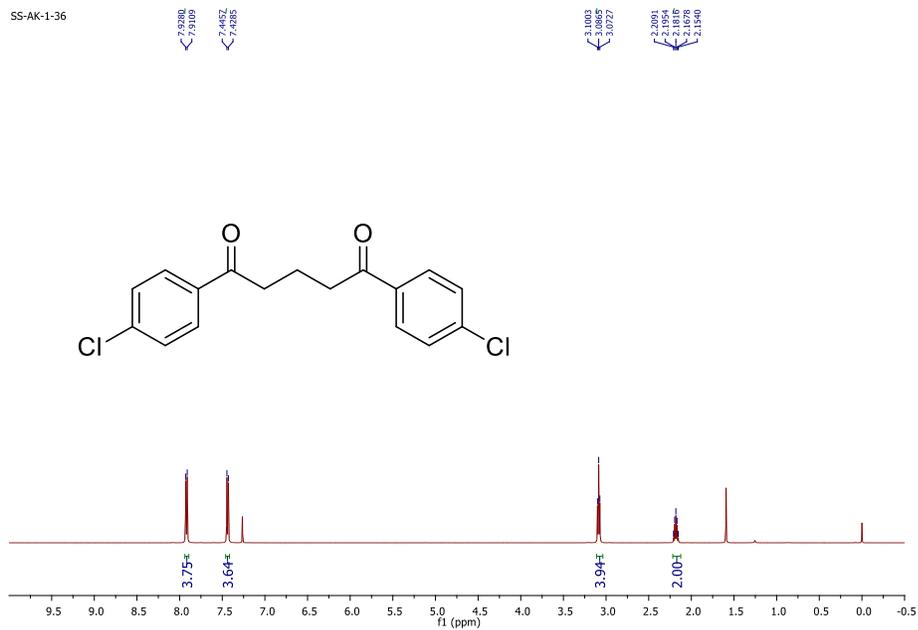


Figure 43: ¹H NMR spectrum (500MHz) of **3dd** in CDCl₃

SS-AK-1-36

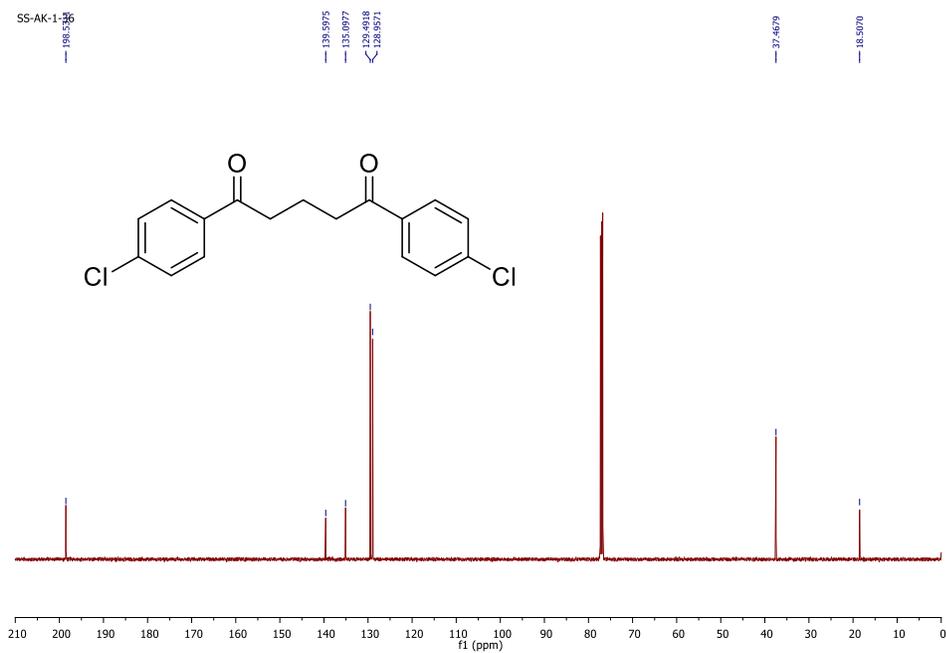


Figure 44: ¹³C NMR spectrum (125MHz) of **3dd** in CDCl₃

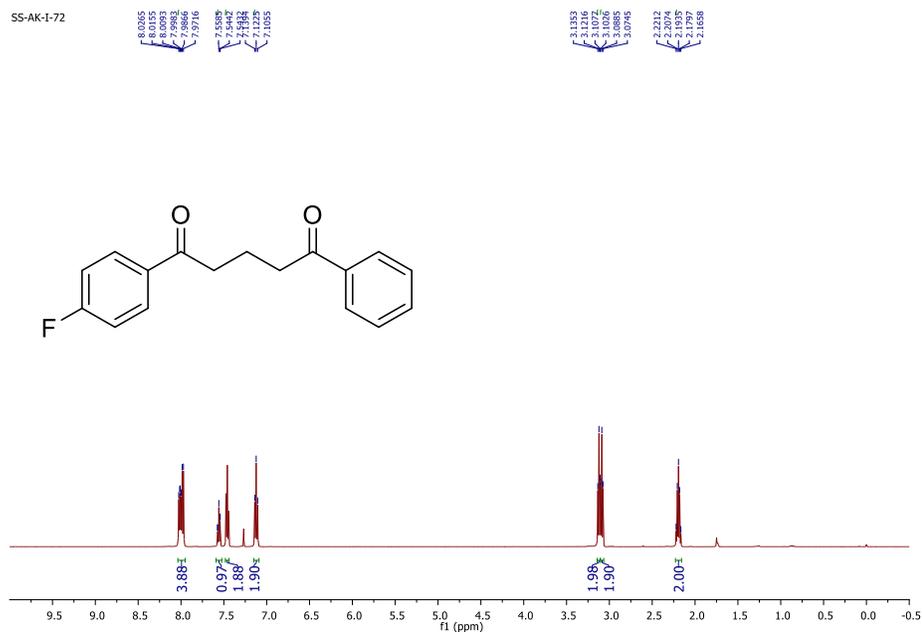


Figure 45: ^1H NMR spectrum (500MHz) of **3ea** in CDCl_3

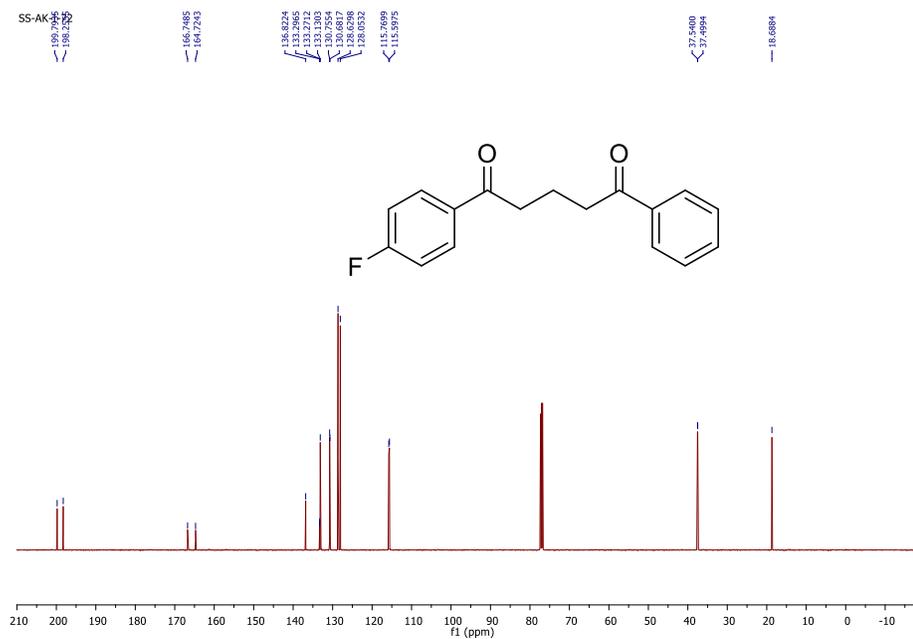
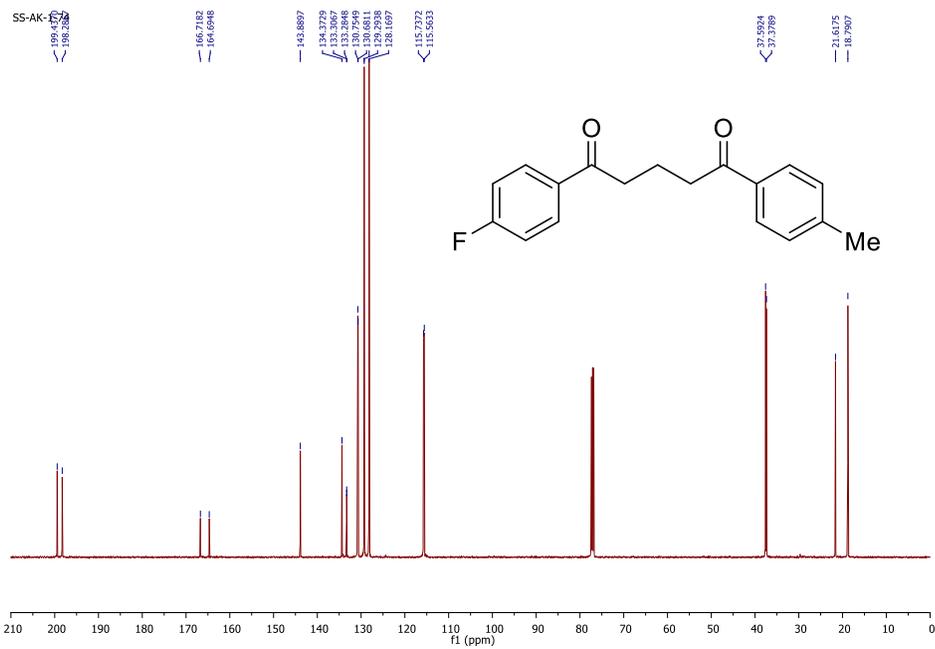
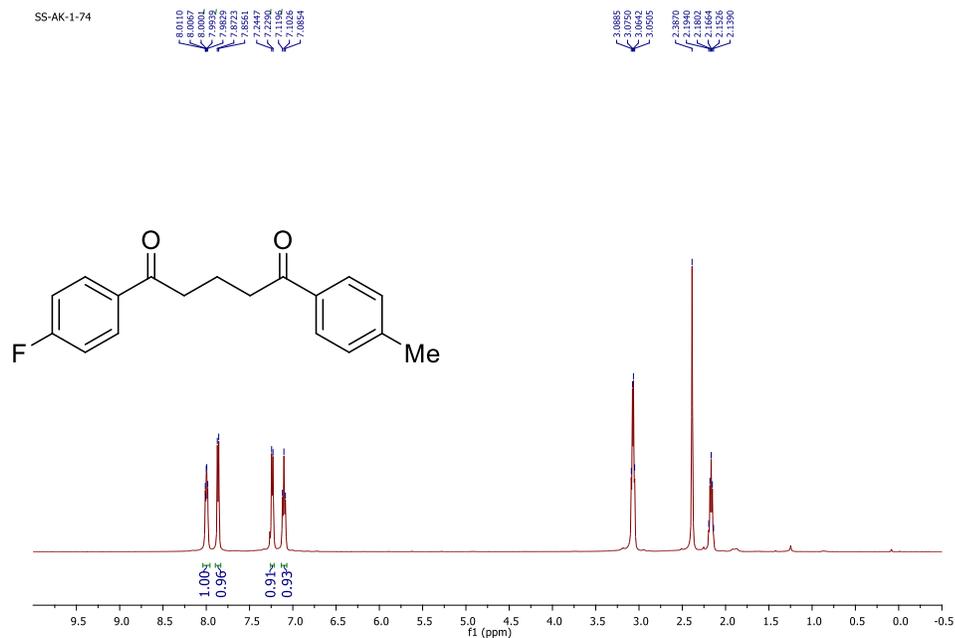


Figure 46: ^{13}C NMR spectrum (125MHz) of **3ea** in CDCl_3



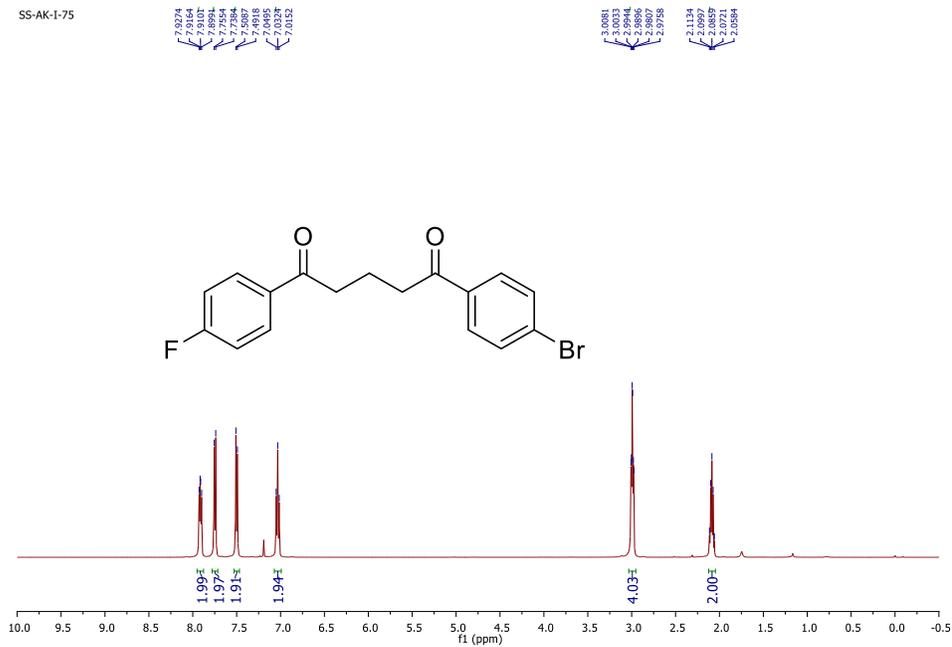


Figure 49: ^1H NMR spectrum (500MHz) of **3ee** in CDCl_3

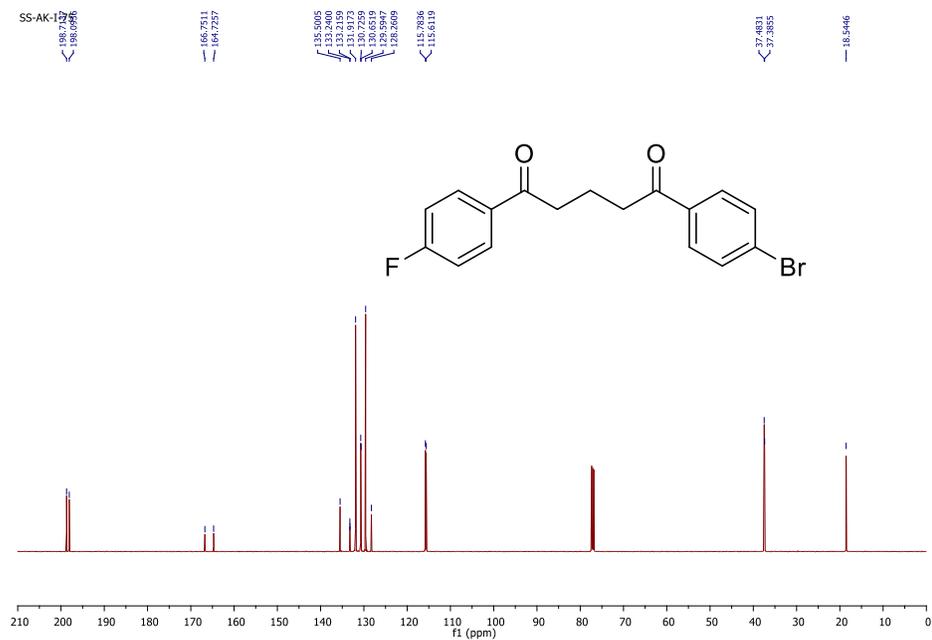


Figure 50: ^{13}C NMR spectrum (125MHz) of **3ee** in CDCl_3

SS-AK-1-76

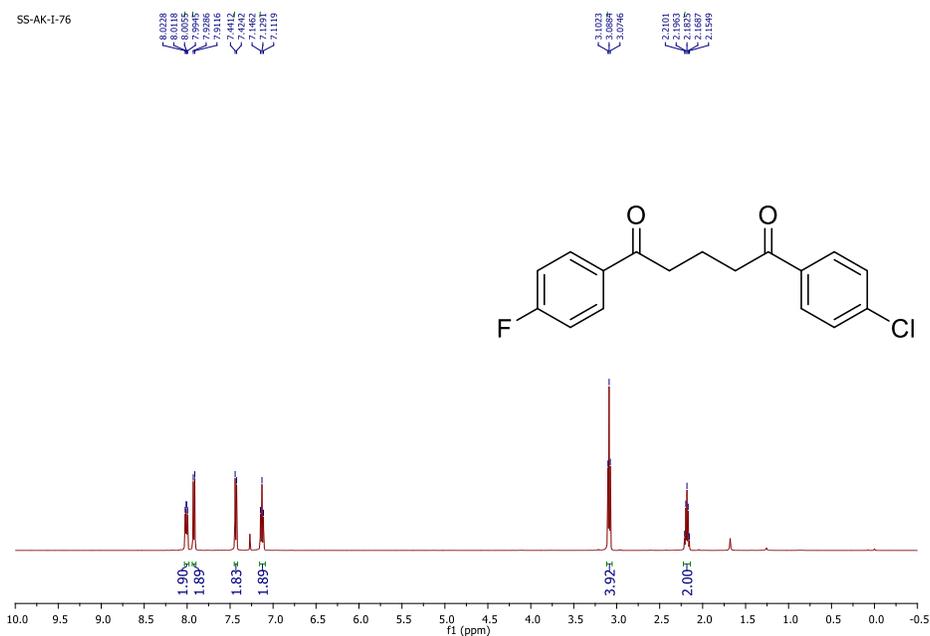


Figure 51: ¹H NMR spectrum (500MHz) of 3ed in CDCl₃

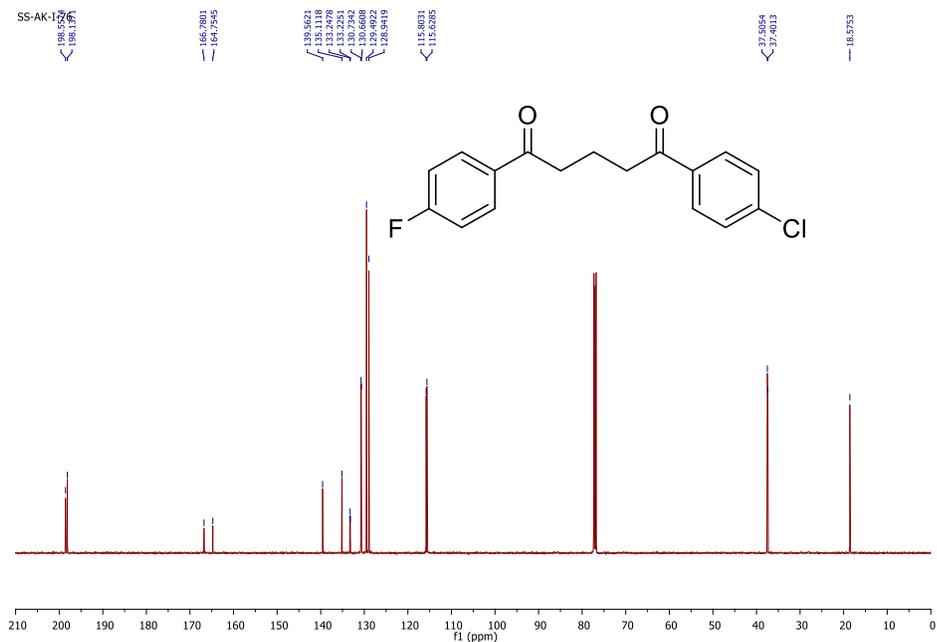


Figure 52: ¹³C NMR spectrum (125MHz) of 3ed in CDCl₃

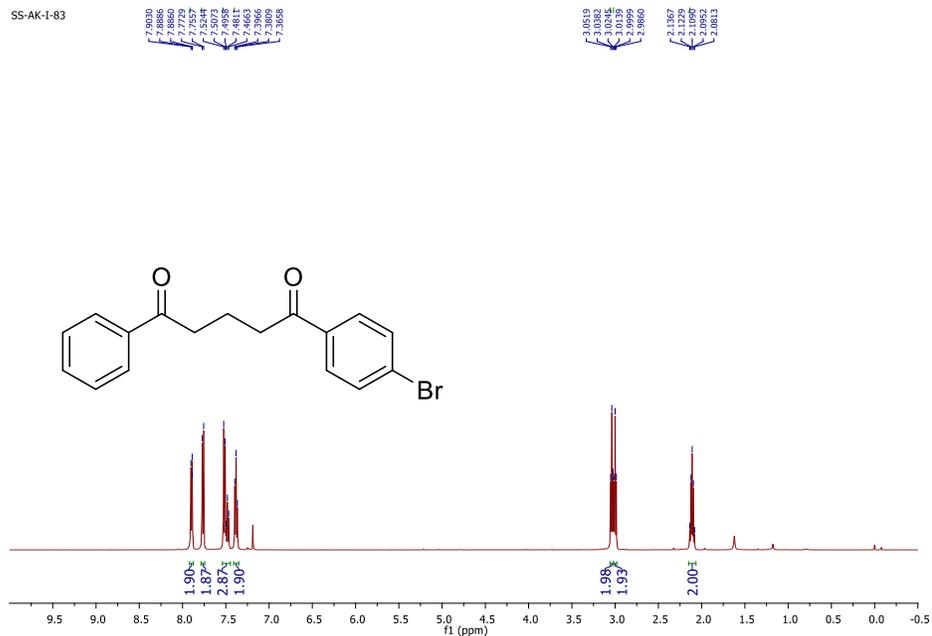


Figure 53: ^1H NMR spectrum (500MHz) of 3ae in CDCl_3

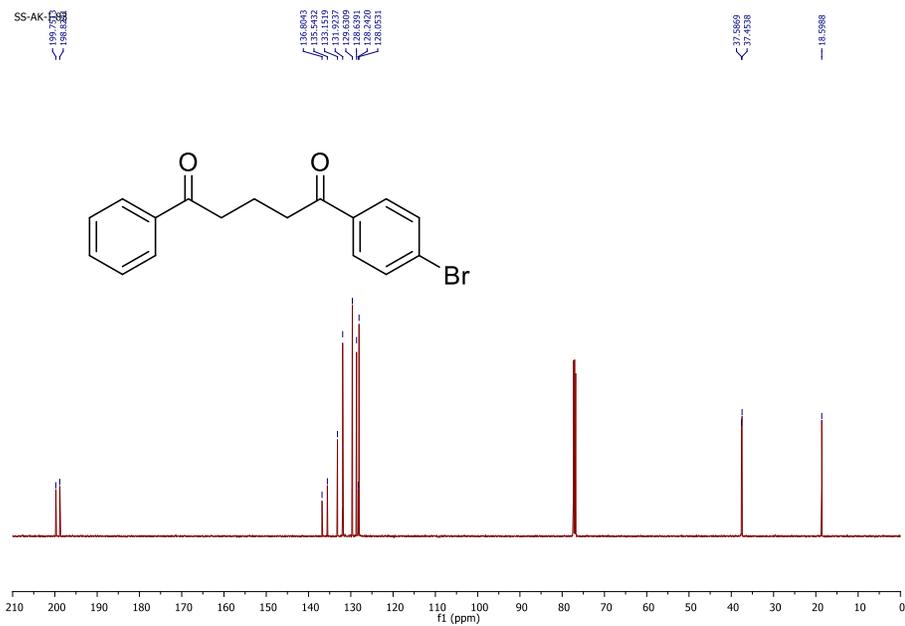


Figure 54: ^{13}C NMR spectrum (125MHz) of 3ae in CDCl_3

SS-AK-I-98

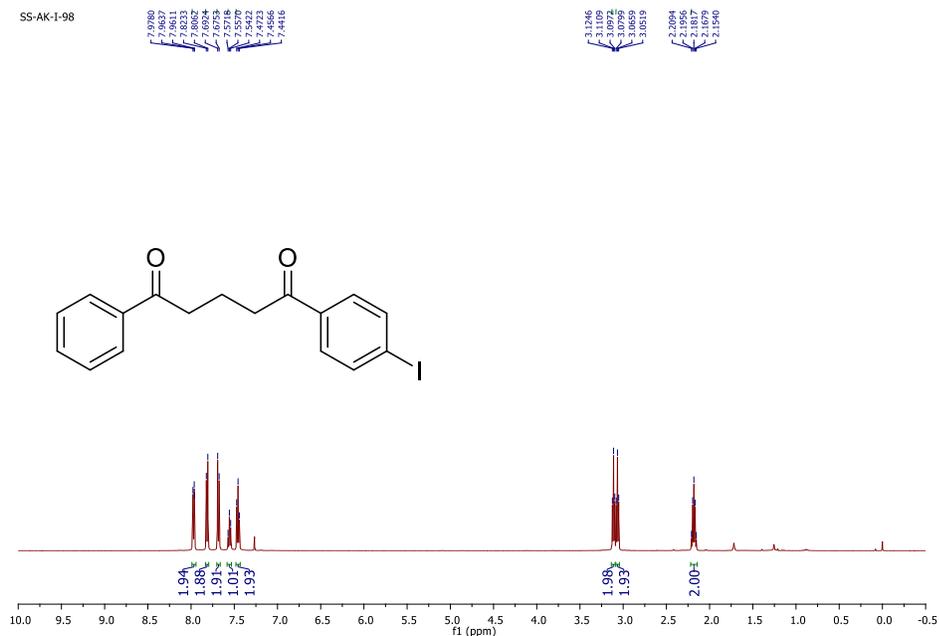


Figure 55: ¹H NMR spectrum (500MHz) of 3af in CDCl₃

SS-AK-I-98



Figure 56: ¹³C NMR spectrum (125MHz) of 3af in CDCl₃

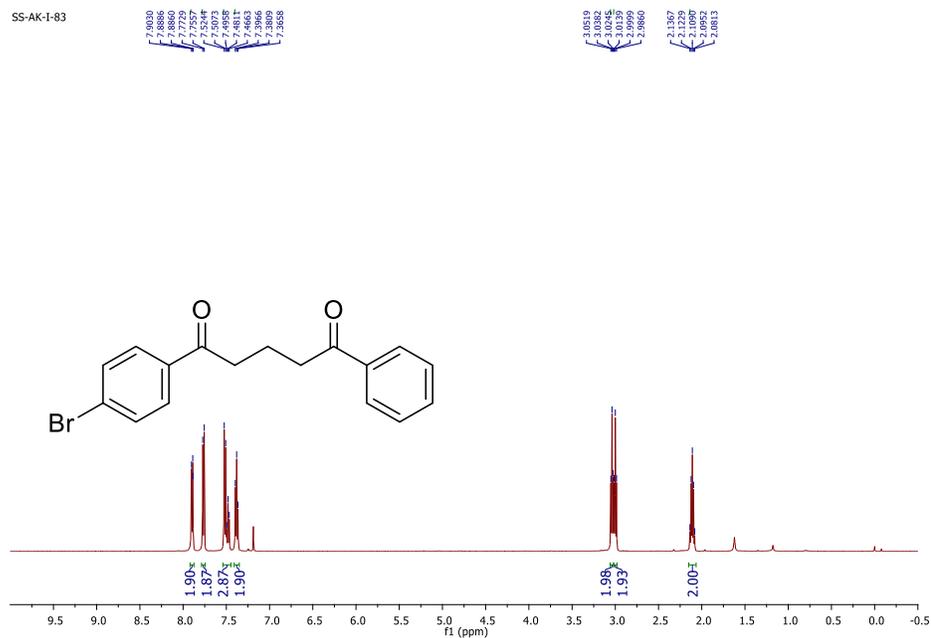


Figure 57: ^1H NMR spectrum (500MHz) of **3fa** in CDCl_3

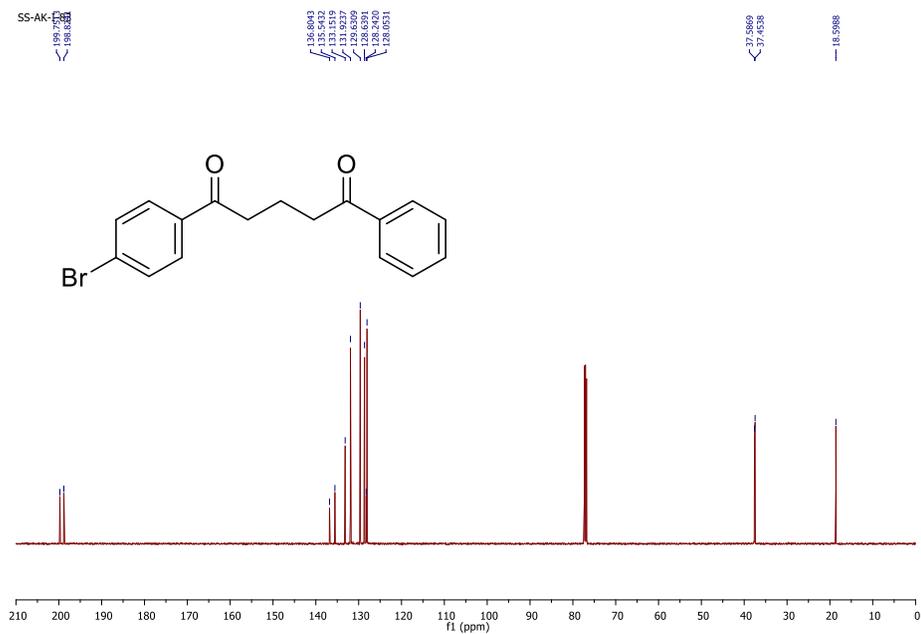


Figure 58: ^{13}C NMR spectrum (125MHz) of **3fa** in CDCl_3

SS-AK-1-67

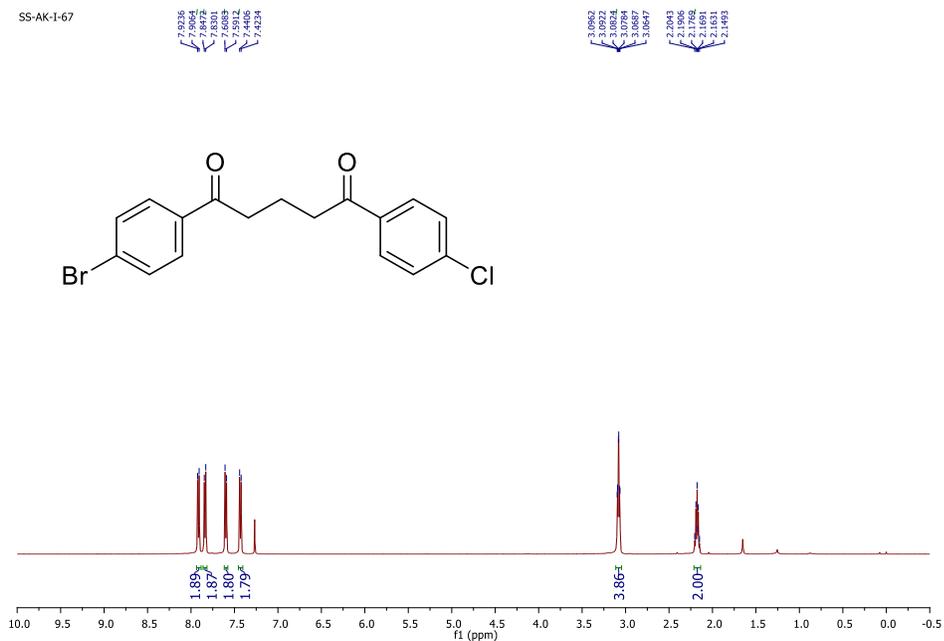


Figure 59: ¹H NMR spectrum (500MHz) of 3fd in CDCl₃

SS-AK-1-67

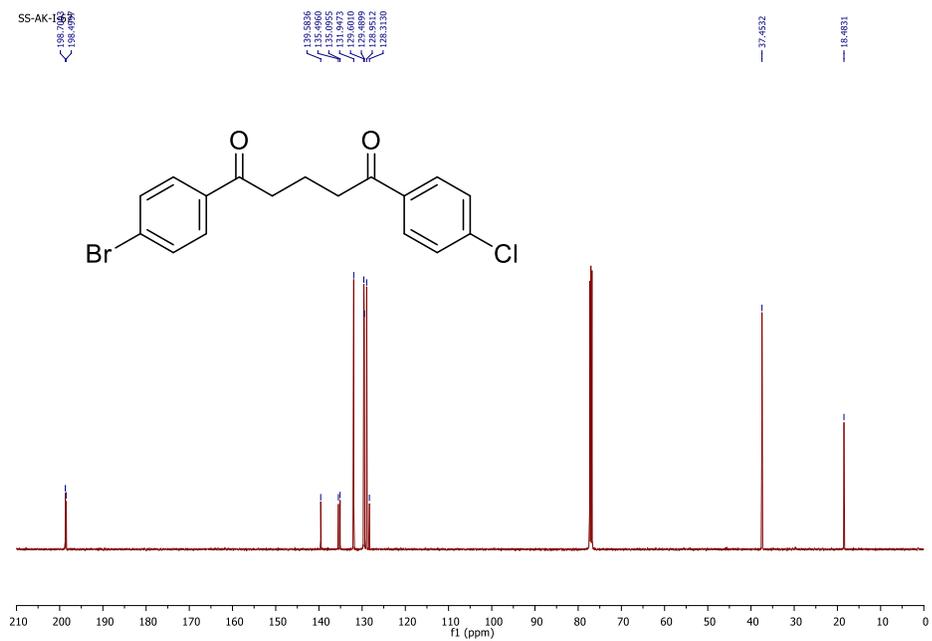
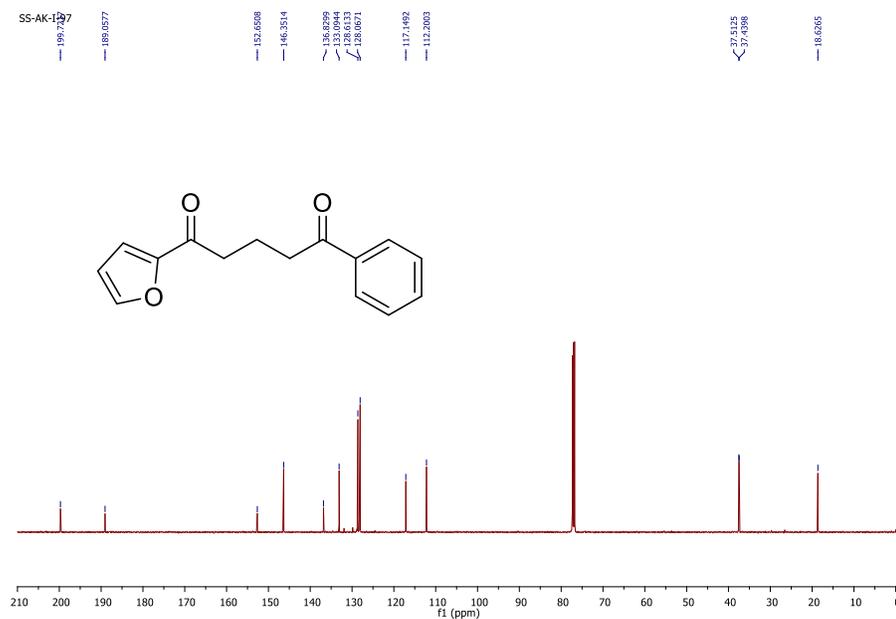


Figure 60: ¹³C NMR spectrum (125MHz) of 3fd in CDCl₃



SS-AR-I-56

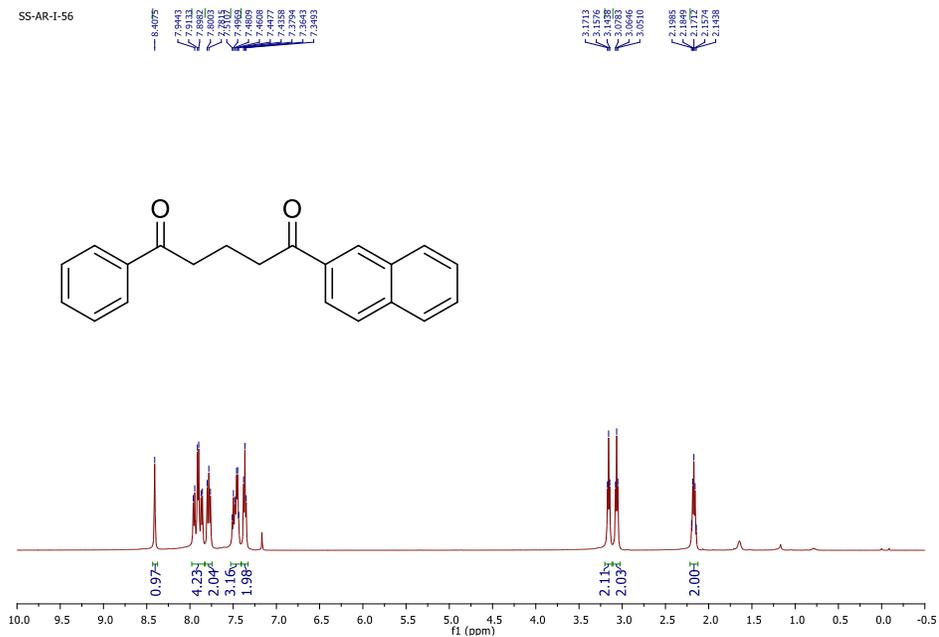


Figure 65: ¹H NMR spectrum (500MHz) of 3ai in CDCl₃

SS-AR-I-56

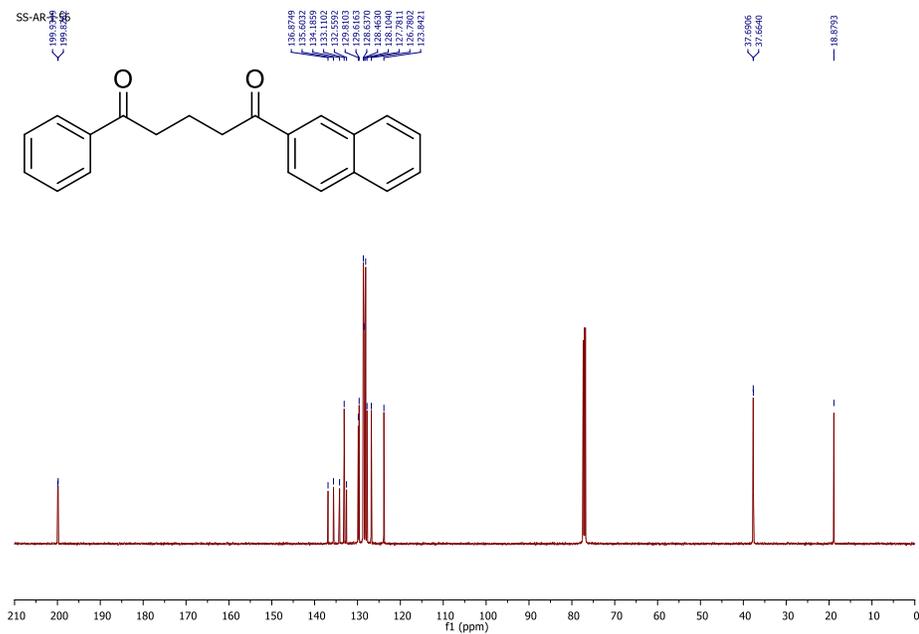


Figure 66: ¹³C NMR spectrum (125MHz) of 3ai in CDCl₃

SS-AR-1-57

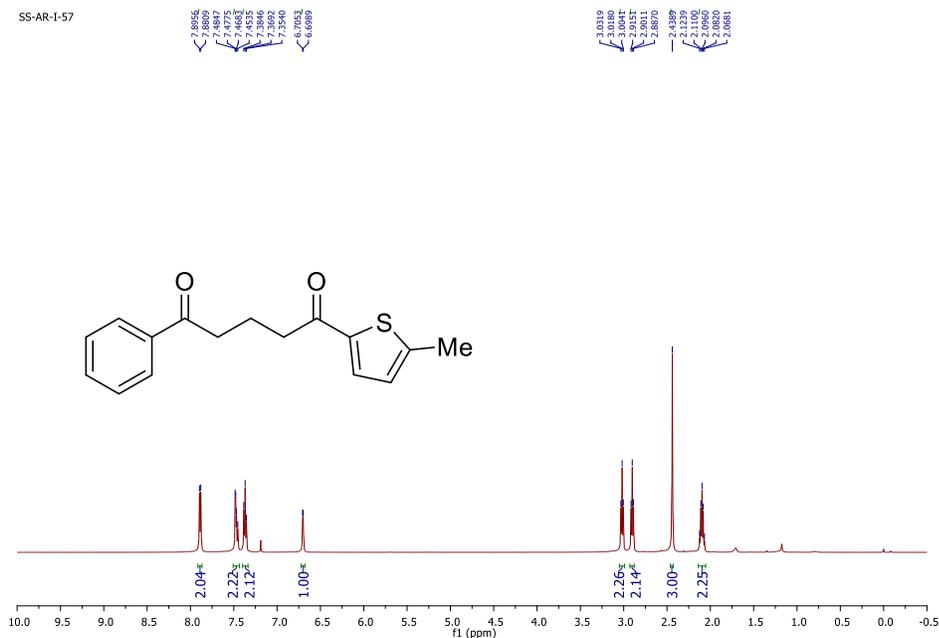


Figure 67: ¹H NMR spectrum (500MHz) of 3aj in CDCl₃

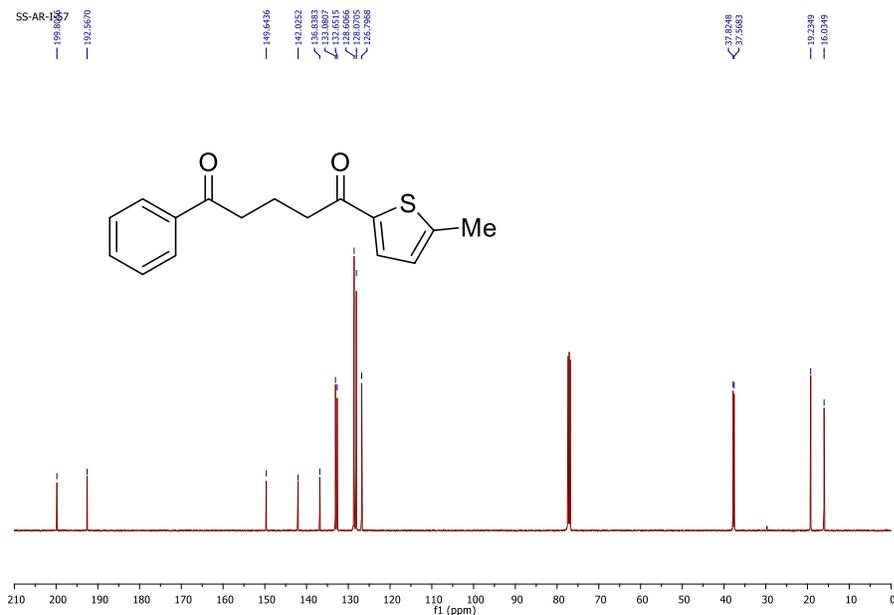


Figure 68: ¹³C NMR spectrum (125MHz) of 3aj in CDCl₃

SS-AK-1-96

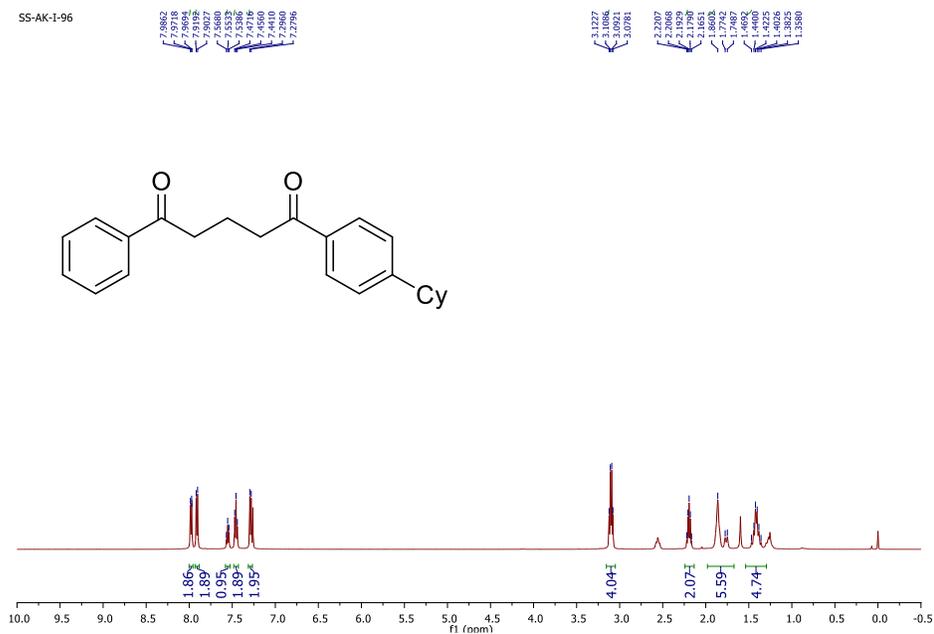


Figure 69: ¹H NMR spectrum (500MHz) of 3ag in CDCl₃

SS-AK-1-96

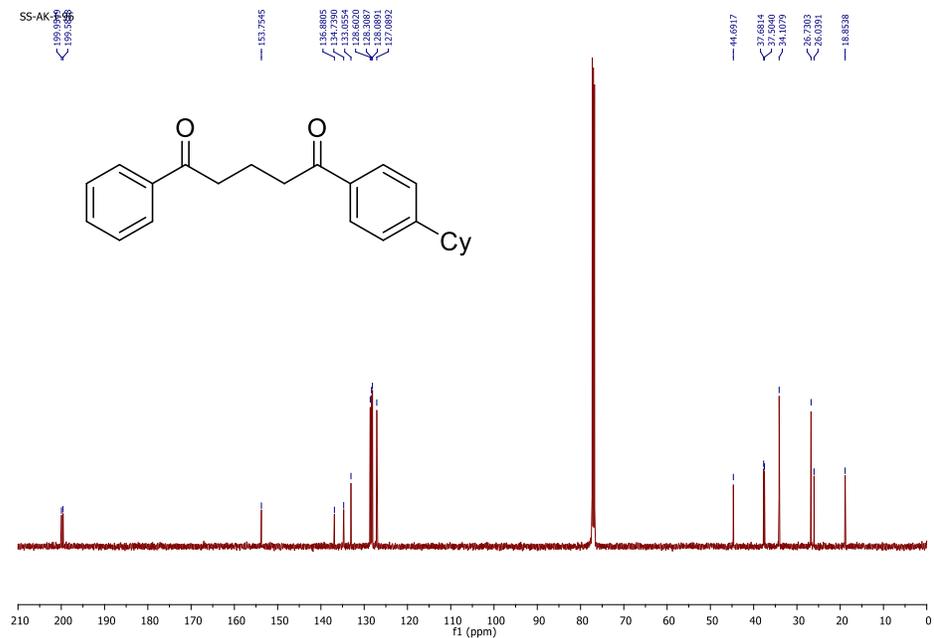


Figure 70: ¹³C NMR spectrum (125MHz) of 3ag in CDCl₃

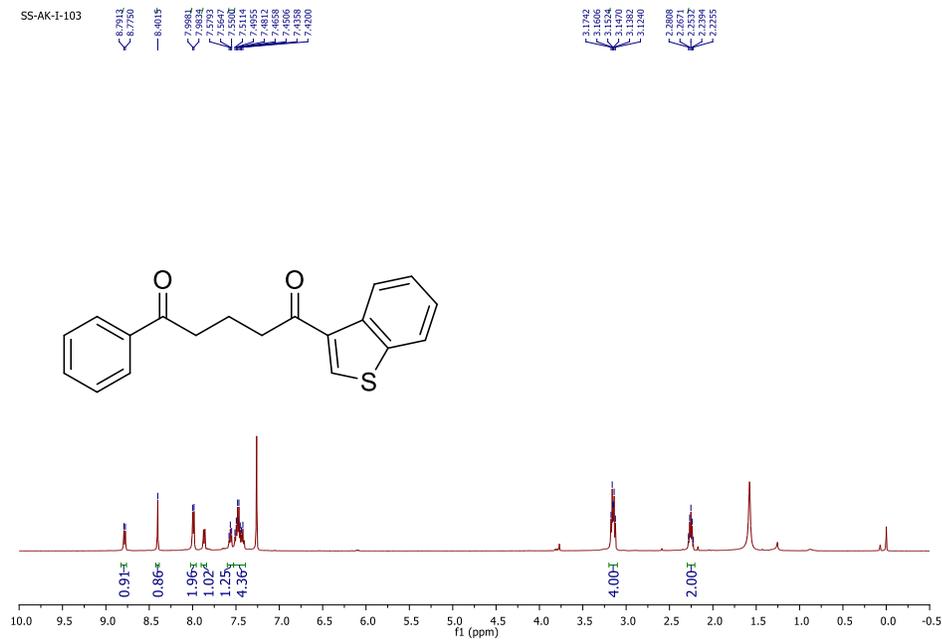


Figure 71: ^1H NMR spectrum (500MHz) of **3al** in CDCl_3

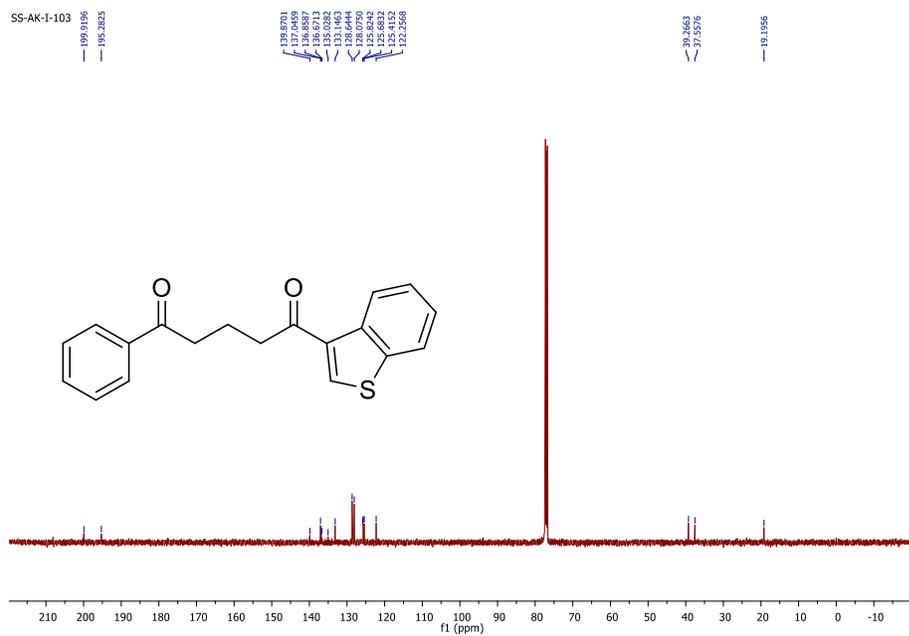


Figure 72: ^{13}C NMR spectrum (125MHz) of **3al** in CDCl_3

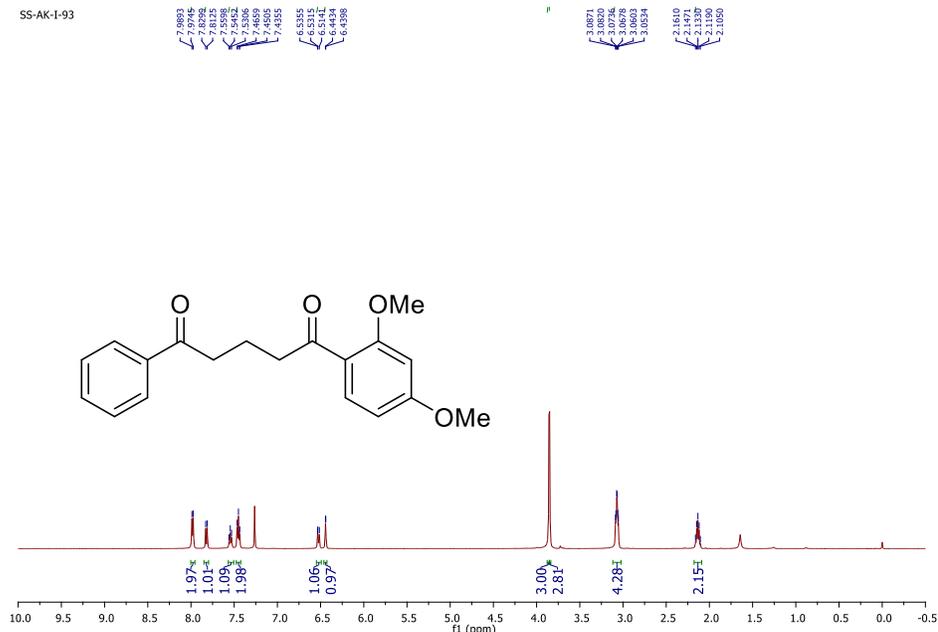


Figure 73: ^1H NMR spectrum (500MHz) of **3am** in CDCl_3

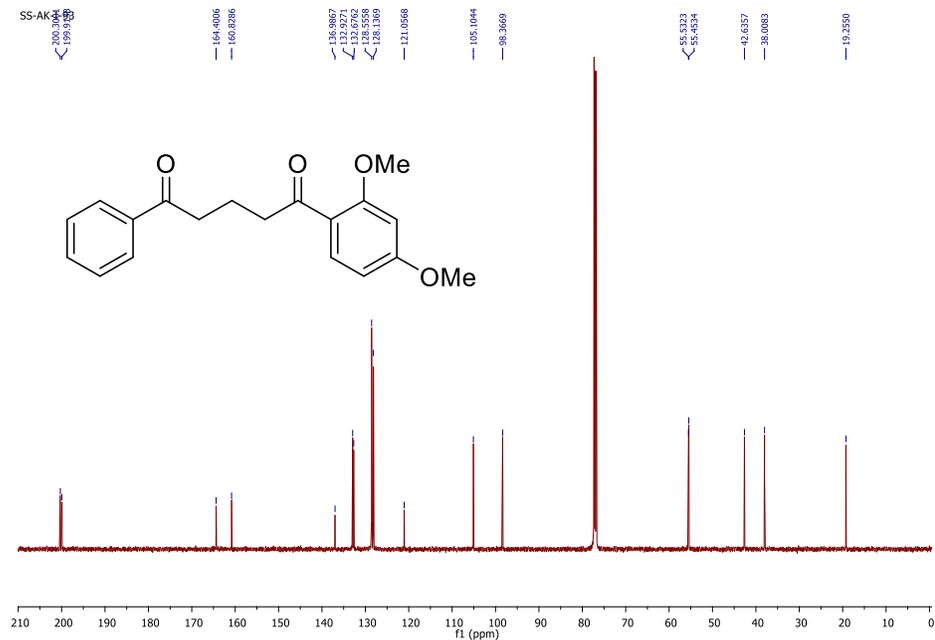


Figure 74: ^{13}C NMR spectrum (125MHz) of **3am** in CDCl_3

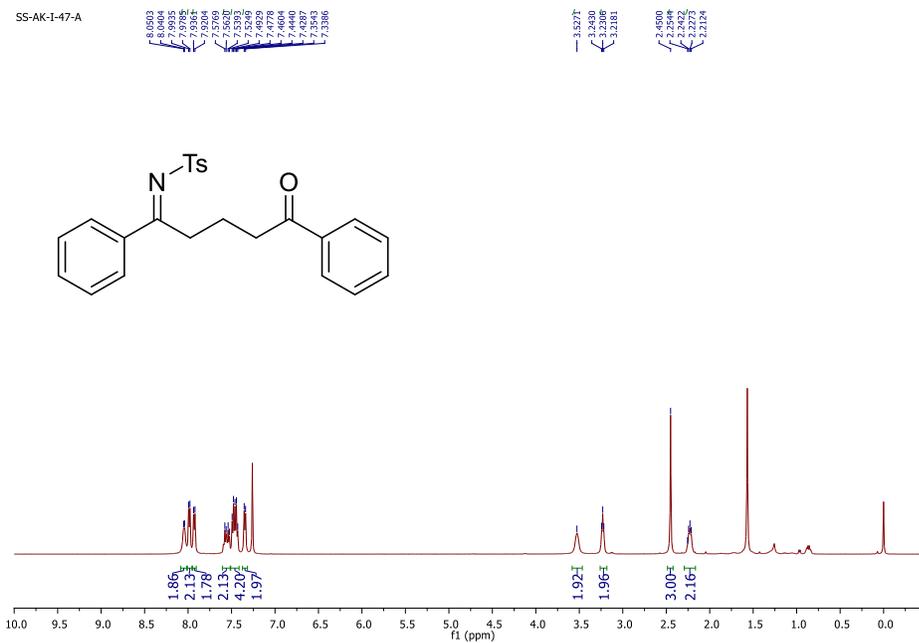


Figure 75: ^1H NMR spectrum (500MHz) of 3aa' in CDCl_3

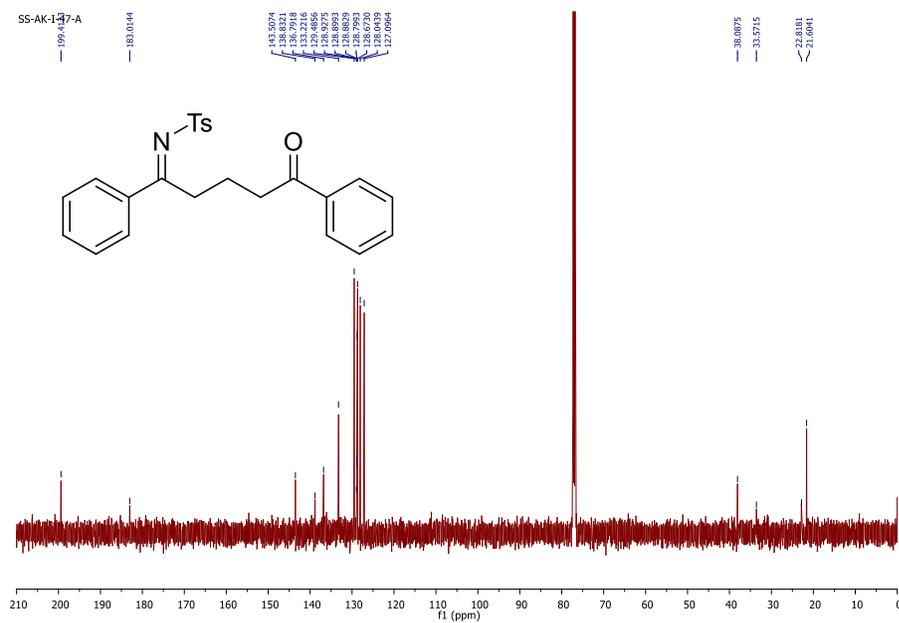
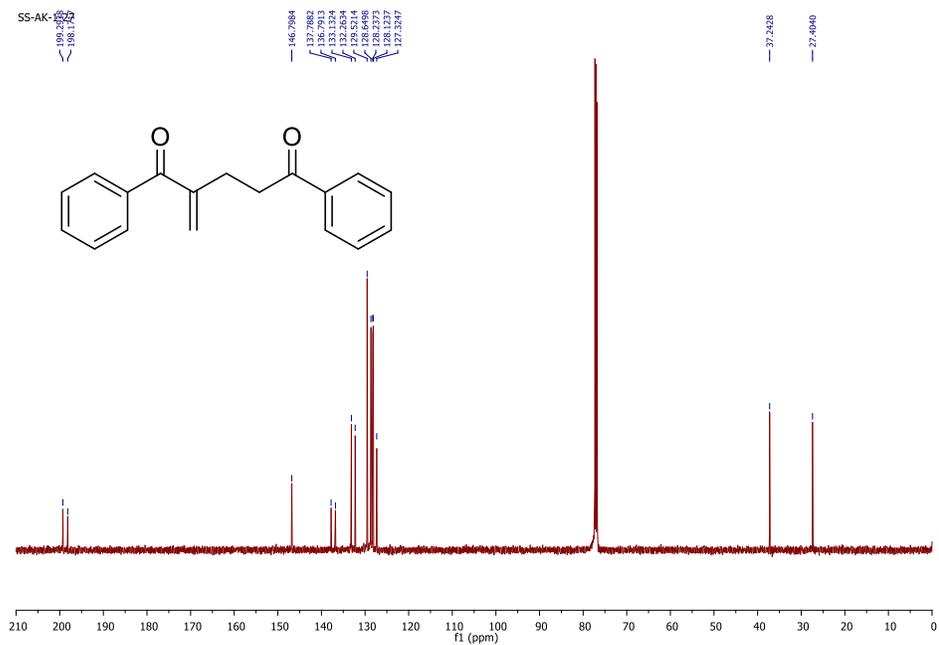
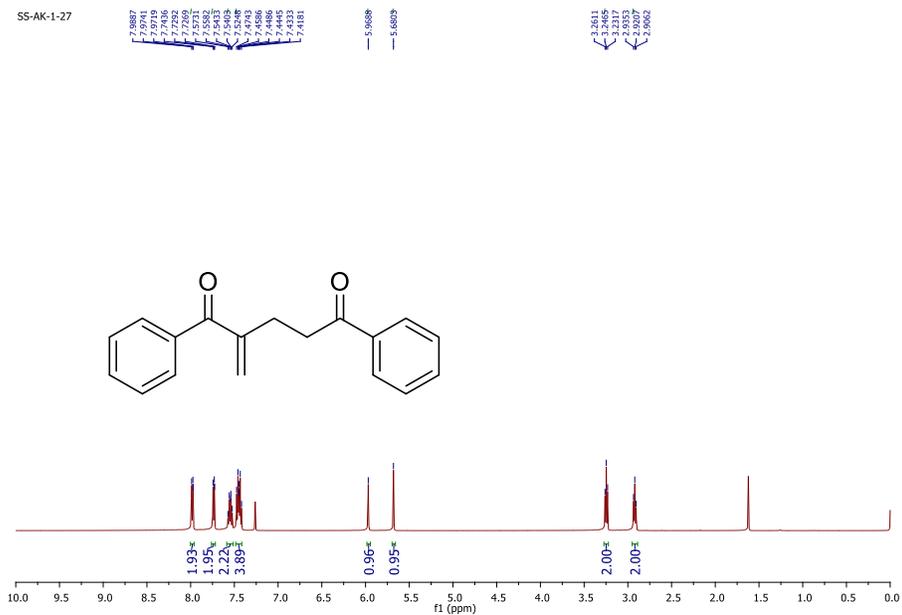


Figure 76: ^{13}C NMR spectrum (125MHz) of 3aa' in CDCl_3



SS-AK-1-70

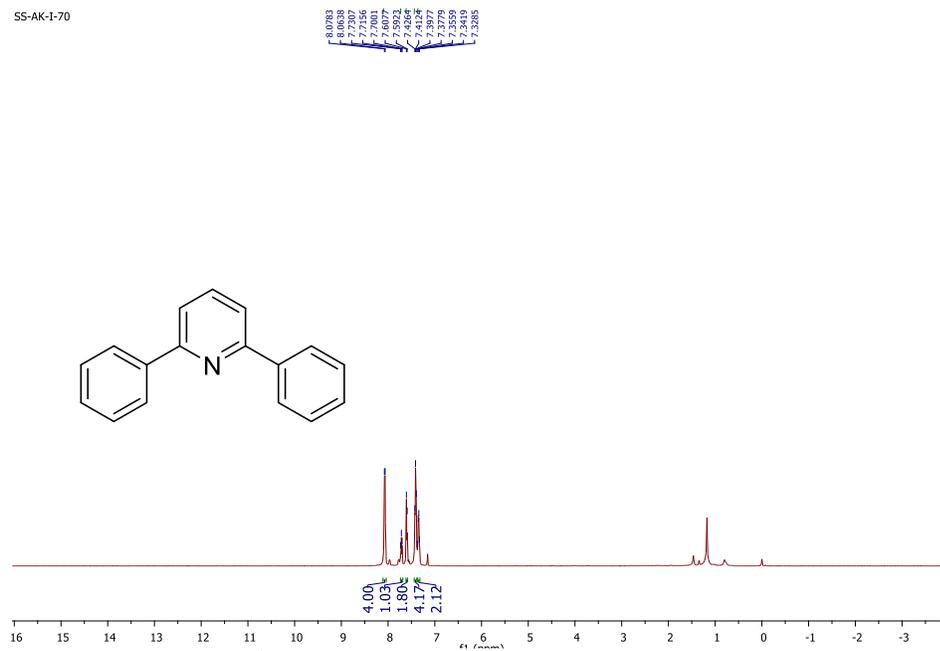


Figure 79: ^1H NMR spectrum (500MHz) of **6aa** in CDCl_3

SS-AK-1-70

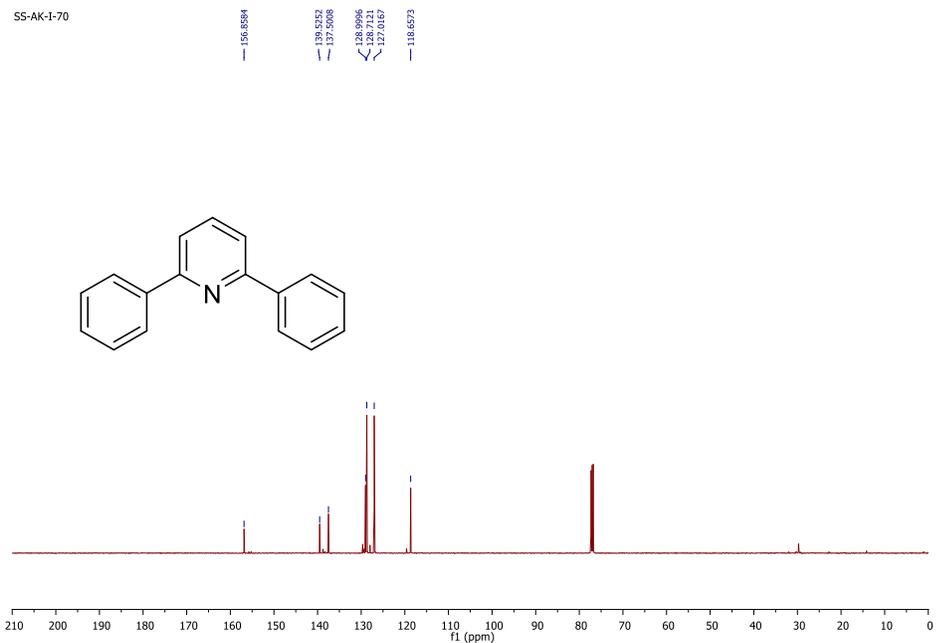


Figure 80: ^{13}C NMR spectrum (125MHz) of **6aa** in CDCl_3

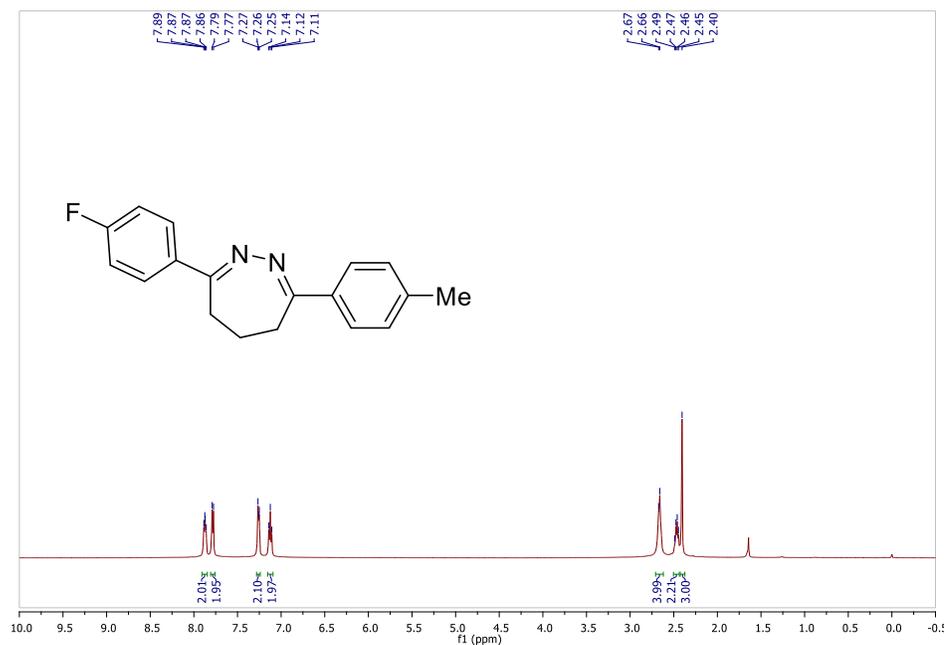


Figure 81: ¹H NMR spectrum (500MHz) of 7aa in CDCl₃

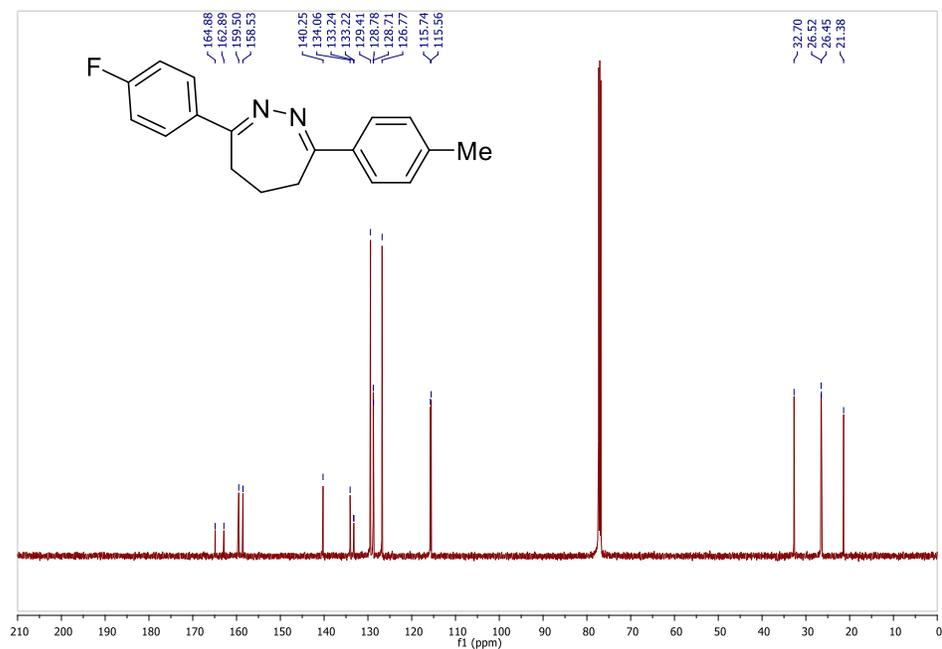


Figure 82: ¹³C NMR spectrum (125MHz) of 7aa in CDCl₃

References

1. Asressu, K.H.; Chan, C. K.; Wang, C. C. One-pot synthesis of 1,5-diketones under a transition-metal-free condition: application in synthesis of 2,4,6-triaryl pyridine derivatives, *ACS Omega*, **2021**, *6*, 7296-7311.
2. Saikia, P.; Gogoi, S.; Boruah, R.C. Carbon-carbon bond cleavage reaction: synthesis of multisubstituted pyrazolo[1,5-a]pyrimidines, *J. Org. Chem.* **2015**, *80*, 6885-6889.
3. Liu, W-y.; Xu, Q-h.; Liang, H.; Chen, B-h.; Liu, W-m.; Ma, Y-x. Preparation of 1,5-diketone derivatives containing ferrocenyl by Michael reaction under solvent-free condition, *J. Organomet. Chem.* **2001**, *637*, 719-722.
4. Yanagisawa, A.; Takashi, H.; Arai, T. One-pot synthesis of 1,5-diketones catalyzed by barium hydroxide, *Tetrahedron* **2007**, *63*, 8581-8585.
5. Goh, K. K. K.; Kim, S.; Zard, S. Z. Free radical variant for the synthesis of functionalized 1,5-diketones, *Org. Lett.* **2013**, *15*, 4818-4821.
6. Guo, R.; Zhang, Z. Expedient synthesis of 1,5-diketones by rhodium-catalysed hydroacylation enabled C-C bond cleavage, *J. Am. Chem. Soc.* **2017**, *139*, 12891-12894.
7. Zhen, J.; Du, X.; Xu, X.; Li, Y.; Yuan, H.; Xu, D.; Xue, C.; Luo, Y. Visible-light-mediated late-stage sulfonylation of boric acid via N-S bond activation of sulfonamides, *ACS Catal.* **2022**, *12*, 1986-1991.
8. Ruano, J. L. G.; Alemán, J.; Cid, M.B.; Parra, A. A general method for preparation of N-sulfonyl aldimines and ketimines, *Org. Lett.* **2005**, *7*, 179-182.
9. Kuwagara, T.; Fukuyama, T.; Picard, B.; Ryu, I.; RuH-catalyzed synthesis of 1,3-aryl ketones via cross coupling of enones with aromatic aldehydes and mechanistic insights, *Adv. Synth. Catal.* **2022**, *364*, 3725-3729.
10. Patel, A. K.; Rathor, S. S.; Samanta, S. Regioselective access to di- and trisubstituted pyridines via metal-oxidant-solvent-free domino reaction involving 3-chloropropiophenones, *Org. Biomol. Chem.* **2022**, *20*, 6759-6765.